



A low cost preparation of VO₂ thin films with improved thermochromic properties from a solution-based process

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ARTICLE INFO

Available online 11 January 2013

Keywords:

VO₂ thin films
Solution-based route
Annealing temperatures
Thermochromic property
W-doped
Phase transition temperature

ABSTRACT

This paper describes a solution-based route to synthesize vanadium dioxide (VO₂) thermochromic thin films on glass substrate by spin-coating technology followed by nitrogen-annealing with vanadium pentoxide (V₂O₅) and oxalic acid (H₂C₂O₄) as source material, which is fairly economical and practical. Surface morphologies indicate that the films obtained by this method are homogeneous and particulate, irregular prisms emerge as the annealing temperatures increase. X-ray diffractions show that films annealed at relatively low temperature are pure monoclinic phase with a preferred orientation of (011). NaV₄O₇ and NaV₆O₁₅ form along with raising the heating temperatures. VO₂ films obtained exhibit excellent visible transparency and switching property at near-infrared wavelengths across the metal–semiconductor transition. Transmittance change at $\lambda = 2000$ nm of VO₂ thin film annealed at 450 °C attains as high as 41.5% and its solar modulation efficiency reaches up to 8.8%. The W-doped VO₂ film at a doping level of 1 at.% exhibits a thermochromic switch at 37 °C with a narrow hysteresis, which will greatly favor the practical application of VO₂-based smart windows.

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1. Introduction

Since Morin first discovered the metal–semiconductor transition (MST) of vanadium dioxide (VO₂), this new frontier has intrigued many researchers for several decades [1]. VO₂ undergoes a thermal-induced reversible first-order phase transition at approximately 68 °C. Above the phase transition temperature, it has a simple tetragonal lattice with the P4₂/mmm rutile space group (R phase) and exhibits metallic property. Below the phase transition temperature, it transforms to a narrow gap (0.7 eV) semiconductor with the P2₁/c space group (M phase) [2,3]. This structure transition is accompanied by significant changes in electrical conductivity, optical transmittance and reflectance in near infrared region. Those properties make VO₂ as a candidate material for a variety of potential advanced applications such as temperature sensing devices, optical switching devices, modulator, data storage medium and smart windows for building/housing energy management, etc. [4–6]. The phase transition temperature (T_T) of VO₂ is one of the most important factors for

the technological applications. Different methods, such as doping and strain control, have been proposed to modify the transition temperature closer to the ambient temperature. So far, tungsten is known as the most efficient dopant [2,6], with T_T lowering of 20 °C per 1 at.% in a typical value. Its range of distribution is summarized in the report [7] depending on various films fabrication techniques.

Various approaches have been explored to fabricate VO₂ thin films such as pulsed laser deposition, RF-sputter deposition, ion implantation and chemical vapor deposition [8–13]. These methods reveal superior quality compared to other film deposition techniques, including conformal coverage, high growth rate, high packing density and strong adhesion to underlying layer [14]. However, these methods are usually complex and expensive, which certainly increase the costs for high-throughput manufacturing. Alternatively, chemical solution deposition process such as sol–gel [15,16] and thermolysis [17,18] are preferred for their low cost and large-scale deposition. However, specific raw materials or complex treatments required during the deposition greatly limit their practical application [19]. Therefore, a facile and practical solution-based method is urgently needed. VO₂ thin films can be achieved under conventional procedures via this environment-friendly method.

In order to exploit VO₂ in a practical building envelope as fenestration application, it is important to deposit VO₂ films with excellent thermochromic properties on glass substrates, which are frequently used in industry due to their good transparency and low cost [20]. In our paper, an attempt was made to fabricate VO₂ thermochromic thin films with high visible transmittance and excellent optical

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modulation in near infrared region on amorphous glass substrates by a simple solution-based route. The effect of annealing temperatures on the morphology, structure and optical properties of VO₂ films were investigated in detail. W-doped VO₂ film was also synthesized to decrease the phase transition temperature (T_t).

2. Experimental details

2.1. Starting materials

Vanadium pentoxide (V₂O₅, AR) and oxalic acid (H₂C₂O₄, AR) were used as starting materials to prepare the vanadium precursor solution. Ethyl alcohol was used as the solvent. Ammonium tungstate hydrate ((NH₄)₅H₅[H₂(WO₄)₆]·H₂O, AR) was chosen for tungsten doping. All of these reagents were used without further purification.

2.2. Preparation of VO₂ films

The precursor solution of VO₂ films was prepared by the following procedures: 4.55 g V₂O₅ was reduced by H₂C₂O₄ under reflux in ethyl alcohol at 120 °C for 10 h, then a transparent blue solution of VOC₂O₄·xH₂O was formed. The reaction occurred as follows [21]:



The tungsten-doped VO₂ film was synthesized by the addition of (NH₄)₅H₅[H₂(WO₄)₆]·H₂O aqueous solution to the VOC₂O₄·xH₂O solution. The amount of the additive was carefully controlled for the realization of a fixed W doping level up to 1 at.%. The precursor solutions were aged for 2 h at room temperature before spin coating.

Slide glasses (25×25 mm²) were cleaned ultrasonically in acetone, ethanol and deionized water to remove organic and cation contaminations on the surface of the glass. This treatment was necessary for the formation of uniform and adherent films. Films were deposited on the glass substrate by spin coating using the precursor solutions at 600 rpm for 10 s and then 3000 rpm for 20 s. After drying at 60 °C for 10 min to remove the excess solvent, smooth films were formed. Then the as-prepared films were transformed to VO₂ films after annealing at various temperatures between 400 and 550 °C by the step of 5 °C/min in nitrogen atmosphere.

