



# Thermal regeneration of amines in vertical, inclined and oscillating CO<sub>2</sub> packed-bed strippers for offshore floating applications

Ion Iliuta\*, Faïçal Larachi

Department of Chemical Engineering, Laval University, 1065, Avenue de la Médecine, Québec, Québec, G1 V 0A6, Canada

## ARTICLE INFO

### Keywords:

FPSO  
Onboard marine CO<sub>2</sub> stripper  
Inclined and oscillating packed-bed columns  
CO<sub>2</sub>-MEA desorption process  
3-D model  
Reactor performance

## ABSTRACT

CO<sub>2</sub>-monoethanolamine desorption process performance was studied for standard vertical, static inclined, and symmetric/asymmetric oscillating packed-bed columns via a dynamic three-dimensional model which links the macroscopic volume-averaged continuity, momentum, energy and species balance equations in the liquid and gas phases with the purpose to describe the stripping packed-bed columns aboard floating production, storage and offloading platforms. CO<sub>2</sub> thermal desorption process is negatively impacted in static inclined and asymmetric oscillating packed-bed columns because of secondary liquid flow, generated by the buoyancy force in the radial and tangential directions, and its consequence on two-phase flow hydrodynamics and temperature fields. The continual variation of the extent of reverse secondary flow in symmetric oscillating packed-bed columns induces a symmetrical oscillatory two-phase flow follows by time-dependent waves for thermal fields and CO<sub>2</sub> thermal desorption performance in the vicinity of the standard vertical stripper steady-state solution. Operation with extra heat allows to obtain higher CO<sub>2</sub> desorption rates and avoids the deterioration of CO<sub>2</sub> desorption performance in static inclined and asymmetric oscillating packed-bed columns. The extra heat is considerable when the inclined and asymmetric oscillating packed-bed columns are operated at elevated reboiler heat duty in the vertical state.

## 1. Introduction

Chemical and petrochemical, steel, cement and aluminum production industrial processes release annually a large quantity of CO<sub>2</sub> into the atmosphere. CO<sub>2</sub> removal by chemical absorption/desorption processes with alkanolamine-based aqueous solutions (monoethanolamine – MEA, diethanolamine, diisopropanolamine, N-methyldiethanolamine, 2-amino-2-methyl-1-propanol – AMP) and blended alkanolamine-based aqueous solutions, in which the amines are regenerated to be reused, are currently massively used for CO<sub>2</sub> capture because of the high-level of CO<sub>2</sub> absorption efficiency (Kohl and Nielsen, 1997; Bougie and Iliuta, 2010; Penders-van Elk et al., 2012; Luis et al., 2012; Rambo et al., 2014). CO<sub>2</sub>-amine absorption/desorption processes are usually carried out in countercurrent packed-bed columns whose performance is largely governed by the mass-transfer efficiency of the random/structured packings used to provide the gas-liquid contacting area (Aroonwilas et al., 2001).

CO<sub>2</sub> thermal desorption in stripping packed-bed columns is central to the design of CO<sub>2</sub>-amine absorption/desorption processes. It is highly-energy intensive and responsible for the main operational cost of the process (Bougie and Iliuta, 2010). Paradoxically, there is relatively

a reduced amount of information related to CO<sub>2</sub> desorption process in the open literature compared to the overwhelming proportion of studies on CO<sub>2</sub> absorption. Recently, the researchers attempted to address these gaps in published different approaches to tackle the problem of high energy consumption in CO<sub>2</sub> thermal desorption process by developing alternatives with amines for low regeneration cost (Bougie and Iliuta, 2010; Warudkar et al., 2013). Cullinane and Rochelle (2004) have studied the carbon dioxide absorption with aqueous potassium carbonate promoted by piperazine, a cyclic diamine with very fast reaction kinetics, and found that the activated potassium carbonate has a lower reaction heat than MEA. Bougie and Iliuta (2010) showed that solutions of hindered amines (2-amino-2-hydroxymethyl-1,3-propanediol-AHPD, 2-amino-2-methyl-1,3-propanediol-AMPD, and 2-amino-2-ethyl-1,3-propanediol-AEPD), and in particular AHPD, are more easy to regenerate as they do not form (or form very few) stable carbamates. Also, Bougie and Iliuta (2010) found that addition of a small amount of piperazine into AHPD aqueous solution allows to obtain the most appropriate solvent for CO<sub>2</sub> capture. Freeman et al. (2010) studied the use of concentrated aqueous piperazine as a fast reacting absorbent for CO<sub>2</sub> capture process. Also, CO<sub>2</sub> capture with ionic liquids, salts that are liquids at room temperature, have been suggested by Hasib-Ur-Rahman

\* Corresponding author.

