



# Van der Waals interactions and the properties of graphite and 2H-, 3R- and 1T-MoS<sub>2</sub>: A comparative study

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## ABSTRACT

We performed *ab initio* calculations based on density functional theory (DFT) in order to explore the influence of van der Waals (vdW) interactions on the structural, energetic, elastic and vibrational properties of graphite and MoS<sub>2</sub> (in the 2H, 1T and 3R polytypes). Different vdW functionals were tested and their performance were evaluated by the comparison with the available experimental data for graphite and 2H-MoS<sub>2</sub>. Several properties of the 1T- and 3R-MoS<sub>2</sub> phases were also predicted.

## 1. Introduction

Two-dimensional (2D) materials drew great attention in the past few years as possible new materials for electronics, among other applications. Up to now, the most studied 2D materials are graphene [1] and semiconducting transition metal dichalcogenides (TMDs), such as MoS<sub>2</sub> [2]. These materials and their heterostructures offer excellent mechanical flexibility, optical transparency, and favorable transport properties for developing electronic, sensing, and optical devices [3]. The intrinsic semiconducting nature of MoS<sub>2</sub> is a major advantage over graphene (which has no intrinsic band gap) as a two-dimensional (2D) channel material in field-effect transistors (FET). Also, being a direct gap material, 2D MoS<sub>2</sub> has great potential for application as optical material as well.

Despite the recent interest in their 2D forms, both systems are also extremely well-known in their traditional 3D, bulk form, i.e. graphite and molybdenite, and they share a few common properties [4]. They both have hexagonal and layered structures, in which the covalent-bonded layers are held together by weak van der Waals forces. Due to its high temperature stability, chemical inertness, high thermal conductivity, low thermal expansion and high ion-exchange capabilities, graphite is widely used in many industrial applications [5]. Also, molybdenite and other TMDs are extensively used in catalysis, energy storage, hydrogen generation, dehydrosulfurization and lubrication [6].

During the synthesis of MoS<sub>2</sub>, both hexagonal (2H) and rhombohedral (3R) polytype layered crystals, which present somewhat different physical properties, can be obtained. For example, the 3R

polytype of MoS<sub>2</sub>, due to its non-centrosymmetric structure, exhibits a valley polarization in photoluminescence emission even in the bulk form. The electronic properties of 3R phase remain largely unexplored and it is not possible to distinguish the polytype *in situ* experimentally. The 1T phase is produced by the oxidation of KMoS<sub>2</sub> or LiMoS<sub>2</sub> and is metastable, being easily converted to the disordered 2H phase after heating in the temperature of 100 °C [8–10]. Due to the complex nature of the layered TMD materials and the great interest in the electronics community for their applications in emerging devices, a more reliable, accurate, and atomistic treatment of the structural and electronic properties of TMD is desired [7].

Density functional theory (DFT) is the tool of choice for the structural investigation of various types of materials [11,12]. Although the chemical bonding between nearest neighbors determines the structure in small molecular systems, for molecular assemblies and for systems of greater sizes, the weak intermolecular forces start to play a more determining role in their stabilization. To cope with this challenge, a good level of theory should have at its disposal the appropriate total energy functionals which take into account the correct weak intermolecular interactions. However, popular density functionals such as the ones based on generalized gradient approximations (GGAs) are unable to describe the correct van der Waals interactions resulting from dynamical correlations between fluctuating charge distributions, therefore they are intrinsically unadapted to describe structural and energetic properties of molecular materials. Several routes have been proposed to circumvent this difficulty. One route is through the so-called “dispersion-corrected” (DFT-D) functionals [13], which consist of a pragmatic

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and inexpensive addition of a semiempirical dispersion potential to the conventional Kohn-Sham DFT energy. In this method, the van der Waals interactions are described via a simple pairwise force field optimized for popular DFT functionals, such as PBE, PW91 and B3LYP. Among numerous attempts to provide consistent parametrizations compatible with a given exchange-correlation functional, the DFT-D2 approach of Grimme seems to be one of the most successful ones [14]. It is noteworthy that a more recent Grimme-like approach exists, namely DFT-D3, although it has not been yet implemented in Quantum Espresso.

