



Microstructure, mechanical properties and strengthening mechanisms of Mg matrix composites reinforced with in situ nanosized TiB₂ particles



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ABSTRACT

In this work, Mg matrix composites reinforced with 2.5 wt% nanosized TiB₂ particles are fabricated by adding Al-TiB₂ master alloy prepared by the chemical reaction of Al-K₂TiF₆-KBF₄ system into molten magnesium. The size distribution of TiB₂ particles extracted from the master alloy and their effects on the microstructure and mechanical properties of AZ91 matrix are investigated. The results show that the TiB₂ particles display a dominant number of particles less than 100 nm and a uniform distribution in the composites. Also, the grain size of composites is decreased compared with AZ91 alloy. The high resolution TEM reveals that the interface between nanosized TiB₂ particles and matrix is clear and bonds well. Besides, the yield strength, ultimate tensile strength and elongation to fracture of composites are improved by 49.4%, 25.5% and 51.1% respectively. Among the strengthening mechanisms of nanosized particles reinforced Mg matrix composites, dislocation strengthening and Orowan strengthening play a more important role in increasing the strength. It is noteworthy that the elongation to fracture of the composites is higher than that of AZ91 matrix, which is contrary to other works.

1. Introduction

Magnesium alloys as the lightest structure material are potential candidates in the aerospace and automotive industries due to their low density, good damping capacity, high specific strength and excellent castability [1–3]. However, low strength and ductility limit wide application of magnesium alloys [1]. In recent years, particulate reinforced magnesium matrix composites (PRMMCs) have attracted increasing attentions because of their excellent properties, such as high specific strength and modulus, low density as well as good wear resistance [4,5]. In general, the reinforcements such as Al₂O₃ [6,7], Y₂O₃ [8,9], SiC [10–12], CNT [13] and TiC [14] are used to fabricate Mg matrix composites by disintegrated melt deposition, stir casting and in situ casting methods. Among these methods, in situ casting shows a great potential in PRMMCs due to several advantages, such as good thermodynamic stability and clean interface between the reinforcement and the matrix [15].

In situ micro particles reinforced Mg matrix composites exhibit a considerably improved strength and wear resistance, but a significantly reduced ductility compared with the unreinforced Mg matrix [14,16–20]. In other words, micro particles often lead to an increase in strength of PRMMCs while ductility would be sacrificed. Some reports have shown that nanosized particles reinforced metal matrix

composites can achieve excellent mechanical properties with a combination of higher strength and ductility [8,21–24]. Yang et al. [25] obtained in situ nano sized AlN/Mg composites with higher ultimate tensile strength and fracture elongation (16–47%) developed via bubbling nitrogen gas into molten Mg-Al alloy. However, most of the reports in literature are mainly focused on micro sized particles reinforced Mg matrix composites, and only a few works study the effect of in situ nanosized reinforcement on the microstructure and mechanical properties of Mg matrix composites. Moreover, TiB₂ ceramic particles are regarded as suitable reinforcement candidates due to its low density (4.5 g/cm³), high elastic modulus (565 GPa), good wear resistance and the same crystal structure with magnesium [26]. In previous reports, nanosized TiB₂ particles reinforced Al matrix composites can be synthesized by the chemical reaction of Al-K₂TiF₆-KBF₄ system [27–29]. To the best of our knowledge, few open literatures have been carried out on in situ nanosized TiB₂/Mg composites. Therefore, it is essential to study the microstructure, interface characteristic, mechanical properties and strengthening mechanisms of in situ nanosized TiB₂/Mg composites.

In present study, the size distribution of TiB₂ particles extracted from Al-20 wt% TiB₂ master alloy prepared by the chemical reaction of Al-K₂TiF₆-KBF₄ system at 900 °C is characterized quantitatively. And then in situ TiB₂/AZ91 composites are fabricated by adding Al-20 wt%

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Table 1
Chemical composition of the AZ91 matrix alloy.

Element	Al	Zn	Mn	Mg
wt%	9.45	0.76	0.27	Balance

TiB₂ master alloy into molten magnesium assisted semisolid stirring technique. Moreover, the effect of TiB₂ particles on microstructure and mechanical properties of Mg matrix composites and strengthening mechanisms are discussed in detail.

2. Experimental procedure

2.1. Processing of in situ TiB₂/AZ91 composites

In this study, pure Mg, Al, Zn and Mg-5%Mn master alloy were used to prepare AZ91 matrix alloy, and the chemical composition of AZ91 alloy was given in Table 1. And 2.5 wt% TiB₂/AZ91 composites were fabricated by in situ mixed salts method with two steps. First, Al-20 wt % TiB₂ master alloy was prepared according to the chemical reaction of Al-K₂TiF₆-KBF₄ system, as given in equal (1) [27]. The Gibbs free energy of this reaction could be calculated as $\Delta G^\circ = -981990 + 112.87T$ using the thermodynamic data from literatures [30,31]. In order to accelerate the reaction rate, the temperature was selected as 900 °C with $\Delta G_{900^\circ\text{C}}^\circ = -849.68\text{ kJ/mol}$, which means the reaction can take place. Two types salts of K₂TiF₆ and KBF₄ with a high purity of 99.5% were mixed well with a molar ratio of Ti/B = 1/2 and the mass of salts corresponded to the composition of Al-20 wt% TiB₂. When the melt temperature reached 900 °C, the mixed salts were slowly added into the molten Al in a graphite crucible in an electrical resistance furnace. The Al melt was stirred at 100 rpm for 5 min using a zircon coated graphite rod every 10 min. After 60 min, the slags floated on the top layer of the Al melt was removed and then the melt was poured into a graphite mould.



