

Accepted Manuscript

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PII: S0167-577X(17)31804-9
DOI: <https://doi.org/10.1016/j.matlet.2017.12.045>
Reference: MLBLUE 23543

To appear in: *Materials Letters*

Received Date: 7 June 2017
Revised Date: 9 November 2017
Accepted Date: 10 December 2017

Please cite this article as: N. Haberkorn, S. Bengio, S. Suárez, P.D. Pérez, M. Sirena, J. Guimpel, Effect of the nitrogen-argon gas mixtures on the superconductivity properties of reactively sputtered molybdenum nitride thin films, *Materials Letters* (2017), doi: <https://doi.org/10.1016/j.matlet.2017.12.045>

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Effect of the nitrogen-argon gas mixtures on the superconductivity properties of reactively sputtered molybdenum nitride thin films

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Abstract

We report on the superconducting properties of nanocrystalline molybdenum nitride thin films grown by reactive DC sputtering at room temperature with a N₂:Ar mixture. Thin films grown using 5 % N₂ concentration display $T_c = 8$ K, which is gradually reduced to 5.8 K for 30 % N₂ concentration, producing changes in nitrogen stoichiometry of the samples from Mo₂N to Mo₂N_{1+x} (0 ≤ x < 0.4). The T_c is abruptly reduced and disappears for N₂ concentration between 30 % and 40 %, which can be attributed to an increment in the disorder due to phase coexistence between cubic γ -Mo₂N and non-superconducting amorphous MoN (dominant for N₂ concentration > 40 %).

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Keywords: nitrides; sputtering; superconductivity.

1. Introduction

Transition-metal nitrides (TMN) display a wide range of electronic and mechanical properties which are promising for technological applications. Superconducting TMN are potential candidates in a wide range of cryogenic devices like tunnel junctions [1] and electromagnetic radiation detectors [2]. The Mo nitrides present several superconducting crystalline phases: γ -Mo₂N (cubic) with $T_c \sim 5$ K [3], β -Mo₂N (tetragonal) with $T_c \sim 5$ K [4] and δ -MoN (hexagonal) with $T_c \sim 12$ K [5]. Different methods have been used in the growth of Mo nitride thin films, such as reactive sputtering [6,7], pulsed laser deposition [8], thermal nitration [9] and chemical routes [10]. A distinctive feature of γ -Mo₂N thin films is the influence of the disorder on T_c , which ranges from 4.5 K to around 8 K for epitaxial and polycrystalline thin films, respectively [11,12].

In this letter, we show that the T_c in γ -Mo₂N_x thin films (grown by reactive DC sputtering at room temperature) can be tuned by modifying the N₂:Ar mixture used during the sputtering process. T_c in thin films can be modified from 8 K to temperatures below 3 K by increasing the N₂ partial pressure in the N₂:Ar mixture from 5% to 40% of the total pressure. This modification can be associated with changes in the

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