



Review

Review of ASP EOR (alkaline surfactant polymer enhanced oil recovery) technology in the petroleum industry: Prospects and challenges



Abass A. Olajire*

Industrial and Environmental Chemistry Unit, Department of Pure and Applied Chemistry, Ladoko Akintola University of Technology, P. M. B 4000, Ogbomoso, Oyo State, Nigeria

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ABSTRACT

Owing to the inefficiency of the conventional primary and secondary recovery methods to yield above 20–40% of the OOIP (original oil in place) as incremental oil, the need for EOR (Enhanced Oil Recovery) techniques to recover a higher proportion of the OOIP has become imperative. ASP (Alkaline/Surfactant/Polymer) is one of such techniques that has proven successful due to its ability to improve displacement and sweep efficiency. Alkaline–surfactant–polymer (ASP) flooding is a combination process in which alkali, surfactant and polymer are injected at the same slug. Because of the synergy of these three components, ASP is widely practiced in both pilot and field operations with the objective of achieving optimum chemistry at large injection volumes for minimum cost. Despite its popularity as a potentially cost-effective chemical flooding method, it is not without its limitations. This paper therefore focuses on the reviews of the application of ASP flooding process in oil recovery in the petroleum industry and its limitations in maximizing oil recovery from onshore and offshore reservoirs. Also discussed are technical solutions to some of these challenges.

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1. Introduction

The high prices of crude oil and the future energy demand worldwide have necessitated the needs for EOR (Enhanced Oil Recovery) processes. Surfactant-based process, especially ASP (Alkaline/Surfactant/Polymer) method has been identified as a cost-effective CEOR (Chemical Enhanced Oil Recovery) process yielding high recovery rates of above 20% in some oilfields like Daqing oilfield in China [1,2]. While the other chemical enhanced oil recovery (CEOR) methods have suffered several drawbacks like adsorptive surfactant loss in a plain surfactant flood or long duration of a dilute alkaline flood, the ASP promises to alleviate such problems [3]. The possession of a combined chemical phase behavior of the injected surfactant and the *in-situ* generated natural surfactant is one of key advantages of the ASP flood over other CEOR methods. Thus, significant developments have made ASP flooding a viable option for field enhanced oil recovery and more attractive than other CEOR methods [4,5].

ASP flooding is a technique which is developed out on the basis of alkali flooding, surfactant flooding and polymer flooding [6–8] with gradual enhancement of oil recovery by decreasing IFT (interfacial tension), increasing capillary number, enhancing microscopic displacing efficiency, improving mobility ratio and increasing macroscopic sweep efficiency [9]. ASP flooding utilizes three types of chemicals – alkali, surfactant and polymer to recover large amounts of waterflood residual oil. This process combines the macroscopic volumetric sweep efficiency improvement from the polymer due to reduction in water–oil mobility ratio with the ability of surfactants (both added and *in situ* soaps) to enhance microscopic sweep efficiency [10–12]. This enhancement results from dramatic reduction of oil–water interfacial tension (IFT) which increases capillary number (N_c) by orders of magnitude to the required range for efficient oil recovery [13,14]. Alkali forms soaps by reacting with naturally occurring organic acid in the crude oil, which interacts synergistically with added surfactant to produce ultra-low IFT [15–17]. The ultra-low IFT is obtained by surfactant distribution between oil and water phase, and surfactant arrangement at interface of oil/water. This is controlled by pH value and ionic strength [18,19]. The alkali injected with surfactant can reduce surfactant adsorption, play the role of ionic strength and lower IFT [20–22]. Addition of polymer increases the viscosity of its

* Tel.: +234 8033824264.

E-mail address: olajireaa@yahoo.com.