Second, about 1000 g Mg, Zn and Mg-5%Mn master alloy were first melted at 750 °C and held at this temperature for 15 min in the mild steel crucible in an electric resistance furnace under an argon protective atmosphere. Subsequently, the Al-TiB₂ master alloy was added into the Mg melt. The melt was kept at 750 °C until the Al-TiB₂ master alloy was dissolved. After the melt was cooled to the semi-solid condition, mechanical stirring was carried out at 300 rpm for 15 min by a stainless-steel stirrer to obtain a uniform distribution of TiB₂ particles in Mg matrix. Then, the composites melt was heated rapidly to 720 °C and casted into a preheated steel mould. In order to compare with the composites, the AZ91 matrix was also prepared by the same method. At last, the matrix alloy and composites were homogenized heat treatment at 410 °C for 24 h followed by quenching in warm water.

2.2. Extraction experiment for obtaining TiB₂ particles

The size distribution and actual weight percentage of TiB₂ particles in Al-TiB₂ master alloy were analyzed by chemical extraction experiment. A small ingot about 10 g cleaned by SiC sand paper was completely dissolved in 10 vol% HCl solution in a breaker for several hours. After dissolution, all particles were deposited at the bottom of the breaker, subsequently the solution of the top was poured out. Residual TiB₂ particles were washed thoroughly for several times until the supernatant reached neutral PH and finally was vacuum-dried at 60 °C for 8 h.

2.3. Mechanical properties test

The mechanical properties of specimens were studied by conducting

Brinell hardness and tensile tests. The hardness test of materials was carried out in Brinell hardness testing at 250 kgf load for a dwell time of 30 s. The tensile test was conducted at room temperature at the mechanical testing machine (MTS, CMT5305, USA) with a tensile rate of 0.5 mm/min, and three specimens with dimensions of 15 mm length, 4 mm width and 2 mm thickness in the gauge section were tested to ensure the reproducibility of data according to the ASTM-E8 standard.

2.4. Material characterization

Phase identification was carried out by X-Ray Diffraction (XRD, Bruker D8 Advance) using Cu K_α radiation ($\lambda = 1.54\text{Å}$) at a scan rate of 0.2°/s. The size distributions of TiB₂ particles extracted from Al-TiB₂ master alloy were measured using Malvern Zetasizer Nano ZS, which could measure particle size ranging from 0.3 nm to 10 μm. Twelve measurements were conducted for TiB₂ particles diluted with ethanol. Actual weight percentage of TiB₂ particles in Al-20 wt% TiB₂ master alloy was measured for three times to obtain an average value.

Microstructure analysis was carried out using optical microscopy (OM), scanning electron microscopy (SEM), field emission scanning electron microscope (FESEM, Zeiss Gemini500) equipped with energy dispersive spectroscopy (EDS) and transmission electron microscopy (JEM-2100Plus). The specimens for OM and SEM analysis were grinded, polished and etched with 5% nitric acid. The equivalent grain size distribution was measured by Image-Pro Plus software. Quantitative estimations of the volume fraction of TiB₂ particles in the TiB₂/Mg composites were statistically analyzed using 20 different SEM micrographs by Image J software. The morphology of fractured surface of tensile specimen was also analyzed by OM and SEM. The TEM specimens were prepared by mechanical polishing to 50 μm thickness, punching 3 mm diameter discs and then ion-thinned.

3. Results and discussion

3.1. Al-TiB₂ master alloy

Fig. 1(a) shows XRD pattern of Al-TiB₂ master alloy. The existence of diffraction peaks of TiB₂ demonstrates that only TiB₂ phase is synthesized in Al-TiB₂ master alloy. It is also observed that diffraction peaks of intermediate product Al₃Ti or AlB₂ can not be detected, indicating completion of reaction (1). This is in agreement with the results [32] that TiB₂ is the most thermodynamically stable phase due to its lowest Gibbs free energy among these three compounds. Fig. 1(b) shows the SEM micrograph of Al-TiB₂ master alloy. It is clear that TiB₂ particles are mainly distributed along the boundaries of α-Al grains. However, some agglomerations of tiny TiB₂ particles are found as shown in Fig. 1(b).

3.2. Extracted TiB₂ particles

TiB₂ particles were extracted for a further study on morphology and size distribution as well as actual weight percentage of TiB₂ particles in Al-TiB₂ master alloy. Fig. 2(a) shows XRD pattern of extracted TiB₂ particles from Al-TiB₂ master alloy. Only TiB₂ phase peaks are detected, suggesting that there is no other intermetallic compound in Al-TiB₂ master alloy. In addition, the XRD result further confirms the purity of extracted TiB₂ particles, which indicates that the interference of other phases can be eliminated for the following calculation of actual weight percentage and the size distributions of TiB₂ particles in the master alloy. Fig. 2(b) presents FESEM morphology of TiB₂ particles extracted from Al-TiB₂ master alloy. As seen in Fig. 2(b), the particles mainly display rectangular and hexagonal shapes. It is worth to note that the morphology of some tiny particles is not typical hexagonal maybe due to the uncompleted growth for TiB₂ crystal [33]. Moreover, the actual weight percentage of TiB₂ particles in Al-TiB₂ master alloy is about 19.6 wt%.

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