



Thermodynamics of premixed combustion in a heat recirculating micro combustor



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ABSTRACT

A thermodynamic model has been developed to evaluate exergy transfer and its destruction in the process of premixed combustion in a heat recirculating micro combustor. Exergy destruction caused by process irreversibilities is characterized by entropy generation in the process. The entropy transport equation along with the solution of temperature and species concentration fields in the wake of flame sheet assumptions have been used to determine the different components of entropy generation. The role of thermal conductivity and thickness of combustor wall, and *Peclet* number on transfer and destruction rate of exergy is depicted in the process of flame stabilization via heat recirculation. The entropy generations due to gas phase heat conduction and chemical reaction are identified as the major sources of exergy destruction. The total irreversibility in pre-flame region is confined only within a small distance upstream of the flame. It has been observed that the local volumetric entropy generation is higher near the axis than that near the combustor wall. The second law efficiency is almost invariant with heat loss from the combustor, *Peclet* number, and thermal conductivity and thickness of combustor wall.

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1. Introduction

The rapid growth of micro devices brings about a strong demand for small-scale power sources. This necessitates the development of micro and mesoscale combustors for power generation in small scale devices like micro gas turbine, micro air vehicle, micro thruster, micro actuator *etc.* The scientific issues and challenges pertaining to micro and mesoscale power generating devices in practice have been addressed in recent reviews [1–3]. The geometry of a micro scale combustor corresponds to a length scale which is close to quenching distance. The ratio of heat loss to heat generation becomes significant due to high surface area to volume ratio of such combustors. As a result, flammability limits get narrowed down and sometimes it becomes very difficult to stabilize a flame within the combustor. There are different approaches for stabilizing flame in micro combustors like heat recirculation via combustor structure [4–12], maintaining high wall temperature of combustor [13–16], use of catalytic combustor [17], use of pulsed micro wave energy [18], use of stepped tube combustor with inserted wire [19]. The addition of hydrogen [20] as an additive to methane–air mixture enhances the concentration of certain

radicals which in turn increases the blow out limit. Amongst the different approaches, heat recirculation has gained considerable interest due to its natural occurrence in small scale structures. The studies on the influence of wall thermal conductivity and wall thickness on the performance of micro channel combustors with heat recirculation [21–24] reveal that an increase in wall thermal conductivity leads to a better heat recirculation thus increasing the flame speed and broadening the flammability range, while increasing the heat loss to the surrounding.

In the wake of growing concern for efficient utilization of natural energy resources, both micro and macro combustors should be thermodynamically efficient in preserving the energy quality while enhancing the burning rate and hence the flame speed. Recirculation of heat from post flame region via combustor wall to the fresh charge in enhancing the flame speed may cause higher entropy generation in the premixed gas phase region due to higher values of local temperature gradient. The increase in entropy generation may also be attributed to enhancement in species diffusion and chemical reaction. Therefore optimal utilization of energy resources should essentially be based on a compromise between the apparently contradicting aspects of entropy generation and the rates of thermo mechanical and thermo chemical transport processes. The importance of exergy based thermodynamic analyses for the performance evaluation of thermal systems has been reported in literature [25–29]. Khu et al. [30] developed a thermodynamic

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