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The adsorption and dissociation of oxygen on Ag (111) supported

χ₃ Borophene

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Abstract: The superstructure of χ_3 borophene on Ag (111) has recently been

synthesized in experiment. In this work, we investigate its structural, electronic

properties and the oxidation mechanism through first-principles calculations. We find

the superstructure of χ_3 borophene on Ag (111) maintain the planar characteristics,

like its free-standing form, owing to the weakly interaction between adsorbate and

substrate. Moreover, oxygen molecule can be spontaneously adsorbed on its

superstructure in a manner of chemical adsorption. Importantly, the energy barrier of

~0.35 eV for oxygen dissociation indicates its relative stability in ambient conditions

compared with silicene. Furthermore, mobility of the active the

O₂-dissociation-induced O atom is poor at room temperature, implying the difficult

migration of O atom on borophene surface. On the other hand, due to the strong B-O

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