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# Large magnetocaloric effect in geometrically frustrated polycrystalline ErMnO<sub>3</sub> compound at cryogenic temperature



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#### ABSTRACT

In this manuscript we report significantly large magnetocaloric effect in the cryogenic temperature region for the chemically stable polycrystalline  $ErMnO_3$  compound. For this compound, the antiferromagnetic transition temperature of the  $Mn^{3+}$  ions is nearly  $T\sim75\,$  K. However, the large non saturating magnetization and magnetocaloric effect are found at a temperature lower than 20 K. Such non saturating nature of magnetization and magnetic entropy changes for the application of external magnetic field is addressed by considering the gradual alignments of dominant paramagnetic moments caused by  $Er^{3+}$  ions.

#### 1. Introduction

During previous few decades, the search for energy materials gets an utmost importance from the technological as well as fundamental point of view [1-20]. Among the several properties of the materials, multiferroic, magnetoresistive properties, magnetocaloric effect (MCE) etc., were extensively studied [1,2,12,21-24]. MCE of a materials is generally described by the isothermal magnetic entropy changes or adiabatic temperature changes due to the application of external magnetic fields. Since magnetocaloric effect is directly related with the suppression of the magnetic randomness in the presence of external magnetic fields, initially the materials having high magnetic moment namely Gd, and Gd-based intermetallic compounds were considered to be giant magnetocaloric materials [1]. However, due to the lower production cost, high chemical stability, significantly larger resistivity, the oxide based perovskite manganites also get a noticeable attention during previous two decades [12]. Additionally, another beneficial aspect is the tunability of operating region of manganites. From the application point of view, magnetocaloric materials broadly categorize into two different operating regions: cryogenic temperature and near room temperature region. There are a numerous studies of MCE at low temperature on the intermetallic compounds and that on oxide based materials, near room temperature [1,12]. However, a few studies related to the large magnetocaloric effect at the cryogenic temperature range for stable oxide based compounds were also reported earlier [12,16].

According to the earlier reports, maximum value of the magnetic entropy changes generally occur near the vicinity of the disorder to order (paramagnetic to ferromagnetic/antiferromagnetic) transition [1,12]. However, in some special cases the large magnetic entropy changes also appears in the relatively lower temperature region [16].

To concentrate on perovskite manganites, the general formula is  ${\rm RMnO_3}$  where R is trivalent element. In most of the cases cubical crystallographic symmetry and antiferromagnetic (A-type) insulating ground state was found. However slight distortion of this crystallographic symmetry gives rise to several interesting properties due to the strong correlation between the spin, lattice and charge degrees of freedom. The large magnetocaloric effect associated with the orientation of the trivalent rare earth ions having large magnetic moment was reported earlier [16]. Regarding this context it is important to mention that in case of the geometrically frustrated magnetic materials large value of the relative cooling power (RCP) which is the product of the maximum value of the entropy change and its temperature span at half maxima can be observed due to higher magnetic randomness. The large value of RCP due to similar fact is also reported recently in intermetallic compounds [25].

In the present study we report the magnetic and magnetocaloric properties of geometrically frustrated polycrystalline  $ErMnO_3$  compound. According to the previous studies this compound form in hexag-

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onal crystal structure with space group  $P6_3CM^{23}$ . The  $ErMnO_3$  compound also known to show multiferroic properties as mentioned in earlier works [21–23]. Our study indicates that the magnetization of this compound increases with decreasing temperature without any signature of saturation down to 5 K. The compound also exhibits significantly large non saturating magnetocaloric effect at low temperature region. Because of such magnetocaloric properties along with the other added advantages this compound can be important from the technological perspectives at the cryogenic temperature region.

#### 2. Sample preparation, characterizations and measurements

The polycrystalline  $ErMnO_3$  compound was prepared by the sol-gel route. Starting materials were preheated  $Er_2O_3$  and  $MnO_2$ .  $Er_2O_3$  was converted to its nitrate forms by adding nitric acid and dissolved into millipore water. Since  $MnO_2$  is directly not converted into its nitrates form, hence first its converted to its oxalate form by using appropriate amount of oxalic acid and millipore water and finally dissolved into the millipore water after adding the nitric acid.