Another route is through the van der Waals density functionals (vdW-DF) [15], which are built by adding a non-local contribution to the electronic energy density-functional so as to describe dispersion interactions between electronic densities at arbitrary geometries. We employed vdW-DF1 [16], vdW-DF2-B86R, vdW-DF2-C09 [17,18] and vdW-DF-CX [19] implementations in this work [20].

In this work, after extensive testing of several approaches for the treatment of dispersion interactions for the well-studied graphite, 1T- and 2H-MoS<sub>2</sub> systems, we calculated and predicted properties of the 3R-MoS<sub>2</sub> and 1T-MoS<sub>2</sub> polytypes, for which there is little information available in the literature [23]. Our results indicate that the vdW-DF2 framework functionals have the best overall performance among the studied functionals for this class of systems.

## 2. Methodology

*Ab initio* calculations were performed in the framework of DFT using the Quantum Espresso package [21]. We used a variety of functionals for the exchange-correlation such as LDA [22], PW91 [24], PBE [25], revPBE [26], PBESOL [27]. The use of a large collection of XC functionals is important for the evaluation of different flavors of the GGA approximation. We used Vanderbilt ultrasoft pseudopotentials to describe the ion cores [28]. The Kohn-Sham orbitals were expanded in a plane wave basis set with a maximum kinetic energy of 60 Ry. Monkhorst-Pack [29] meshes of  $6 \times 6 \times 2$  k-point sampling in the first Brillouin Zone were used for the MoS<sub>2</sub> bulk in the polytypes 2H and 3R,  $7 \times 7 \times 5$  for the graphite bulk and  $3 \times 3 \times 3$  for the 1T polytype.

Dispersion interactions were included via the DFT-D2 [14], vdW-DF1 [16], vdW-DF-B86R, vdW-DF2-C09 [17,18] and vdW-DF-CX [19] approaches. In the case of DFT-D2, a semiempirical pairwise interatomic potential is added to the DFT converged total energy. This additional term results from the computation of the dispersion energy taking into account the periodic boundary conditions by the means of a real-space summation. Formally, the resulting total energy is then given by:

$$E_{DFT-D2} = E_{DFT} + E_{disp} \quad (1)$$

where  $E_{DFT}$  is the converged total energy with the chosen functional and  $E_{disp}$  is the contribution of the dispersion interactions computed within an interatomic pairwise approach. The term  $E_{disp}$  can be written as:

$$E_{disp} = -s_6 \sum_{i=1}^{N_{at}-1} \sum_{j=i+1}^{N_{at}} \frac{C_6^{ij}}{R_{ij}^6} f_{dmp}(R_{ij}) \quad (2)$$

where  $N_{at}$  is the total number of atoms,  $C_6^{ij}$  denotes the dispersion coefficient between the atomic sites  $i$  and  $j$ ,  $R_{ij}$  is the distance between atoms  $i$  and  $j$ ,  $s_6$  is a semiempirical parameter corresponding to an overall scaling of dispersion interactions and, to avoid singularities between very close neighbors when  $R_{ij}$  is very small, a damping function  $f_{dmp}$  is used. In the present work, the  $s_6$  parameter was fitted for PBE and PW91 with the experimental values of the parameter  $c$ . Therefore, the predicted properties from this approach were all computed by these fitted parameters. For graphite, the optimal values are 0.45 and 0.50 for PW91-D2 and PBE-D2, respectively. In the case of the 2H-, 3R- and 1T-MoS<sub>2</sub>, the optimized values are 0.95 for both PBE-D2 and PW91-D2 for 3R- and 1T-, 0.90 for PW91-D2 and 0.95 for PBE-D2

for 2H. For the sake of simplicity, “-D2” will be appended to the name of the exchange-correlation functional in order to state that the DFT-D2 approach was applied.