E-mail address: [ion.iliuta@gch.ulaval.ca](mailto:ion.iliuta@gch.ulaval.ca) (I. Iliuta).

**Nomenclature***Notation*

$a$	Gas-liquid interfacial area, $\text{m}^2/\text{m}^3$
$a_s$	Packing specific surface area (surface packing area/column volume), $\text{m}^2/\text{m}^3$
$c_{p,\alpha}$	Specific heat capacity of $\alpha$ -phase ( $\alpha = g, l$ ), $\text{J/kgK}$
$C_j$	Concentration of species $j$ , $\text{kmol}/\text{m}^3$
$d_p$	Effective particle diameter, $\text{m}$
$D$	Column diameter, $\text{m}$
$D_{j\ell}$	Molecular diffusivity coefficient of species $j$ in liquid phase, $\text{m}^2/\text{s}$
$D_{\ell(g)}$	Liquid and gas dispersion coefficients, $\text{m}^2/\text{s}$
$E_1$	$E_2$ Ergun constants, –
$f_e$	Wetting efficiency, –
$F_{g\ell}$	Gas-liquid drag force, $\text{N}/\text{m}^3$
$F_{gs}$	Gas-solid drag force, $\text{N}/\text{m}^3$
$F_{ls}$	Liquid-solid drag force, $\text{N}/\text{m}^3$
$g$	Gravitational acceleration, $\text{m}/\text{s}^2$
$H$	Reactor height, $\text{m}$
$H_j^\alpha$	Molar enthalpy of species $j$ in $\alpha$ -phase, $\text{kJ}/\text{kmol}$
$N_j$	Interfacial molar flux, $\text{kmol}/\text{m}^2\text{s}$
$P$	Reactor pressure, $\text{Pa}$
$P_j$	Partial pressure of species $j$ , $\text{Pa}$
$P_c$	Capillary pressure, $\text{Pa}$
$r$	Radial coordinate, $\text{m}$
$R$	Ideal-gas constant or reactor radius
$R_j$	Reaction rate of the component $j$ , $\text{kmol}/\text{m}^3\text{s}$
$t$	Time, $\text{s}$
$T_a$	Period of angular motion
$T_\alpha$	$\alpha$ -Phase temperature, $\text{K}$
$T_{int}$	Temperature at the gas-liquid interface
$u_\alpha$	Interstitial velocity of $\alpha$ -fluid, $\text{m}/\text{s}$
$v_{s\alpha}$	$\alpha$ -Phase superficial velocity, $\text{m}/\text{s}$
$x$	Liquid film coordinate, $\text{m}$
$z$	Axial coordinate, $\text{m}$

*Greek letters*

$\alpha$	Angle of packed-bed column inclination with respect to the horizontal plane
$\alpha_{g\ell}$	Heat transfer coefficient at the gas-liquid interface, $\text{J}/\text{m}^2\text{sK}$
$\alpha_{max}$	Amplitude of the angular motion
$\varepsilon$	Packed bed porosity, –
$\varepsilon_\alpha$	$\alpha$ -Phase holdup, –
$\varepsilon_b$	Porosity in the bulk region of the packed bed, –
$\delta_\ell$	Liquid film thickness, $\text{m}$
$\Delta H_j^{vap}$	Vaporization enthalpy of species $j$ , $\text{kJ}/\text{kmol}$
$\Delta H_r$	Reaction enthalpy, $\text{kJ}/\text{Kmol}$
$\lambda_{\alpha,r}^{ef}$	Radial effective thermal conductivity of $\alpha$ -phase, $\text{J}/\text{msK}$
$\mu_\alpha$	$\alpha$ -Phase dynamic viscosity, $\text{kg}/\text{ms}$
$\mu_\alpha^{ef}$	$\alpha$ -Phase effective viscosity (combination of bulk and shear terms), $\text{kg}/\text{m s}$
$\nu_B$	Stoichiometric coefficient of amine in reaction
$\rho_\alpha$	$\alpha$ -Phase density, $\text{kg}/\text{m}^3$
$\rho_{pb}$	Packed bed density, $\text{kg}/\text{m}^3$
$\sigma_\ell$	Surface tension, $\text{N}/\text{m}$
$\theta$	Azimuthal coordinate, $\text{m}$

