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Proton beam flux dependent work function of mono-layer MoS₂

Sangwoo Kwon^a, Soo Ho Choi^a, You Joong Kim^a, Im Taek Yoon^b, Woorchul Yang^{a,*}

^a Department of Physics, Dongguk University, Seoul 04620, Republic of Korea

^b Quantum Functional Semiconductor Research Center, Dongguk University, Seoul 04620, Republic of Korea

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ABSTRACT

Monolayer (ML)-molybdenum disulfide (MoS₂) with a direct band gap of ~1.8 eV exhibits considerable potential for advanced electronic and optical device applications. The surface electronic properties of ML-MoS₂ need to be controlled for developing novel MoS₂-based devices. In this study, we investigated the work function variation of chemical vapor deposition-grown ML-MoS₂ that was controlled by proton irradiation. The crystallinity of the ML-MoS₂ was confirmed by micro-Raman and Photoluminescence measurements. The work functions of the ML-MoS₂ irradiated with varying proton beam flux were measured by Kelvin probe force microscopy. As the ML-MoS₂ were exposed to the proton beam flux ranging from 1×10^{12} to 1×10^{14} protons/cm² at the same beam energy of 10 MeV, the contact potential difference of the MoS₂ increased up to about 0.108 V with increased proton beam flux. Based on the referenced value of the Au work function (~5.1 eV), the work functions of non-treated ML-MoS₂ and proton-irradiated ML-MoS₂ with a proton beam flux of 1×10^{14} protons/cm² were determined to be 5.031 eV and 4.992 eV, respectively. The decrease of the work function of the MoS₂ with increased proton beam flux is due to the defect formation induced by proton irradiation. We suggest the possibility of engineering the surface potential and electronic properties of ML-MoS₂ through controlling proton irradiation conditions.

1. Introduction

Two-dimensional (2D) transition metal dichalcogenides (TMDCs), which are semiconductors of the MX₂ type (M: a transition metal atom, X: a chalcogen atom) [1,2], are layered materials that are widely studied for harvesting fundamental science and emergent applications in nanoelectronics and optoelectronics. Molybdenum disulfide (MoS₂) is a prototypical TMDC material [3]. Monolayer MoS₂ (ML-MoS₂) is composed of three-atomic-thick S-Mo-S stacks. Unlike ML-MoS₂ with a direct bandgap (~1.8 eV), the bulk MoS₂ has an indirect bandgap of ~1.23 eV [4]. Moreover, chemical vapor deposition (CVD) methods for the large-area growth of ML-MoS₂ have recently been optimized [5,6].

The variation of the electrical properties in MoS₂ has already been studied for the layer number of MoS₂. However, focus needs to be placed on engineering the material properties in ML-MoS₂ through external treatments. One of the many methods used for changing the electrical property of 2D materials is to control defect-sites [7]. For example, M. R. Islam et al. [8] demonstrated that the electrical property of ML-MoS₂ can be significantly tuned from the semiconducting to the insulating regime via defect engineering by oxygen plasma. ML-MoS₂ has two types of atomic defects: sulfur (S)-vacancy and molybdenum (Mo)-vacancy. The formation energy of S-vacancy is lower than that of

Mo-vacancy. Thus, S vacancies are energetically more favorable than Mo vacancies [9]. The intrinsic defects can be produced unintentionally in mechanically exfoliated or CVD-grown MoS₂. However, it is not a repetitive phenomenon. We therefore focused on employing the proton beam irradiated with the high proton energy to intentionally produce atomic-defects in ML-MoS₂. Several experimental studies have been carried out on the characterization of 2D materials related to the defects produced by proton beam irradiation. For instance, S. Mathew et al. [10] studied the effect of proton beam irradiation on the layer number of graphenes and substrates. They also reported that proton-irradiated MoS₂ exhibited ferrimagnetic behavior [11]. While the electrical properties of field effect transistor devices based on proton-irradiated multilayer MoS₂ were investigated by Kim et al. [12]. However, no fundamental studies have been carried out on the variation of the surface potential in proton-irradiated ML-MoS₂. This study focused on the correlation of the surface potential of ML-MoS₂ with defective-sites produced by proton ions.

