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Energy-saving thermally coupled ternary extractive distillation process by combining with

mixed entrainer for separating ternary mixture containing bioethanol

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Abstract

The major intrinsic obstacle of extractive distillation is the high energy consumption. It is an

actual problem for reducing energy consumption of extractive distillation processes. Two

thermally coupled ternary extractive distillation processes were studied to separate the ternary

azeotropic mixture tetrahydrofuran/ethanol/water using a single component solvent (dimethyl

sulfoxide) and a mixed solvent (dimethyl sulfoxide and ethylene glycol) as entrainer. The

optimal conditions of all ternary extractive distillation processes were obtained based on the

minimal total annual cost. Thermodynamic efficiency and CO₂ emissions index were also

considered to evaluate the energy efficiency and environmental impact of alternative ternary

extractive distillation processes. The results show that the use of mixed entrainer can result in

reduction in both energy consumption and total annual cost for the same ternary extractive

distillation configuration. Comparisons of the conventional ternary extractive distillation process

and thermally coupled ternary extractive distillation process 1 (combining extractive distillation

column with entrainer-recovery column) with mixed entrainer show that a thermally coupled

extractive distillation sequence with a side rectifier present the best results. However, thermally

coupled ternary extractive distillation process 2 (combining extractive distillation column with

entrainer-recovery column) does not show good result due to existence of remixing effect.

Keywords: extractive distillation; thermally coupled; total annual cost; azeotrope

1. Introduction

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