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Folding and stability of integral membrane proteins in amphipols

ABSTRACT

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42 Introduction

Folding integral membrane proteins to their functionally active form:still a challenge

Integral membrane proteins - hereafter transmembrane pro-45 teins (TMPs)¹ – are key biological players because of their roles in 46 bioenergetics, transmembrane transport, signaling, cell and tissue 47 organization, etc. They are therefore primary targets for pharmaceu-48 tical drugs. Structural and pharmacological studies of TMPs are ham-49 50 pered by their low abundance and by difficulties in overexpressing them in a functional form. TMPs can be overexpressed either homol-51 52 ogously or heterologously, for example in Escherichia coli, yeast, etc. Plasmid-based overexpression can be designed to target TMPs either 53

to a membrane, which often results in low yields, or to cytoplasmic 54 inclusion bodies, which yields larger amounts of protein, but in a 55 misfolded and aggregated form that has to be folded to a functional 56 state - a difficult achievement. Folding outside the cell environment 57 is also necessary when TMPs are expressed in vitro in a cell-free sys-58 tem. Successful folding in detergent or mixed detergent/lipid 59 micelles has been demonstrated for both known structural classes 60 of TMPs, the α -helix bundle and the β -barrel. Over 50 α -helical TMPs 61 have been folded or refolded in vitro to date [1]. Among the first ones 62 to be studied were bacteriorhodopsin (BR) [2-4], the light harvesting 63 complex LHCII [5,6], several G protein-coupled receptors (GPCRs) 64 such as olfactory receptors [7,8], the BLT1 receptor of leukotriene 65 LTB_4 [9], or the serotonin 5- $HT_{4(a)}$ receptor [10], channel proteins 66 like the homotetramer KcsA [11], the small multidrug transporter 67

Amphipols (APols) are a family of amphipathic polymers designed to keep transmembrane proteins

(TMPs) soluble in aqueous solutions in the absence of detergent. APols have proven remarkably efficient

at (i) stabilizing TMPs, as compared to detergent solutions, and (ii) folding them from a denatured state to

a native, functional one. The underlying physical-chemical mechanisms are discussed.

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¹ Abbreviations used: A8-35, a poly(sodium acrylate)-based amphipol comprising ~35% free carboxylates, ~25% octyl grafts, ~40% isopropyl groups, whose number-average molar mass is ~4.3 kDa; APol, amphipol; BLT1 and BLT2, two human G protein-coupled receptors of leukotriene LTB₄; BO, BR, respectively bacterio-opsin and bacteriorhodopsin from *Halobacterium salinarum*; CAC, critical association concentration; CB1, human cannabinoid receptor 1; CD, circular dichroism; CHAPS, 3-[(3-cholamidopropyl)dimethyl-ammonio]-1-propanesulfonate; cmc, critical micellar concentration; 2D, 3D, two- and three-dimensional; DAGK, diacylglycerol kinase; dBO, delipidated BO, obtained after solubilizing purple membranes in organic solvents; DDM, *n*-dodecyl- β -*p*-maltoside; DHPC, dihexanoylphosphatidylcholine; DS, dodecylsulfate; EM, electron microscopy; ESI-IMS-MS, electrospray ionization mass spectrometry coupled with ion mobility spectrometry; FomA, major outer membrane protein A from *Fusbacterium nucleatum*; GHS-R1a, human ghrelin receptor 1a; GPCR, G protein-coupled receptor; His-tag, polyhistidine tag; 5-HT_{4(a)}, human serotonin receptor; LDAO, *N*-lauryl-*N*,*N*-dimethylamine-*N*-oxide; LHCII, light-harvesting complex II; LPS, lipopolysaccharide; MD, molecular dynamics; OG, *n*-octyl- β -*p*-glucoside; OmpA and OmpT, outer membrane proteins A and T from *Escherichia coli*; OTG, octylthioglucoside; PAGE, polyacrylamide gel electrophoresis; PagP, lipid A palmytoyltransferase from *E. coli*; PDS and SDS, potassium and sodium dodecylsulfate, respectively; PM, purple membrane; SEC, size-exclusion chromatography; SUV, small unilamellar vesicles; TMP, transmembrane protein; tOmpA, transmembrane domain of OmpA; UV, ultraviolet; Z3-14, 3-[dimethyl(tetradecyl)azaniumyl]propane-1-sulfonate (also called sulfobetaine 3-14, Zwittergent 3-14).