In this study, we prepared ML-MoS₂ samples exposed to various proton beam fluxes and measured the variation in work function by Kelvin probe force microscopy (KPFM) to analyze the modification of electronic properties related to the defective-sites on ML-MoS₂. We discuss the electronic band structure of ML-MoS₂ before and after

* Corresponding author.

E-mail address: wyang@dongguk.edu (W. Yang).

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proton irradiation with a varying proton flux. We confirmed that the results of the KPFM measurements are consistent with those obtained from the first-principles (density functional theory) calculation.

2. Experiment details

ML-MoS₂ was grown in a CVD furnace with a 2 in. quartz tube including zone 1 and zone 2 of the furnace. Zone 1 was used for evaporation of S powder and zone 2 was used for evaporation of the Mo precursor and growth of MoS₂ on the substrate. The distance between zone 1 and zone 2 is 35 cm. The Si substrate with 300 nm thick SiO₂ was prepared by treatment with piranha solution (H₂SO₄:H₂O₂ = 3:1) for converting hydrophobic surface to hydrophilic surface. Mo precursor (ammonium molybdate tetrahydrate solution, (NH₄)₆Mo₇O₂₄·4(H₂O)) of 4 ml was coated on the substrate via spin-coating. The substrate was then placed in the center of zone 2 while S powder was loaded on the center of zone 1. Prior to growth, the system was purged using high purity argon gas (99.999%) at 500 sccm for 10 min to remove residual gases. The temperatures of zone 1 and zone 2 were elevated from room temperature to 350 °C and 800 °C for 15 min, respectively. During growth, the temperature was maintained for 10 min. After growth, the furnace was rapidly cooled to room temperature. An argon flow rate of 500 sccm at atmospheric pressure was maintained throughout the entire growth process. Fig. 1(a) shows that the MoS₂ flakes are well-grown on the SiO₂/Si substrate. First, to confirm that the as-grown MoS₂ flakes are a monolayer structure, optical microscopy (Nikon LV-IM, Nikon) and micro-Raman and photoluminescence (PL) spectroscopy (XperRAM100, Nanobase Inc., Korea) were used with a 532 nm laser excitation (average power of < 0.5 mW; exposure time of 1 s) at room temperature in ambient condition. After the characterization of the ML-MoS₂, the flakes were transferred onto the Au substrate by conventional transfer method. Considering the proton beam spot size (~10 mm), the 5 × 5 mm² sized ML-MoS₂ was transferred onto the center of the 1 × 1 cm² Au substrate. The samples were then annealed at 350 °C under atmosphere for KPFM (N8-NEOS, BRUKER Inc., Germany) measurement.

The proton irradiation of the ML-MoS₂ samples was conducted on a proton beam line in a MC-50 cyclotron at the Korea Institute of Radiological and Medical Sciences (KIRAMS). The employed energy and current of the proton beam were fixed at 10 MeV and 10 nA, respectively. The proton irradiation times were 10, 101, and 1019 s, which correspond to the proton beam fluxes of 1 × 10¹², 1 × 10¹³, and 1 × 10¹⁴ ions/cm² exposed to the ML-MoS₂/Au samples, respectively. The prepared samples are positioned at the center of the beam line and all of our experiments were carried out at room temperature.

The work function of the samples before and after proton beam irradiation was measured by using the KPFM technique. KPFM is an operating mode available in the atomic force microscopy (AFM) systems, and was used for measuring the variation in the local surface potential on the surface by employing a conducting AFM tip coated

with Cr/Pt (Multi-75, Budgetsensors. Inc., Bulgaria). The measurement of the surface potential using KPFM is generally conducted by the interaction between the AFM tip and sample surface. In the KPFM measurement, the distinct work function between the sample and the substrate can be determined from the Contact Potential Difference (V_{cpd}). The contrast of local brightness in the KPFM image can be used to obtain V_{cpd}, which is defined as follows [13,14], V_{cpd} = (φ_{tip} - φ_{sample})/(-e) where φ_{tip} and φ_{sample} are the work functions of the AFM tip and the sample, respectively, and e is the electronic charge. When the tip and sample make contact with each other, the electrons move from the material with a low work function to that with a high work function until the Fermi levels of all materials are equal. The result voltage difference is referred to as V_{cpd}.

