



# Effect of CO<sub>2</sub> in the aeration gas on cultivation of the microalga *Nannochloropsis oculata*: Experimental study and mathematical modeling of CO<sub>2</sub> assimilation



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## ABSTRACT

The effects of CO<sub>2</sub>-supplementation on growth and biomass productivity of the microalga *Nannochloropsis oculata* are discussed for cultures with and without pH control. In otherwise non-limiting photoautotrophic cultures, the supply of inorganic carbon controls the algal biomass concentration and productivity. Inorganic carbon is nearly always supplied as CO<sub>2</sub>, but is taken up by the cells mainly as bicarbonate. The culture pH determines the speciation of the dissolved inorganic carbon and its availability for uptake. In air-sparged batch cultures, the pH control at 6.5, 7.0 and 8.0 by injection of CO<sub>2</sub> as needed, did not affect the biomass concentration and productivity relative to the control culture (no pH control) sparged with air. In the absence of pH control, the supplementation of air with CO<sub>2</sub> at 0.34 to 1.34% v/v levels had no effect on the biomass concentration, but the pH oscillated with the day–night cycle. Compared to the control culture, the range of pH oscillations was reduced if CO<sub>2</sub> was added to air at the specified levels. A mathematical model was developed to explain the effect of the culture pH and the CO<sub>2</sub> level in the aeration gas, on the rates of photosynthetic CO<sub>2</sub> assimilation and CO<sub>2</sub> absorption.

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## 1. Introduction

Microalgae are being extensively examined as potential sources of lipids for making biodiesel. In addition to oils, the microalgal biomass contains proteins, carbohydrates and other materials. Consequently, microalgae are potential sources of foods, feeds and feedstock for producing diverse chemicals and fuels. Microalgae transform sunlight into chemical energy (ATP and NADPH) that is used in the photosynthetic carbon reduction cycle, or Calvin–Benson cycle, to produce organic carbon. The first step of the Calvin–Benson cycle is catalyzed by the enzyme ribulose-1,5-bisphosphate carboxylase/oxygenase (RuBisCO). In both carboxylation and oxygenation reactions one of the substrates is ribulose-1,5-bisphosphate (RuP<sub>2</sub>). The other substrate is either oxygen (in oxygenase reaction) or carbon dioxide (for carboxylation reaction). Because of the low level of carbon dioxide (10–12 μM) in seawater in equilibrium with the atmosphere and the relatively low affinity of RuBisCO for carbon dioxide, algal biomass production in industrial processes requires a CO<sub>2</sub>-enriched medium to enhance the rate of photosynthesis and reduce photorespiration. Notwithstanding this, some

marine phytoplanktonic species possess mechanisms that increase the concentration of CO<sub>2</sub> in the proximity of the active site of RuBisCO [22].

A CO<sub>2</sub>-enrichment mechanism involves external carbonic anhydrases for dehydration of bicarbonate (HCO<sub>3</sub><sup>-</sup>) to maintain a constant equilibrium concentration of CO<sub>2</sub> at the site of the CO<sub>2</sub> transporter [51]. In some species, bicarbonate, the main form of the dissolved inorganic carbon (DIC), is taken up via an active transport system [12,13,15,33,46]. Inside the cells, HCO<sub>3</sub><sup>-</sup> is dehydrated by internal carbonic anhydrases located near the active site of RuBisCO to generate CO<sub>2</sub>, the required substrate.

The inorganic carbon required by a culture system is typically supplied by bubbling, or sparging, CO<sub>2</sub>-enriched air through the culture medium. The effect of a CO<sub>2</sub>-enriched air supply on the biomass production parameters of a microalgal culture depends not only on the algal strain, but also on the culture conditions (e.g. temperature, salinity, ionic strength of the medium, concentrations of the other nutrients, and the pH) being used. The culture conditions define the main form of the dissolved inorganic carbon available, its concentration, and the expression of any carbon concentrating mechanisms [22]. CO<sub>2</sub>-enriched air is commonly used to reduce the culture pH that would otherwise increase as a consequence of the uptake nutrients such as HCO<sub>3</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> [49]. In addition to the effect of pH on carbon speciation, pH can also affect the growth rate of microalgae by influencing the rates of biochemical reactions and the properties of the cell membrane [47].

