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ABSTRACT

Defective black titania (TiO₂) was synthesized by the solution plasma technique at ambient temperature and pressure. The effects of the electrolyte solution medium type (KCl and HNO₃) and concentration (0.3 and 3.0 mM) as well as the plasma discharge time (1-4 h) with a Ti electrode were investigated. The 3.0 mM HNO₃ solution provided the highest energy per second discharging into the plasma, resulting in both a high synthesis rate of black TiO₂ and a high degree of defective structures, as monitored in terms of the Ti^{3+}/Ti^{4+} ratio, which can shorten the band gap energy (E_g) of the obtained black TiO₂. A long plasma discharge time (4 h) induced the formation of large particles of black TiO₂, which appeared as a highly defective structure. Overall, the black TiO₂ prepared by discharged plasma for 4 h in 3.0 mM HNO₃ solution (BT_{N,304}) provided the highest photocatalytic activity for glycerol

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