



Design of 0–3 type nanocomposites using hydrothermal sintering

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ABSTRACT

We report here the successful design of 0–3 type nanocomposites where 30 nm ferromagnetic metallically conducting cores of manganite $\text{La}_{0.66}\text{Sr}_{0.34}\text{MnO}_3$ (LSMO) are discretely distributed in an insulating silica matrix. Starting from LSMO@ SiO_2 core@shell nanoparticles, hydrothermal sintering process was used as a low temperature densification route (300 °C, 350 MPa, 90 min) in presence of 0.2 M aqueous sodium hydroxide solution. This process based on a pressure solution creep in the contact zones between nanoparticles allows the design of complex microstructures preventing the grain growth of the cores and the formation of interphases between the cores and the matrix.

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Advanced materials such as new high-performance composites are our allies for a sustainable future. For instance, the introduction of an insulating phase at the grain boundaries in polycrystalline metallically conducting manganites with $(\text{La}, \text{Sr})\text{MnO}_3$ (LSMO) or $(\text{La}, \text{Ca})\text{MnO}_3$ (LCMO) compositions promotes the role of tunneling of spin-polarized carriers between neighboring perovskite grains [1–8]. Consequently, the macroscopic transport is strongly dependent on external magnetic field, and such composites are promising as elements with large negative magnetoresistance (positive magnetoconductance) for the design of spintronic devices. The development of these systems, mainly issued from the densification of mixtures of LSMO and insulating phase powders, is strongly dependent on the mastery of efficient sintering processes. However, the co-sintering is hindered by differences in thermal stability, the rate and the onset temperature of shrinkage, and the physical and/or chemical compatibilities between components. It has been claimed that classical thermal treatments may induce the evolution of the manganese oxidation state [9], the grain growth [10], the segregation of the insulating phase [2,5,11,12] or the formation of an interphase between the two components (see supplementary material, part SM1). This difficult control of the microstructure and the formation of side phases may then have a detrimental influence on the magnetoconductive properties because the tunnel barrier structure

becomes more complex and new spin-nonconserving conducting channels (spin-flip, thermally activated) are contributing.

We propose here a strategy based, on the one hand, on grain boundary engineering to design 0–3 type nanocomposites starting from core@shell nanoparticles and, on the other hand, on the implementation of a low temperature sintering process (see Fig. 1). In this way, after sintering, LSMO cores are expected to be uniformly and discretely embedded in an insulating matrix of three-dimensional-connectivity (3D) that originates from the silica shell densification. Here, silica also ensures the confinement of defects at the interfaces and should impede the core growth. This complex multimaterial microstructure is expected to be regular and tailored with a simultaneous control of the amount and of the chemical nature of tunnel barriers [13,14]. Hydrothermal sintering has been chosen as the low temperature densification route. Among the water-assisted densification routes such as cold sintering process CSP [15–18], water-assisted flash [19] or SPS [20] processes, hydrothermal sintering offers the specificity to operate in a close system, which promotes the diffusion of chemical species, hence of the reactivity. In this process, as in the CSP, the dissolution occurs in the contact zone between particles while the precipitation operates at the less stressed surface of particles, *i.e.* the pore surface. The involved solvent also acts as a mass transport medium and enhances creep at the grain/grain interface to promote densification [21–23]. It may be noted that, during sintering, water remains liquid and is expelled by densification in specific spaces for water removal [21,23] while in Cold Sintering Process, the system is open and water evaporates. In the first case, the

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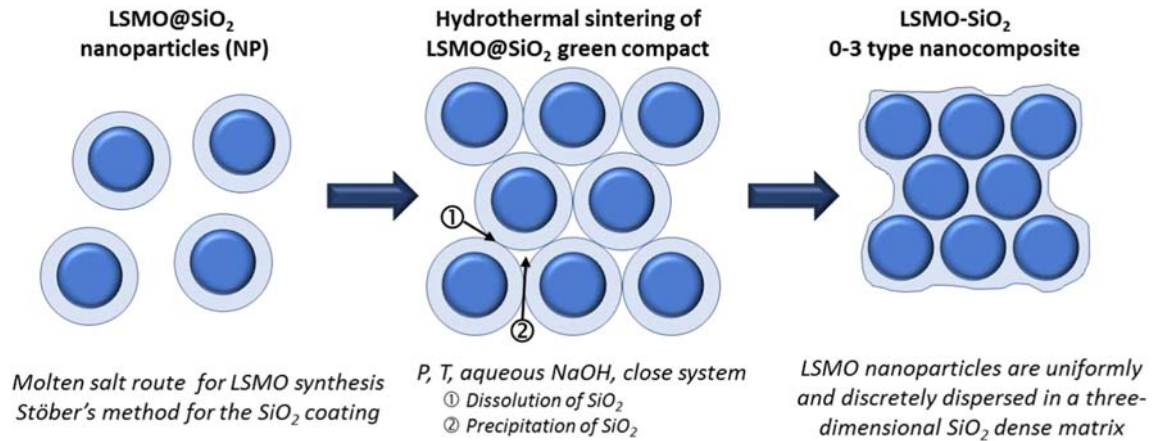


Fig. 1. Bottom up strategy deployed for the fabrication of 0-3 type composite.

process is rather thermodynamically controlled while in the second one it is mainly controlled by kinetics, the predominance of the elementary steps depending on the closed/open system.