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**Nomenclature**

$a_B$	total bubble surface area
$abs$	absorptance
$A_c$	total cell surface area
ASW	artificial seawater
ATP	adenosine triphosphate
$B_e$	concentration of $\text{HCO}_3^-$ in the medium
$B_i$	concentration of $\text{HCO}_3^-$ in the cell
$c_1$	parameter defined by Eq. (58)
$c_2$	parameter defined by Eq. (59)
$Ca$	carotenoid content of the biomass
$C_{\text{absorption}}$	fraction of the supplied $\text{CO}_2$ absorbed in the bioreactor
$C_{\text{Chl}}$	chlorophyll content (mass per unit biomass) in the cell
$C_{\text{Chl,HI}}$	chlorophyll content in the cell under high light
$C_{\text{Chl,LI}}$	chlorophyll content in the cell under low light
$C_e$	concentration of $\text{CO}_2$ in the medium
$C_H$	concentration of protons in the medium
$C_{Hi}$	concentration of protons in the cell
Chl <i>a</i>	chlorophyll <i>a</i> content of the biomass
$C_i$	concentration of $\text{CO}_2$ in the cell
$C_{\text{OH}}$	concentration of hydroxyl anions in the medium
$D$	dilution rate
DIC	dissolved inorganic carbon
$e^-$	electron
$f$	parameter correcting for spectral quality of the light
$F$	molar flow rate of the aeration gas
$F_C^{\text{in}}$	inlet molar flow rate of $\text{CO}_2$
$F_N^{\text{in}}$	inlet molar flow rate of $\text{N}_2$
$F_O^{\text{in}}$	inlet molar flow rate of $\text{O}_2$
$F_C^{\text{out}}$	outlet molar flow rate of $\text{CO}_2$
$F_O^{\text{out}}$	outlet molar flow rate of $\text{O}_2$
$F_C^{\%}$	molar flow rate of $\text{CO}_2$ in the gas bubbles reaching the headspace
$F_O^{\%}$	molar flow rate of $\text{O}_2$ in the gas bubbles reaching the headspace
$H_C$	Henry's law constant for $\text{CO}_2$
HI	high irradiance
$H_O$	Henry's law constant for $\text{O}_2$
$I$	incident irradiance
$I_2$	fraction of the incident irradiance ( $I$ ) absorbed by photosystem II
$j$	potential electron transport rate per unit mass of chlorophyll
$J$	potential electron transport rate
$J_{\text{max}}$	maximum rate of carboxylation allowed by light reactions at given partial pressures of $\text{CO}_2$ and $\text{O}_2$
$k_1$	forward rate constant of $\text{CO}_2$ hydration
$k_{-1}$	reverse rate constant of $\text{CO}_2$ hydration
$k_3$	forward rate constant of $\text{CO}_2$ hydration in acidic conditions
$k_{-3}$	reverse rate constant of $\text{CO}_2$ hydration in acidic conditions
$k_3^*$	$k_3$ value estimated at the intracellular pH
$k_4$	forward rate constant of $\text{CO}_2$ hydration in alkaline conditions
$k_{-4}$	reverse rate constant of $\text{CO}_2$ hydration in alkaline conditions
$k_4^*$	$k_4$ value estimated at the intracellular pH
$k_{\text{LB}}$	mass transfer coefficient between gas bubbles and the liquid
$k_{\text{LH}}$	mass transfer coefficient between gas in the headspace and the medium
$K_2$	dissociation constant of carbonic acid to bicarbonate

$K_3$	equilibrium constant of $\text{CO}_2$ hydration in acidic conditions
$K_4$	equilibrium constant of $\text{CO}_2$ hydration in alkaline conditions
$K_5$	dissociation constant of the bicarbonate to carbonate ion product of water
$K_6$	ion product of water
$K_b$	Michaelis constant of the intracellular carbonic anhydrase for $\text{HCO}_3^-$
$K_c$	Michaelis constant of the intracellular carbonic anhydrase for $\text{CO}_2$
$K_C$	Michaelis constant of RuBisCO for $\text{CO}_2$
$K_O$	inhibition constant of RuBisCO for $\text{O}_2$
$K_f$	Michaelis constant of $\text{HCO}_3^-$ transport out of the cell
$K_i$	Michaelis constant of $\text{HCO}_3^-$ transport into the cell
$L$	lipid content of the biomass
$LI$	low irradiance
$N$	cell concentration
NADPH	nicotinamide adenine dinucleotide phosphate (reduced form)
$O_e$	concentration of $\text{O}_2$ in the medium
$O_i$	concentration of $\text{O}_2$ in the cell
$p_{\text{CO}_2}^{\text{out}}$	partial pressure of $\text{CO}_2$ in the outlet gas
$p_{\text{O}_2}^{\text{out}}$	partial pressure of $\text{O}_2$ in the outlet gas
$p_C^{\%}$	partial pressure of $\text{CO}_2$ in the bubbles reaching the headspace
$p_O^{\%}$	partial pressure of $\text{O}_2$ in the bubbles reaching the headspace
$P_C$	cell permeability to $\text{CO}_2$
$P_O$	cell permeability to $\text{O}_2$
$P_T$	total pressure
$Q$	volumetric flow rate of gas
$Q_f$	volume flow rate of feed
$QY$	quantum yield
$r_1$	rate of $\text{HCO}_3^-$ dehydration by intracellular carbonic anhydrase
$r_2$	rate of $\text{CO}_2$ hydration by carbonic anhydrase
$r_3$	rate of $\text{HCO}_3^-$ dehydration in the cell
$r_4$	rate of $\text{CO}_2$ hydration in the cell
$r_6$	rate of $\text{HCO}_3^-$ transport from the medium to the cell
$r_7$	rate of $\text{HCO}_3^-$ transport from the cell to the medium
$r_8$	rate of carboxylation
$r_9$	rate of $\text{CO}_2$ diffusion due to membrane permeability
$r_{10}$	rate of $\text{O}_2$ diffusion due to membrane permeability
$r_{11}$	rate of oxygenation by RuBisCO
$r_{12}$	rate of $\text{CO}_2$ production by dark respiration
$r_a$	rate of $\text{CO}_2$ assimilation per mole of Chl in the biomass
$r_B$	mean bubble radius
$r_C$	mean cell radius
$r_{\text{dark}}$	dark respiration rate per unit leaf surface area
$R$	gas constant
$R_B$	rate of $\text{HCO}_3^-$ production in the liquid phase
$R_C$	rate of $\text{CO}_2$ production in the liquid phase
$R_{\text{dark}}$	dark respiration rate
$R_{\text{dark, HI}}$	dark respiration rate under high light
$R_{\text{dark, LI}}$	dark respiration rate under low light
RuBisCO	ribulose biphosphate carboxylase-oxygenase
RuP <sub>2</sub>	ribulose-1,5-bisphosphate
$S_H$	area of the interface between the liquid and the headspace in the bioreactor
$S$	salinity
$S_{\text{CO}}$	specificity of RuBisCO
$t$	time
$T$	absolute temperature
TS	total sugar content in the biomass

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