



## Nanostructured TiO<sub>2</sub> thin films sensitized by CeO<sub>2</sub> as an Inexpensive Photoanode for enhanced photoactivity of water oxidation

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### Abstract:

Nanostructured CeO<sub>2</sub>/TiO<sub>2</sub> bilayer thin films were fabricated by sequential deposition of CeO<sub>2</sub> on TiO<sub>2</sub> with varying the thickness of CeO<sub>2</sub> using sol-gel spin-coating method. All samples were characterized by XRD, SEM, and UV-Visible absorption spectroscopy with an inspiration to use them as photoanode in photoelectrochemical splitting of water. The CeO<sub>2</sub>/TiO<sub>2</sub> bilayer with the optimized thickness of CeO<sub>2</sub> and TiO<sub>2</sub> showed enhanced photoelectrochemical response of 2.1 mA cm<sup>-2</sup> at 0.95V/Ag/AgCl as compared to pristine CeO<sub>2</sub> and TiO<sub>2</sub>. The improved photocurrent density of CeO<sub>2</sub>/TiO<sub>2</sub> bilayer probably due to rapid transfer of charge carriers or electrical gradient at CeO<sub>2</sub>/TiO<sub>2</sub> interface. This scalable and cost effective work with stable PEC performance of CeO<sub>2</sub>/TiO<sub>2</sub> also shows the potential for realistic PEC applications.

**Keywords:** CeO<sub>2</sub>/TiO<sub>2</sub>, Bilayer, Photoelectrochemical response, Cost effective, Spin Coating.

### Introduction:

Solar hydrogen generation by solar water splitting in PEC cells is the most forward-looking method in the recent time on account of hydrogen being widely accepted as clean energy carrier [1-2]. This solar-to-hydrogen conversion process initiated from the first PEC water splitting response of titanium dioxide (TiO<sub>2</sub>) by Honda and Fujishima in 1972 [3]. Nowadays, TiO<sub>2</sub> is well known semiconductor as photoanode due to its suitable band positions, abundant, non-toxic, and stability. However, previous studies on TiO<sub>2</sub> have found that its wide band gap of 3.2 eV absorbs only UV part of the solar spectrum and high recombination rate of photogenerated electron-hole pairs due to its short diffusion length of charge carriers are main drawbacks that limited the photocatalytic activity and visible light response of TiO<sub>2</sub> [4-5]. In this regards, extensive surveys have been carried out on refining the photoelectrochemical response of TiO<sub>2</sub>. Among the numerous techniques coupling of TiO<sub>2</sub> with other low band gap semiconductor with matching band edges positions will be potential method to improve the utilization of visible light and facilitation of photo-generated electrons

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