## Accepted Manuscript

Crystal Morphology-Based Kinetic Study of Carbon Dioxide-Hydrogen-Tetra-nbutyl ammonium bromide Hydrates Formation in a Static System

Y.U. Yi-Song, X.U. Chun-Gang, L.I. Xiao-Sen

PII:	S0360-5442(17)31823-6
DOI:	10.1016/j.energy.2017.10.117
Reference:	EGY 11761
To appear in:	Energy
Received Date:	05 June 2017
Revised Date:	21 August 2017
Accepted Date:	25 October 2017

Please cite this article as: Y.U. Yi-Song, X.U. Chun-Gang, L.I. Xiao-Sen, Crystal Morphology-Based Kinetic Study of Carbon Dioxide-Hydrogen-Tetra-n-butyl ammonium bromide Hydrates Formation in a Static System, *Energy* (2017), doi: 10.1016/j.energy.2017.10.117

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



## Highlights

- > TBAB hydrate and TBAB/gas mixture hydrate are two independent processes.
- > Flocculent crystal formed in the gas-liquid interface is easy to form dense layer.
- > Hydrate morphology becomes constraint of gas uptake under high operating pressure.

Download English Version:

## https://daneshyari.com/en/article/8072432

Download Persian Version:

https://daneshyari.com/article/8072432

Daneshyari.com