*Subscripts/superscripts*

$g$	Gas phase
$i$	Gas-liquid interface
$in$	Reactor inlet
$l$	Liquid phase
$\ell f$	Liquid film
$r$	Radial direction
$s$	Solid phase
$trans$	Transfer
$vap$	Vaporization
$z$	Axial direction
$w$	Reactor wall

et al. (2012), Hasib-Ur-Rahman and Larachi (2012), Zhang et al. (2012), Hasib-Ur-Rahman and Larachi (2013a, 2013b), Acidi et al. (2014) and Iliuta et al. (2014).

Nowadays, the offshore gas and oil industry located in zones of deep ocean waters far of continental coasts is concerned by the integration of  $\text{CO}_2$  removal in countercurrent random/structured packed-bed columns aboard floating production, storage and offloading (FPSO) nonstationary units. However, there are major challenges in advancing, the most prominent one being the impact of sea conditions on the performance of floating countercurrent packed-bed columns for  $\text{CO}_2$  absorption/desorption processes. The offshore environment confronted with sea perturbations can initiate various ship motions which generate pressure drop and liquid holdup fluctuations, liquid maldistribution, interphase mass transfer worsening followed by significant  $\text{CO}_2$  removal performance deviations in packed-bed columns (Kim et al., 2007; Gu and Ju, 2008; Iliuta and Larachi, 2016a). Therefore, the design of  $\text{CO}_2$  absorption and desorption countercurrent packed-bed columns exposed to variable wind, marine currents and waves must take these effects into consideration (Pham et al., 2015; Iliuta and Larachi, 2016a, 2017) and should include valuable corrective actions (supplementary capacity, liquid redistribution, etc.) to neutralize or to avoid any performance deviation with regard to the land-based  $\text{CO}_2$  removal units (Assima et al., 2015).

Even if  $\text{CO}_2$  absorption performance in vertical countercurrent packed-bed columns has been extensively examined, only a small

number of studies have focused on the absorption performance of packed-bed columns in inclined states with the aim to understand the behavior of  $\text{CO}_2$  absorption process aboard FPSO nonstationary platforms. Pham et al. (2015) investigated the impact of ship inclination and motion, at small inclination angles ( $0$ – $4.5^\circ$ ), on liquid holdup, liquid maldistribution, two-phase pressure drop and  $\text{CO}_2$  removal efficiency ( $\text{CO}_2$ -MEA absorption process) in an isothermal countercurrent structured packed-bed column via a gas-liquid Eulerian computational fluid dynamics model. Iliuta and Larachi (2017) analyzed 3-D flow fields and  $\text{CO}_2$ -MEA absorption performance in standard vertical, static inclined (inclination angles from  $0$  to  $15^\circ$ ) and asymmetric/symmetric oscillating countercurrent random packed-bed columns via a dynamic non-isothermal three-dimensional model which couples the macroscopic volume-averaged momentum, mass, energy and species conservation equations in the liquid and gas phases and the differential equations describing the diffusion and chemical reaction in the liquid film surrounding the gas-liquid interface.  $\text{CO}_2$ -MEA absorption process in inclined and asymmetrically oscillating packed-bed columns faces evident deviations as regards the standard vertical packed-bed column while symmetrically oscillating packed-bed columns gives oscillatory absorption performance waves, with amplitude and propagation frequency altered by the packed-bed column oscillations, in the vicinity of the steady-state solution of the standard vertical packed-bed column. The impact of static inclination on the  $\text{CO}_2$  removal efficiency is more important than the effect of the rolling motion, when the angle of

Download English Version:

<https://daneshyari.com/en/article/8089375>

Download Persian Version:

<https://daneshyari.com/article/8089375>

[Daneshyari.com](https://daneshyari.com)