3. Results and discussion

The optical properties of CVD-grown MoS₂ were investigated by micro-Raman and PL spectroscopy as shown in Fig. 1. The optical microscopy image showed that high-quality MoS₂ flakes were grown on the SiO₂ substrate, as shown in Fig. 1 (a). The triangular shaped MoS₂ flake was selected for investigation of optical properties. Two typical Raman-active phonon modes of E_{2g}¹ (centered at ~384 cm⁻¹) and A_{1g} (centered at ~403 cm⁻¹) were observed and the frequency spacing between the E_{2g}¹ and A_{1g} peaks was about 19 cm⁻¹, as shown in Fig. 1(b). Our Raman results were consistent with the Raman results of ML-MoS₂ reported from other groups [15]. Also, the bandgap of the MoS₂ flake was determined to be ~1.86 eV through PL spectrum measurement (Fig. 1(c)), which indicates a monolayer MoS₂ [5,6]. It was thus confirmed that the CVD-grown MoS₂ flakes have a monolayer structure.

Before the AFM measurements, the number of protons irradiated on the few nano-sized areas of ML-MoS₂ was first quantitatively calculated. In the top view of ML-MoS₂ shown in Fig. 2, the distance between the two atoms (Mo or S) is about 3.122 Å [16]. The proton flux of 1 × 10¹⁴ protons/cm² on a substrate with 1 × 1 cm² scale corresponds to the irradiation of a proton on the surface of the 3 × 3 unit cell (dashed line in Fig. 2) of ML-MoS₂, as shown in Fig. 2. After proton irradiation with a proton flux of 1 × 10¹², 1 × 10¹³ and 1 × 10¹⁴ protons/cm² on each ML-MoS₂/Au substrate sample, we conducted AFM measurements of the ML-MoS₂ flakes with similar shapes and sizes selected by optical microscopy. Considering the proton beam size (~1 cm), the entire surface (5 × 5 mm²) of the ML-MoS₂ was affected by the protons.

Fig. 3 shows the AFM morphology and KPFM images of the ML-MoS₂/Au samples before and after proton beam irradiation with proton flux of 1 × 10¹², 1 × 10¹³, and 1 × 10¹⁴ protons/cm². For all morphologies of the samples (Fig. 3(a)–(d)), the ML-MoS₂ are easily distinguished from the Au substrate. The region of the relatively flat surface exhibits the ML-MoS₂ while the region of the rough surface indicates the Au substrate. After transferring ML-MoS₂ flakes onto the Au substrate, an annealing process was performed to stabilize the

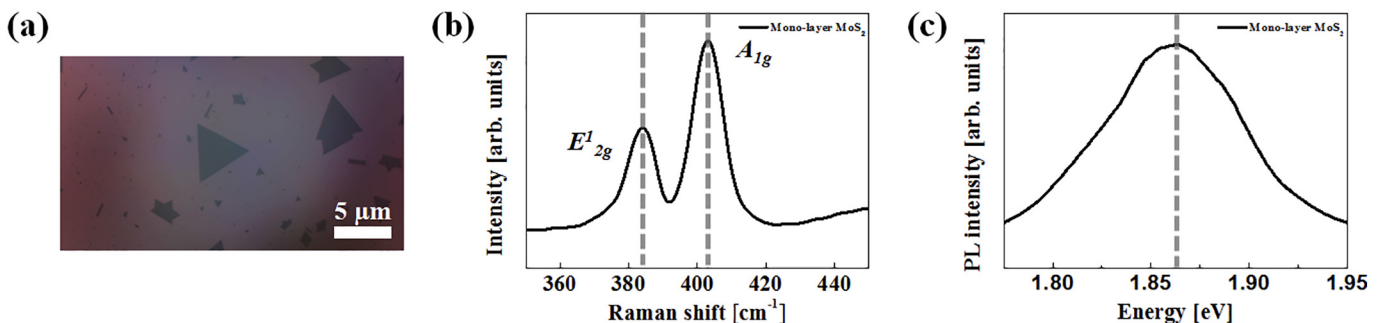


Fig. 1. (a) Optical microscopy image of ML-MoS₂: high-quality ML-MoS₂ flakes are grown on the SiO₂/Si substrate. (b) Micro-Raman of ML-MoS₂: E_{2g}¹ and A_{1g} peaks are centered at 384.0 cm⁻¹ and 402.8 cm⁻¹, respectively. (c) PL spectrum of ML-MoS₂: the bandgap of ML-MoS₂ is 1.86 eV.

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