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Review on the recent improvements in sonochemical and combined sonochemical oxidation processes – A powerful tool for destruction of environmental contaminants

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ABSTRACT

The last two decades research demonstrate that the presence of organic contaminants in drinking water has increased every year and the increasing demand for the technology to reclaim the water from the effluents released by the industries has persuaded us to write the review in this topic. The combination of advanced oxidation processes (AOPs) namely photocatalysis, Fenton and ozonolysis combined with sonolysis harvests the degradation of wide spectrum of organic contaminants. The various parameters including the geometry of the sonochemical reactors influencing the sonochemical reactions have been discussed. The combination of sonolysis with photocatalysis, Fenton and ozonolysis enhances the production of non-selective radicals to improve the degradation process thus avoiding the secondary pollution. The economical and technical feasibilities of the AOPs are very essential for the initialization of the combined AOPs for the degradation of environmental contaminants. The review envisions to improve our understanding and outline directions for future research.

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1. Introduction

In the forthcoming years, worldwide crisis may arise for energy, food and water; however, the three are inter-connected and water is essential for all social and economic developments among the nations [1]. During the past two decades, considerable increase in the concentration of organic contaminants in surface water has been observed [2,3]. On the other hand diseases spread through

Table 1
Some reactions involved in advanced oxidation processes [19,20].

Advanced oxidation process	Reaction steps
Sonolysis	$H_2O \rightarrow \cdot OH + OH^-$ $H_2O \rightarrow (1/2)H_2 + (1/2)H_2O_2$
Semiconductor photocatalysis (TiO₂-semiconductor)	$TiO_2 + h\nu \rightarrow TiO_2^- + \cdot OH$ (or TiO_2^+) $TiO_2^- + O_2 + H^+ \rightarrow TiO_2 + \cdot OH + H_2O$ $TiO_2^- + 2H^+ \rightarrow TiO_2 + \cdot H_2$ $H^+ + H_2O \rightarrow \cdot OH + H^+$
Fenton reactions	$Fe^{2+} + H_2O_2 \rightarrow Fe^{3+} + \cdot OH + OH^-$
Ozonolysis	$O_3 + H_2O \rightarrow 2HO_2\cdot$ $O_3 + 2HO_2\cdot \rightarrow \cdot OH + 2O_2$
Sonophotocatalysis (with H₂O₂)	
(i) Water sonolysis	$H_2O \rightarrow \cdot OH + OH^-$ $H_2O \rightarrow (1/2)H_2 + (1/2)H_2O_2$
(ii) Reaction of H₂O₂ with H atoms (formed from water sonolysis)	$H_2O_2 + H\cdot \rightarrow H_2O + \cdot OH$
(iii) Photolytic dissociation of H₂O₂	$H_2O_2 + h\nu \rightarrow 2\cdot OH$
(iv) Reaction of H₂O₂ with superoxides (formed in the presence of TiO₂ and under UV irradiation)	$H_2O_2 + O_2^{\cdot -} \rightarrow \cdot OH + OH^- + O_2$
(v) Reaction of H₂O₂ with electrons (conduction band electrons are generated from semiconductor photocatalyst under UV irradiation UV + hydrogen peroxide + ozone)	$H_2O_2 + e^- \rightarrow \cdot OH + OH^-$ $O_3 + OH^- \rightarrow \cdot OH$ $3O_3 + h\nu \rightarrow 2\cdot OH$ $H_2O_2 + h\nu \rightarrow 2\cdot OH$ $H_2O_2 + O_3 \rightarrow 2\cdot OH$
Ozone + sonolysis	$H_2O \rightarrow \cdot OH + OH^-$ $H_2O \rightarrow (1/2)H_2 + (1/2)H_2O_2$ H_2O_2 $O_3 \rightarrow O_2(g) + O(^3P)(g)$ $O(^3P)(g) + H_2O(g) \rightarrow 2\cdot OH$

water which is not suitable for usage in day to day life of human beings. As a result of the usage of polluted water, water borne diseases spread throughout the world and in the worst case it causes the mortality of children [4,5]. Naturally available drinking water is rare and limited nonetheless it is needed for all domestic and commercial applications which clearly illustrates how even a small shortage of water could become a threat to mankind. In day to day life, the environmental pollution is appears to be very much responsible for the negative consequences in the lives of all beings on the earth. The ingestion of chemical substances into the water bodies by the consumers destroys the quality of water continuously, which is called aquatic pollution. The contaminating (foreign) substances are generally classified into organic and non-organic contaminants, while Adewuyi classifies the contaminating pollutants into various categories [6]. The discharge of the un-mineralized contaminants, for example effluents released from the textile industries, pulp and paper industries, pharmaceutical industries etc., into the water resources causes environmental pollution. Similarly, non-organic substances, for example, excessive addition of agricultural chemicals causes water as well as land pollution [7]. The release of these un-neutralized pollutants into the environment creates detrimental effects and adulteration of these chemicals into the water bodies not only creates health hazards to the human beings but also affects the aquatic organisms. For example the presence of 4-nonylphenol (endocrine disrupter) in sewage disposal plants creates the feminizing effects in fishes which is a serious problem in terms of ecological conservation [8–10].

To avoid the hazardous effects created by the presence of these chemicals, many methodologies such as adsorption [11], biodegradation [12,13], photocatalysis [14] and ozonation [15] are proposed one by one from the early 1970s for the initiation of degradation of environmental pollutants [16–18]. As a result of these innovations, some awareness has been created among the public by the researchers about environmental pollutants. Many research groups have started working towards the complete removal of these pollutants from the environment by proposing, developing and adopting the new technologies for the degradation

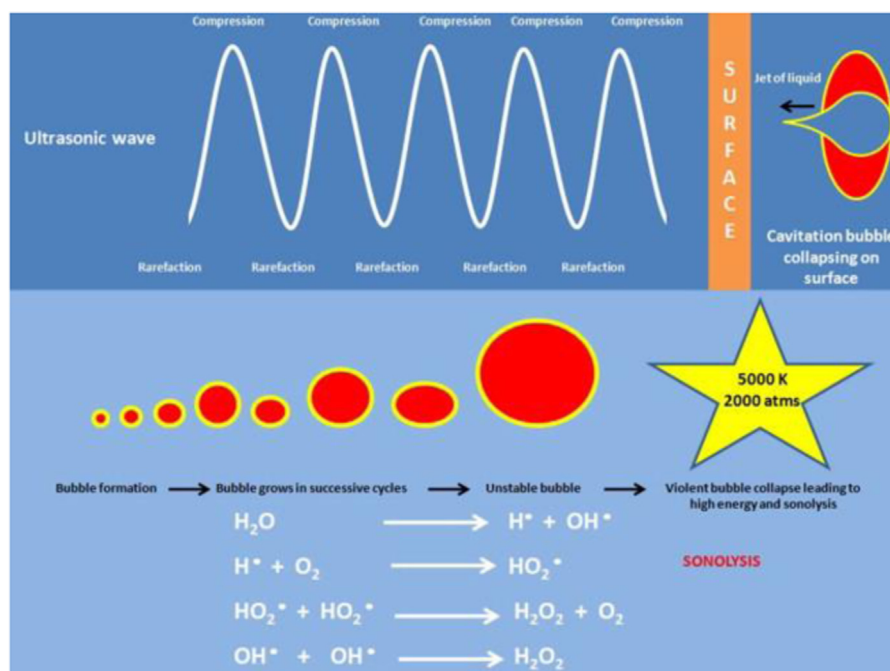


Fig. 1. Generation of an acoustic bubble [39].

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