In the case of the vdW-DF1, vdW-DF2-B86R, vdW-DF2-C09 and vdW-DF-CX approaches, dispersion interactions are computed self-consistently from the electronic density. The exchange-correlation functional framework is written as:

$$E_{xc}[n] = E_x^{GGA} + E_c^{LDA} + E_c^{nl}, \quad (3)$$

where  $E_c^{LDA}$  is the local-density approximation correlation functional,  $E_x^{GGA}$  is the semilocal exchange energy and  $E_c^{nl}$  is the non-local dispersion, generally written as

$$E_c^{nl}[n] = \frac{1}{2} \iint d^3r_1 d^3r_2 n(\vec{r}_1) n(\vec{r}_2) \phi(q_1, q_2, r_{12}), \quad (4)$$

where  $r_{12} = |\vec{r}_1 - \vec{r}_2|$ ,  $q_i$  is the inverse length scales at positions  $\vec{r}_i$  and  $\phi$  is the dispersion kernel.

All structures were found by minimizing the total energy and the atomic forces with the convergence criteria of  $10^{-4}$  Ry and  $10^{-3}$  Ry/Bohr, respectively.

The exfoliation energies of systems were determined by the following equation

$$E_{exf} = E_{monolayer} - E_{bulk} \quad (5)$$

where  $E_{monolayer}$  and  $E_{bulk}$  are the total energies of the monolayer and the bulk systems, respectively. The bulk energy was normalized by the number of the layers per unit cell. Cohesive energies were calculated as

$$E_{coh} = \sum_i E_i - E_{bulk} \quad (6)$$

where  $E_{bulk}$  and  $E_i$  are the total energies of the bulk and of a single atom  $i$  in vacuum.

Bulk moduli were calculated by fitting  $E(V)$  curves to the Murnaghan's equation of state:

$$E(V) = E_0 + V \left( \frac{B}{B'} \right) \left[ \left( \frac{V_0}{V} \right)^{\frac{B'}{B'-1}} + 1 \right] - \frac{BV_0}{(B'-1)} \quad (7)$$

where  $E_0$  is the minimum energy,  $V_0$  is the equilibrium volume,  $B$  is the bulk modulus, and  $B' = \frac{dB}{dP}$  is the pressure derivative of the bulk modulus.

Of particular interest in the evaluation of vdW functionals is the calculation of the elastic constant  $C_{33}$

$$C_{33} = \frac{2c_0}{\sqrt{3}a_0^2} \frac{\partial^2 E}{\partial c^2}, \quad (8)$$

since it involves the relative displacements between vdW-interacting rigid layers. In the equation above,  $a_0$  and  $c_0$  are in-plane and out-of-plane lattice constants at equilibrium. Closely related to  $C_{33}$  is the frequency  $\omega_{LB}$  (in  $\text{cm}^{-1}$ ) of the interlayer-breathing (or ZO') phonon mode:  $\omega_{LB} = \sqrt{\frac{C_{33}}{\mu \pi^2 v_t^2}}$  [30], where  $\mu$  is the area mass density for the monolayer,  $v$  is the light speed (in cm/s) and  $t$  is the interlayer distance.

## 3. Results and discussion

### 3.1. Structural analysis

Table 1 presents the main unit cell parameters obtained by the different approaches. The largest deviations with respect to the experimental values occur for the  $c$  parameter, certainly because of the poor treatment of the dispersion interactions for some functionals. For graphite, LDA and vdW-DF2 framework present the lowest relative errors, whereas the values obtained with PBE, PW91 and revPBE present the highest deviations from the experimental values. Regarding the dispersion corrections, the DFT-D2 approach tends to overbind the graphite layers. It should be noted that the good structural description

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