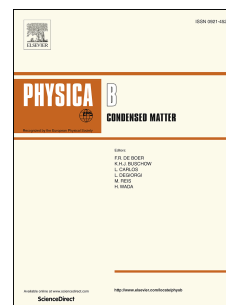


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Structural, electronic, elastic, and thermodynamic properties of CaSi, Ca₂Si, and CaSi₂ phases from first-principles calculations

X.D. Li^a, K. Li^{a,1}, C.H. Wei^a, W.D. Han^a, N.G. Zhou^b

(^aSchool of Mechanical and Electrical Engineering, Nanchang University, Nanchang 330031, China;

^bSchool of Materials science and Engineering, Nanchang University, Nanchang 330031, China)

Abstract: The structural, electronic, elastic, and thermodynamic properties of CaSi, Ca₂Si, and CaSi₂ are systematically investigated by using first-principles calculations method based on density functional theory (DFT). The calculated formation enthalpies and cohesive energies show that CaSi₂ possesses the greatest structural stability and CaSi has the strongest alloying ability. The structural stability of the three phases is compared according to electronic structures. Further analysis on electronic structures indicates that the bonding of these phases exhibits the combinations of metallic, covalent, and ionic bonds. The elastic constants are calculated, and the bulk modulus, shear modulus, Young's modulus, Poisson's ratio, and anisotropy factor of polycrystalline materials are deduced. Additionally, the thermodynamic properties were theoretically predicted and discussed.

Keywords: First-principles calculations; Electronic structures; Elastic constants; Thermodynamic properties

1. Introduction

The Ca-Si intermetallic compounds including CaSi, Ca₂Si, and CaSi₂ are composed of inexpensive elements and are widely used in energy, aerospace, electronics, and other fields [1–4]. These compounds have several unique properties, CaSi can be used as a hydrogen storage material because H atoms can be easily absorb onto its surface, creating CaSiH₃ [5]. Ca₂Si is an environment-friendly semiconductor, with a 1.9-eV band gap at room temperature, which can be used for photoelectric and thermoelectric conversion [6]. The 6R structure of CaSi₂ was discovered by Kautsky in the 1950s [7]. CaSi₂ particles, which can be formed in Si-containing magnesium alloys by adding Ca into the melt, can act as heterogeneous nuclei of Mg₂Si phases, modifying their morphology [8]. In addition, CaSi₂ particles have good thermal stability and thus enhance the heat-resistant capability of magnesium alloys.

The thermodynamic data of Ca-Si intermetallic compounds are still incomplete [9,10], and making it important to systematically investigate their various properties. In this work, we carried out first-principles calculations to investigate the electronic, elastic, and thermodynamic properties of CaSi, Ca₂Si, and CaSi₂ phases, in order to evaluate their structural and thermal stability.

2. Computational methods

Calculations of the total energy and electronic, elastic, and thermodynamic properties were performed with the Cambridge sequential total energy package (CASTEP), a first-principles plane-wave pseudopotential based on density functional theory (DFT) [11]. The

¹ Corresponding author. Tel.: +86 18970029945; fax: +86 79183969622. E-mail address: like.1@126.com (K.Li).

Present address: School of Mechanical and Electrical Engineering, Nanchang University, 999 Xuefu Road, Honggutan District, Nanchang, 330031, China.

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