2.3. Characterization

The surface morphology of the VO₂ thin films were investigated using a field emission scanning electron microscope (FESEM, S-4800, Hitachi Japan) under the operating voltage of 2 kV. X-ray diffraction patterns were recorded by an X-ray diffractometer (XRD, PANalytical X'pert Pro MPD) in the 2θ range of 10–80° with the step of 0.0083° using Cu-Kα radiation (λ = 1.54178 Å). The accelerating voltage is 40 kV and the current is 40 mA. Ultraviolet–visible–near infrared spectrophotometer (UV–Vis–NIR, Lambda 750) was employed to characterize the optical switching properties of the films, from 250 to 2500 nm at temperature 30 °C and 100 °C, respectively. Temperature was measured with the assistance of a sensor in contact with the films which was connected with FP23 temperature controlling unit. Hysteresis loops were measured by collecting the transmittance of films at a fixed wavelength (2000 nm) at an approximate temperature interval of 2 °C.

Detail parameters of the hysteresis loops are obtained as follows. Plots of d(Tr)/d(T) & T are obtained from the hysteresis loops. Fitting each of the d(Tr)/d(T) & T curves by Origin pro 8.0 software, and a peak with well-defined maxima is yielded. The temperature ($T_{c,h}$ and $T_{c,c}$) corresponding to the maxima is defined as the phase transition temperature (T_c) of the heating and cooling branch with additional suffix *h* and *c*, respectively. The phase transition temperature (T_t) of VO₂ thermochromic films is defined to be $T_t = (T_{c,h} + T_{c,c})/2$ [22].

3. Results and discussion

3.1. XRD

Fig. 1 shows the X-ray diffraction patterns of VO₂ films annealed at different temperatures. A broad amorphous background peak, which stretches from 15 to 38°, is attributed to the amorphous glass substrates [20]. All of the film samples exhibit apparent diffraction peaks at 2θ = 27.86° and 55.65° (denoted by star (☆)), which can be indexed to (011) and (220) planes of M-phase VO₂ (JCPDS card No. 72-0514, P2₁/c, a = 0.574 nm, b = 0.452 nm, c = 0.538 nm, and β = 122.61°), respectively [3]. In addition, the (011) peak is dominant, while the (220) peak is weak, indicating that the film has a strong preferred orientation [23]. The intensity of the (011) plane significantly increases as enhancing the annealing temperature, indicating that the crystallinity of film is improved.

For films annealed at 450, 500 and 550 °C, the film samples exhibit an additional weak diffraction at ~18° (denoted by inverse triangle (▼)), which is assigned to NaV₄O₇, and this peak intensifies as raising the annealing temperature from 450 to 500 °C. Two new peaks (denoted by empty rhombus (◇)) appeared in the XRD spectrum of the film annealed at 550 °C are identified to be NaV₆O₁₅ (curve (d) in Fig. 1), but the diffraction intensity of NaV₄O₇ decreases simultaneously. The emergence of NaV₄O₇ and NaV₆O₁₅ could be particularly attributed to Na ion diffusion from the glass substrate as increasing the temperature of post-annealing.

3.2. SEM

The thin films obtained from our method are observed to be yellowish-brown, which is the characteristic color of monocline/rutile phase VO₂. With increasing the annealing temperature, the characteristic yellow-brown color bleaches out slightly, which might be attributed to the formation of NaV₄O₇ and NaV₆O₁₅. The morphology and microstructure of VO₂ films annealed at 400, 450, 500 and 550 °C are presented in Fig. 2, respectively. The distinctive morphological changes have been observed due to the different annealing temperatures. For film annealed at 400 °C (Fig. 2(a)), the homogeneous film with uniform particulates covers the glass substrate entirely and exhibits a porous structure in nano-scale. The average grain size is estimated to be 30–40 nm. As seen in Fig. 2(b), slight grain growth is displayed in film treated at 450 °C, and a trace of irregular prisms appear at the same time. The amount of prisms increases as the annealing temperature increases to

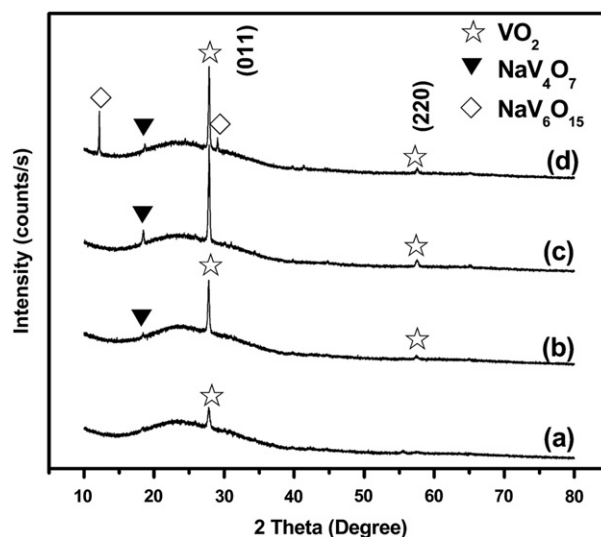


Fig. 1. XRD patterns of VO₂ films prepared at different annealing temperatures on glass substrates. (a) 400 °C; (b) 450 °C; (c) 500 °C; and (d) 550 °C.

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