Nomenclature

AAS	alkyl–aryl sulphonate	N67	Neodol 67-7PO (propoxylated) sulfate
A	alkaline	NaOH	sodium hydroxide (or caustic soda)
AOS	α -olefin sulphonate	Na ₂ CO ₃	sodium carbonate (or soda ash)
AP	alkali–polymer	NaHCO ₃	sodium bicarbonate
APG	alkyl polyglycosides	NaBO ₂	sodium metaborate
API	American Petroleum Institute	Na ₂ SO ₄	sodium sulfate
ASP	alkaline/surfactant/polymer	N_c	capillary number
AS	alkali–surfactant	N	original oil in place
B_o	formation volume factor of the oil	N_p	accumulative oil recovered
CaSO ₄	anhydrite (calcium sulfate)	N_b	bond number
CaSO ₄ ·2H ₂ O	gypsum (calcium sulfate dihydrate)	N_t	total trapping number
CaCO ₃	calcium carbonate	OOIP	original oil in place
Ca(OH) ₂	calcium hydroxide	O/W	oil-in-water
CEOR	chemical enhanced oil recovery	OPEX	operating cost
DETPMP	diethylenetriaminepenta (methylene phosphonic acid)	P	polymer
EA	ethoxylated alcohol	P_r	permeability reduction factor
EOR	enhanced oil recovery	PAA	polyacrylic acid
EO	ethoxylate	PAM	polyacrylamide
E_{ro}	oil recovery efficiency	PARCOM	Paris Commission
E_{do}	microscopic or displacement efficiency	PASP	polyaspartates
E_{vo}	macroscopic or volumetric sweep efficiency	PO	propoxylate
E_a	areal displacement efficiency	PPCA	poly phosphono carboxylic acid
E_v	vertical displacement efficiency	PS	petroleum sulphonate
FPSO	floating, production, storage and offloading	PV	pore volume
FSO	floating storage and offloading	R	resistance factor
g	acceleration due to gravity	R_r	residual resistance factor
HA _o	concentration of acid in oil	S	surfactant
HA _w	concentration of acid in water	SAC	strong acid cation
HEC	hydroxyl ethyl cellulose	SIS	select ion sequestration
HPAM	partially hydrolyzed polyacrylamide	S_{oi}	initial oil saturation
IFT	interfacial tension	S_{or}	residual oil saturation
IOS	internal olefin sulfonate	V_p	permeability variation
K_D	partition coefficient of HA between oil and water	WAG	water alternating gas injection
K_a	reaction constant	WAC	weak acid cation
k_{rw}	relative permeability of water	μ_o	oil viscosity
k_{ro}	relative permeability of oil	μ_w	water viscosity
k_a	single phase permeability (absolute permeability)	ν	brine velocity (or Darcy's velocity)
k_p	permeability of polymer solution	λ_w	water mobility
k_w	permeability of water	λ_o	oil mobility
k_{wp}	permeability of water after polymer injection	λ_p	polymer solution mobility
M	mobility ratio	λ_{wp}	mobility of water after polymer injection
Mg(OH) ₂	magnesium hydroxide	σ_{ow}	oil/water interfacial tension
Mg	magnesium	θ	contact angle between the wetting phase and the rock
		$\Delta\rho$	oil/water density difference

aqueous phase [23], so that the mobility of aqueous phase decreases. Thus, the decrease in mobility ratio greatly increases sweep efficiency.

Substantial research works are being carried out worldwide on ASP flooding process by different researchers [24–29]. The development of ASP EOR technology and advances on surfactant chemistry have brought a renewed attention for chemical floods [30], especially to boost oil production in mature and waterflooded fields. Currently, there are numerous active ASP flooding projects worldwide, with the ASP flooding implemented at the Daqing field in China considered as one of the largest ASP ongoing projects [31]. Several current ASP oilfield applications are reported in the literature [31,32]. Meanwhile, disadvantages identified during the implementation process of the ASP flooding technology, such as, severe scaling in the injection lines and strong emulsification of the

produced fluid [33–37] also limited its further application in the field. This study reviewed and assessed some of the recent advances and prospects made by the application of ASP flooding process in oil recovery in the petroleum industry and its limitations to maximizing oil recovery from onshore and offshore reservoirs. Also discussed is how these challenges could be technically addressed.

2. Enhanced oil recovery (EOR) processes

The ultimate goal of EOR processes is to increase the overall oil displacement efficiency, which is a function of microscopic and macroscopic displacement efficiency. Based on the overall materials balance of the reservoir, the overall oil recovery efficiency (E_{ro}) can be defined as:

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