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68 EmrE [12], and the enzyme diacylglycerol kinase (DAGK) [13,14] (for 69 reviews, see e.g. Refs. [1,15,16]). About 40 β -barrel TMPs have been 70 (re)folded in vitro to date [1,17–20], ranging from simple 8-stranded 71 β -barrels like OmpA [21,22], OmpX [23] or PagP [24–26] to large 72 22-stranded active transporters like FepA [27,28]. The list also 73 includes eukaryotic β -barrels such as the 19-stranded VDAC human 74 isoforms 1 [29] and 2 [30,31]. It is quite possible that all β -barrel 75 TMPs can be folded in vitro without any cellular folding machinery. 76 An unsuccessful attempt to fold a β -barrel in solution has been reported for the β -barrel domain of AIDA [32]. However even for 77 78 AIDA, folding was successful when the protein was bound to a 79 nickel-bearing column [33].

80 α-Helical TMPs like BR or KcsA have been folded from denatured forms in organic solvent like trifluoroacetic acid or mixtures 81 82 of formic acid and ethanol [2,11]. Folding of BR was initiated by 83 transferring denatured bacterio-opsin (BO, the apoprotein) first 84 into micelles of the denaturing detergent sodium dodecylsulfate 85 (SDS) and then, usually in the presence of retinal, BR's cofactor, into 86 "mild" detergents, mixtures of detergent and lipids, or pure lipids 87 [2-4]. Rather than diluting it into a large excess of the folding med-88 ium [2,3], dodecylsulfate (DS) can be precipitated as its potassium 89 salt, PDS [4]. β -Barrel TMPs such as outer membrane protein A (OmpA, previously called Omp II) from E. coli were first refolded 90 91 from their SDS-denatured forms by replacing the SDS either with 92 lipopolysaccharide [21] or octylglucoside [22]. Subsequently, most 93 (re)folding protocols for β -barrel TMPs have been based on their 94 solubilization and denaturation in concentrated solutions of urea 95 (8–10 M) [34] or other chaotropic agents (reviewed in Ref. [1]). Denatured β -barrel TMPs are generally folded by transferring them 96 97 to either detergent micelles or lipid membranes under concurrent 98 strong dilution of the denaturant.

The mechanisms of folding of TMPs of both structural classes 99 100 have been the subject of extensive research. For α -helical TMPs, BR (see e.g. Refs. [35,36]) or DAGK [37,38] have often served as 101 102 models. Folding of α -helical membrane proteins is a sequential 3-103 stage process, in which helices insert independently in a first stage 104 and associate laterally in a second stage, which is followed, in a 105 third stage, by the formation of additional tertiary structure, such 106 as re-entrant loops, and/or by the binding of prosthetic groups 107 (for discussions, see Refs. [35,39]). For β -barrel TMPs, a concerted mechanism has been reported for OmpA, in which folding and 108 insertion are coupled [40,41] (for reviews, see Refs. [20,42-44]. 109

Using detergent- or lipid-based protocols, folding yields tend to 110 111 be lower than desired for many pharmacologically and physiologically interesting TMPs, e.g. GPCRs. For example, the leukotriene B₄ 112 113 BLT1 receptor has been refolded to a moderate yield of \sim 30% in 114 mixed micelles of detergent and lipid [9]. To date, this is still one 115 of the highest yields reported using classic methods. A homolog 116 of BLT1, the BLT2 receptor, is an example of a GPCR that cannot 117 be folded with a decent yield (more than a few percents) in any 118 detergent or detergent/lipid mixture that has been tried to date 119 [45]. The development of successful refolding strategies for novel TMPs of interest using classic approaches is extremely time-con-120 121 suming. It is therefore of great interest to examine new methods.