In this way, hydrothermal sintering appears as a powerful tool to favor the densification of silica, the shell being in contact with the solvent, while the manganite cores and the manganite/silica interfaces remain totally inert. This process based on mechanical-chemical mechanisms thus should ensure conditions that comply with the characteristics of the specifications. The purpose of this paper is then to provide a proof of concept for this nanostructuring strategy.

The manganite $\text{La}_{0.66}\text{Sr}_{0.34}\text{MnO}_3$ (LSMO) nanoparticles with perovskite structure were synthesized using molten salt route. The silica coating of these as-obtained 30 nm LSMO nanoparticles was performed according to Stöber's method. The targeted shell thicknesses were 5 and 10 nm, as-obtained core@shell nanoparticles being LSMO@SiO₂(5) and LSMO@SiO₂(10), respectively. For each sample, a mixture of core@shell nanoparticles with 20 wt% of a 0.2 M aqueous sodium hydroxide solution was prepared and shaped in the form of green pellets with $\approx 55\%$ of compactness. The synthesis and characterization of nanoparticles (LSMO and LSMO@SiO₂) and the shaping of hydrated green compacts are described in the supplementary material, parts SM2 and SM3, respectively.

In the hydrothermal sintering process, a powder with water or a hydrated green compact is externally and mechanically compressed in an autoclave under hydrothermal conditions and over short periods of time. The solvent is expelled during densification and recovered in specific spaces for solvent retreat [21]. The experimental conditions implemented for the densification of composites were chosen identical to those we previously used for the densification of 50 nm silica nanoparticles, reaching 86% of relative density [23]. It was shown that the use of the mineralizer NaOH at low concentration promotes the dissolution of silica. In a first step, each hydrated green compact is subjected in the hydrothermal sintering device to a pressure increase up to 127 MPa in order to avoid the vaporization of water during the subsequent heating of the system. The temperature is then increased up to 300 °C with a heating rate of 10 °C/min. When the temperature reaches 150 °C, the pressure is increased, in 15 s, up to 350 MPa. The sample is maintained in the autoclave at 300 °C under 350 MPa for 90 min. Finally, the pressure is released as the system cools spontaneously. In this way, LSMO-SiO₂(5) and LSMO-SiO₂(10) composites are obtained from starting silica shell thickness of 5 and 10 nm, respectively. All characterization procedures are detailed in the supplementary material, part SM4.

For the thinnest silica shell (Fig. 2a), the as-obtained LSMO-SiO₂(5) composite exhibits a microstructure that consists of a cohesive assembly of well-faceted particles embedded in a vitreous matrix (Fig. 2b). One can observe highly dense blocks where LSMO

nanoparticles do not seem to be directly in contact with each other but whose morphologies are still distinct. Some small pores can be distinguished on the HR SEM image. Their nature is confirmed by the mercury intrusion porosimetry measurements that reveal the mesoporous nature of the composite (Fig. 2c) as the pore size ranges from 10 to 30 nm. Considering the basic mechanisms involved, the initial mesopores originating from the compact assembly of core@shell nanoparticles in the green material are gradually filled by precipitation. The residual presence of small mesopores in the composite thus reveals that the progression of densification has been incomplete. The relative density is evaluated at $83 \pm 5\%$. For the thickest silica shell (Fig. 2d), the microstructure of the composite LSMO-SiO₂(10) has drastically evolved (Fig. 2e). In this case, some larger pores are distributed in a highly dense homogeneous matrix. On the left bottom of Fig. 2e, one can observe that, as expected, LSMO cores are uniformly and discretely embedded in a three-dimensional-connectivity silica network. The mercury intrusion porosimetry measurements show that the material is predominantly macroporous with a maximum pore size of 200 nm. However, around 12 vol% of mesopores still contribute to the total porosity. The relative density is evaluated at $77 \pm 5\%$. We can conclude that, when the silica shell thickness increases, the densification mechanisms are more efficient as they promote the filling of the initial mesopores. However, this densification propagation involves also a rapid release of water that was accommodated in the microstructure via the formation of macropores, hence leading to a decrease of the relative density. One can reasonably formulate the hypothesis that the different microstructures of LSMO-SiO₂(5) and LSMO-SiO₂(10) originate from the difference of compressibility between the manganite and the silica. The compressibility strongly influences the stress gradient (induced by external uniaxial compression) within particles and consequently the activation of dissolution/precipitation phenomena [22]. The thicker the silica shell is, the more the core@shell particles densify similarly to pure silica nanoparticles, the formation of macropores facilitating the rapid water release in compressible materials, as we previously observed on 50 nm SiO₂ nanoparticles [22,23].

To unambiguously confirm the 0-3 type microstructure of the LSMO-SiO₂(5) composite, HR TEM characterization has been performed (Fig. 3a-b). The microstructure is homogeneous and shows that highly crystalline manganite nanoparticles are individually embedded in an amorphous matrix. The average distance between the crystallized cores equals 2–2.5 nm, far from the initial distance before sintering (10 nm). This impressive decrease of the *inter-core* distance is due (i) on the one hand, to the polycondensation of native silanol groups in the shells which induces a contraction of silica [22] and (ii) on the other hand, to the mass transport of silica from the *inter-grain* contact zones toward neighboring pores that are gradually filled by

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