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Manifestation of phase transitions in the crystal field spectra of  $[(CH_3)_2CHNH_3]_4Cd_3Cl_{10}:Cu$  crystals

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#### ACCEPTED MANUSCRIPT

# Manifestation of Phase Transitions in the Crystal Field Spectra of [(CH<sub>3</sub>)<sub>2</sub>CHNH<sub>3</sub>]<sub>4</sub>Cd<sub>3</sub>Cl<sub>10</sub>:Cu Crystals

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On the basis of absorption spectroscopy data the temperature evolution of the Cu<sup>2+</sup> ion environment in the crystal of [(CH<sub>3</sub>)<sub>2</sub>CHNH<sub>3</sub>]<sub>4</sub>Cd<sub>3</sub>Cl<sub>10</sub> doped with Cu<sup>2+</sup> (IPACC:Cu) was studied. As it follows from the temperature dependences of heat capacity and absorption coefficient at a constant wavelength IPACC:Cu crystals are characterised by I order phase transitions at  $T_1 = 358 \text{ K}$  and  $T_3 = 253 \text{ K}$  whereas the continuous changes observed at  $T_2 = 293$  K would be related to the first order phase transition close to the second order. For the detailed analysis of the crystal field spectra in the temperature range from  $T_3$  till  $T_1$  the special program package CrysTool 2.0 based on the quantum mechanical models first of all on the model of normalized spherical harmonics (NSH) has been employed. It has been found that similarly to the crystal of ((CH<sub>3</sub>)<sub>2</sub>CHNH<sub>3</sub>)<sub>4</sub>Cd<sub>3</sub>Cl<sub>10</sub> (IPACC) the investigated doped crystal contains the distorted octahedral metal-halogen complexes of two types and the degree of their distortion is considerable changed in vicinity of the phase transition at  $T_2 = 293$  K. The phase transitions at  $T_2$  and  $T_3$  were found to be shifted toward the low temperatures in comparison with initial IPACC crystal whereas the high temperature transition at  $T_1$  is shifted to the higher temperatures.

*Keywords:* Transition metal alloys and compounds; Optical materials; Optical spectroscopy; Crystal and ligand fields; Phase transitions.

#### 1 Introduction

Unusual structural architectures are formed in halogeno-cadmate (II) compounds. Owing to the chemical flexibility (the flexibility which accommodates different organic counterions as well as inorganic components) of these hybrid organic-inorganic materials, it

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