Amphipols as new tools for the refolding of integral membraneproteins

124 Over the last eight years, a new class of non-detergent surfac-125 tants, namely synthetic amphipathic polymers called amphipols 126 (APols) [46], has emerged as very promising tools for folding dena-127 tured TMPs to their native state. APols are short polymers com-128 prised of both hydrophilic groups and hydrophobic chains. They 129 can substitute to detergents, providing a milder environment to 130 TMPs while keeping them water-soluble (for reviews, see Refs. 131 [47–51]). All of the three dozen TMPs tested to date form soluble

complexes with APols in the absence of detergent, whatever their 132 size, origin, function and secondary, tertiary and guaternary struc-133 ture (reviewed in Ref. [51]). As compared to preparations in deter-134 gent solutions, most TMPs remain active for a much longer time 135 when solubilized in the form of APol/TMP complexes (see e.g. Refs. 136 [45,46,52–55]). APols bind to TMPs by adsorbing specifically onto 137 their hydrophobic transmembrane surface, as demonstrated by 138 NMR spectroscopy [56-60], electron microscopy (EM) [61-69], 139 and molecular dynamics (MD) simulations [70]. Unless exchanged 140 for other surfactants, APols do not dissociate from the surface of 141 TMPs, even at very high dilution, keeping the protein water-soluble 142 [47,71,72] (reviewed in Ref. [51]). 143

Several types of APols have been synthesized and validated (Fig. 1) (reviewed in Refs. [51,73]). Among these, A8-35 has been most extensively studied. A8-35 (Fig. 1A) is a poly(acrylic acid) partially amidated with octylamine and isopropylamine, leaving \sim 35% of the carboxylic groups free [46]. In aqueous solution at pH > 7, A8-35 assembles into small micelle-like particles with a mass of ~40 kDa [74-76]. Other APols developed more recently include glycosylated nonionic APols (NAPols) [77,78] (Fig. 1B and C), sulfonated APols (SAPols) [79] (Fig. 1D) or phosphorylcholinebased APols (PC-APols) [80,81] (Fig. 1E) (reviewed in Refs. [50,51,73]). SAPols [79] are derived from a precursor of A8-35 lacking the amidation with isopropylamine. Instead, about \sim 40% of the carboxyl groups are amidated by taurine (2-aminoethanesulfonic acid) (Fig. 1D). The sulfonate groups of taurine do not protonate even at pH 0, which keeps SAPols water-soluble at pH < 7, whereas under such conditions A8-35 aggregates [74,75]. Homopolymeric NAPols (Fig. 1B) are synthesized by homotelomerization of a monomer carrying two glycosyl residues [77]. Heteropolymeric NAPols (Fig. 1C) have been obtained either by cotelomerization of hydrophobic and hydrophilic monomers [82] or by randomly grafting hydrophobic chains onto a glycosylated homotelomer [83]. PC-APols are zwitterionic at neutral and basic pH, cationic at acidic pH. NAPols and PC-APols, as SAPols, remain water-soluble under acidic conditions.

APols, originally designed to keep TMPs water-soluble while 168 preserving their activity [46], have been used for a wide range of 169 applications (reviewed in Ref. [51]). Their properties and those of 170 TMP/APol complexes have been reviewed in Refs. [49-51] and 171 detailed protocols for their implementation provided in Ref. [84]. 172 This review focuses on their use for folding TMPs and for stabiliz-173 ing them against denaturation by heat or chaotropic agents. Since 174 α -helix bundle TMPs and β -barrel TMPs fold according to very dif-175 ferent principles (see e.g. Refs. [35,39,41,85]), our discussion is sub-176 divided into separate sections for these two categories of TMPs. 177 Historically, APol-assisted folding of TMPs was first demonstrated 178 for three model TMPs, BR, an α -helical TMP, and two β -barrel TMPs, 179 OmpA and FomA [86]. It was then extended to two more β -barrel 180 TMPs and six GPCRs (Table 1). 181

Amphipol-assisted folding of α -helix bundle membrane proteins

Folding of BR in A8-35

BR, a light-driven proton pump [87], has served as a popular 185 model protein in studies on TMP folding since the seminal work 186 of Khorana and coworkers [2] (for reviews, see e.g. Refs. 187 [1,35,88,89]). When overproduced, BR accumulates in the plasma 188 membrane of *H. salinarum* in the form of 2D protein/lipid crystals, 189 the so-called purple membrane (PM). Its covalently but loosely 190 bound chromophore, retinal, confers it a characteristic purple 191 color. When PM is solubilized in SDS, BR denatures to BO, releasing 192 its chromophore, which causes an absorption peak shift from 193

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