After dissolving the constituent elements, next stage of this process is gel formation. For this part, individual clear solutions were mixed-up in a beaker and stirrer some time for proper homogeneity of the solution and appropriate citric acid was mixed. The beaker, containing the homogeneous precursor solution was kept in a water bath at  $\sim\!80\,^{\circ}\mathrm{C}$  for slow evaporation of the extra water. After evaporation of water the gel was formed and it was decomposed at slightly higher temperature ( $\sim\!200\,^{\circ}\mathrm{C}$ ) to get black porous powder. The powder was pelletized and finally annealed at 1300  $^{\circ}\mathrm{C}$  for 36 h and crystalline compound was formed

Room temperature x-ray diffraction study have been performed in a Rigaku-TTRAX-III diffractometer. To study the magnetic and magnetocaloric effect, a Super conducting Quantum Interference Device (SQUID) (Quantum Design) magnetometer was employed.

#### 3. Results and discussion

X-ray diffraction study indicates the chemically single phase nature of the compound. The Rietveld refinement (profile fitting) indicates that the space group symmetry of that compound is  $P6_3$ CM (hexagonal structure) which is similar as reported earlier [23].

Magnetization as a function of temperature in the presence of external magnetic field H = 100 Oe was measured and shown in Fig. 1. Magnetization of the compound was measured in two different protocols which are denoted by ZFC (Zero Field Cooled) and FC (Field Cooled). In case of ZFC method, sample was first cooled down from the room temperature in the absence of any external magnetic field and after stabilizing the temperature (T = 5 K), magnetic field of 100 Oe was applied and magnetization (as a function of temperature) in the warming cycle was measured from T = 5–300 K. Whereas, for FC protocol, sample was cooled down in the presence of 100 Oe external magnetic field and temperature dependence magnetization was measured during warming cycle from T = 5–300 K.

The temperature dependence of the magnetization (shown in Fig. 1) exhibits similar nature with decrease of temperature except slightly greater value of magnetization in FC protocol (at low temperature region). The greater value of FC magnetization compared to ZFC inhibits the anisotropic behavior of the sample. The origin of the anisotropic behavior is not the subject matter of this manuscript. It can be only mention that the value of anisotropy is very small. The inverse susceptibility  $(1/\chi)$  of this compound calculated from the FC data and given in Fig. 1, indicates almost liner in nature.

For further investigation about the nature of the magnetic ground state of the compound, magnetic field dependence of the magnetization was measured at  $T=5\,$  K and given in Fig. 2. Before starting the measurement, initially the sample was cooled down to  $T=5\,$  K from room temperature ( $T=300\,$  K) in the absence of any magnetic field. The

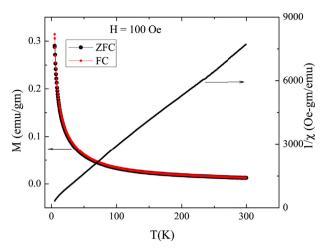


Fig. 1. Magnetization in ZFC and FC protocols (both coincide each other) and inverse susceptibility (calculated from FC data) as a function of temperatures for  $H=100\,$  Oe external magnetic field.

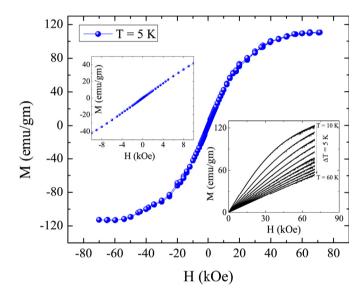


Fig. 2. Magnetization as a function of external magnetic field at  $T=5\,$  K. Upper inset is the enlarge view at the lower magnetic field region and lower inset is the some representatives magnetic isotherms at different temperature.

experimental data of magnetization as a function of magnetic field indicates that the magnetization value almost saturates at 70 kOe magnetic field value. Such nature again indicates negligible magnetic anisotropy in the system. Another important fact from the application point of view is even at low temperature region, there is negligible hysteresis loop in magnetization data (enlarged view is given in upper inset of Fig. 2). Some selected magnetic isotherms at different temperatures are given in lower inset of Fig. 2.

As mentioned in the introduction part, the multiferroicity in this compound was reported earlier [21]. In this present study, magnetic properties indicates that the saturation of the magnetization appears for  $H=70\,$  kOe at  $T=5\,$  K. Additionally no hysteretic nature (another beneficial aspect of refrigerant materials) even at the low temperature region motivated us to study the magnetocaloric responses of this material. The magnetocaloric properties (isothermal magnetic entropy change) of this materials were estimated from the magnetic isotherms data using Maxwell's thermodynamic relation which is given below

$$\Delta S = \int_{0}^{H} (\partial M/\partial T) dH \tag{1}$$

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