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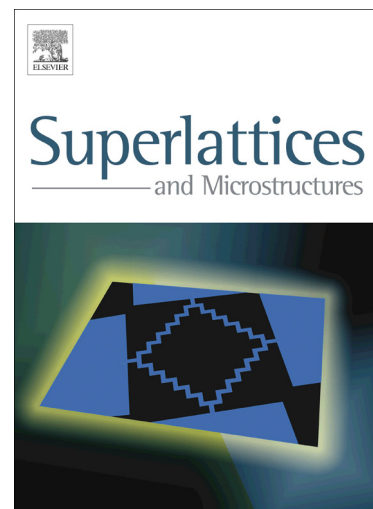
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# Enhanced Optical Properties of $W_{1-x}Mo_xO_3 \cdot 0.33H_2O$ Solid Solutions with Tunable Band Gaps

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## Abstract

A series of  $W_{1-x}Mo_xO_3 \cdot 0.33H_2O$  ( $x = 0, 0.25, 0.50, 0.75$ ) nano/microstructures and  $MoO_3 \cdot 0.55H_2O$  with elongated morphology were prepared by using hydrothermal technique. Aqueous acidified solutions of ammonium metatungstate hydrate ( $(NH_4)_6H_2W_{12}O_{40} \cdot xH_2O$ ) and ammonium heptamolybdate tetrahydrate ( $(NH_4)_6Mo_7O_{24} \cdot 4H_2O$ ) were hydrothermally reacted to yield the desired nano/microstructures. In the  $WO_3 \cdot 0.33H_2O$  crystal lattice can be substituted with up to 75 % Mo without structural alterations. When the Mo atoms increase, from 0 to 75 at. %, the band gap of the, as-prepared,  $W_{1-x}Mo_xO_3 \cdot 0.33H_2O$  material decreases from 2.55 to 2.15 eV. In order to corroborate experimental data, first-principle calculations using DFT and DFT+U framework were employed which revealed indirect semiconductors up to  $x=0.75$ . We suggest that the increase in the Mo fraction (25, 50 and 75%) by hydrothermal synthesis (pressure and temperature) is responsible for the narrowing of the band gap.

**Keywords:** oxide semiconductors, electron microscopy (STEM, TEM and SEM), visible and ultraviolet spectrometers, band-structure, crystallography, optical properties, band gap tunable.

## 1. Introduction

Hydrated transition metal oxides such as  $WO_3 \cdot nH_2O$  and  $MoO_3 \cdot nH_2O$  ( $n = 0, 0.33, 1$  or  $2$ ) have been studied extensively due to their special electronic and optoelectronic properties. These materials have enormous potential applications in the fields ranging from condensed-matter physics to solid-state chemistry [1], such as photo-electrochemical energy conversion [2], gas sensors [3], photo-catalysts [4], lithium-ion batteries [5], solar cells [6], electron emitters [7] and optical storage media [8]. Hydrated oxides, compared with their single metal oxide constituents ( $WO_3$  and  $MoO_3$ ), have potential application in electrochemical devices. In the case of solid solutions, molybdenum-tungsten oxide ( $W_{1-x}Mo_xO_3 \cdot 0.33H_2O$ ) materials show more promise due to the ability to control the components, structural characteristics tailoring, physical/chemical properties modulation. Furthermore, an improvement in the performance in the above-mentioned applications is expected due to the “synergistic effect” in the composites, when Mo is included in the lattice. Recently, tremendous effort has been dedicated to the preparation [9, 10, 11], formation, mechanism study [12, 13, 14], and property investigation [15, 16, 17, 18, 19, 20, 21, 22, 23, 24, 25, 26, 27, 28, 29] of  $W_{1-x}Mo_xO_3 \cdot 0.33H_2O$ . The  $W_{1-x}Mo_xO_3 \cdot 0.33H_2O$  showed

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