



## Review paper

How reliable is the intramolecular distribution of  $^{15}\text{N}$  in  $\text{N}_2\text{O}$  to source partition  $\text{N}_2\text{O}$  emitted from soil?

Charlotte Decock\*, Johan Six

Department of Environmental System Science, ETH-Zurich, Tannenstrasse 1, 8092 Zurich, Switzerland

## ARTICLE INFO

## Article history:

Received 10 May 2012

Received in revised form

9 April 2013

Accepted 12 May 2013

Available online 6 June 2013

## Keywords:

Stable isotopes

Site preference

Source-partitioning  $\text{N}_2\text{O}$ 

Nitrification

Denitrification

Nitrifier denitrification

Fungal denitrification

 $\text{N}_2\text{O}$  reductionGlobal  $\text{N}_2\text{O}$  budget

## ABSTRACT

$\text{N}_2\text{O}$  is a potent greenhouse gas and plays an important role in the depletion of stratospheric ozone. Hence, many efforts are now geared toward quantifying and mitigating  $\text{N}_2\text{O}$  emissions from soil in various ecosystems. This requires an in-depth understanding of the mechanisms and processes underlying  $\text{N}_2\text{O}$  emissions, which has been methodologically challenging. Recently, it has been suggested that the intramolecular distribution of  $^{15}\text{N}$  in the  $\text{N}_2\text{O}$  molecule (known as site preference or SP) can indicate which processes contribute to  $\text{N}_2\text{O}$  fluxes. Here, we assess, through guidance by a framework of recommended validation steps, the suitability of SP to source partition  $\text{N}_2\text{O}$  emitted from soils. In individual studies, significant effects of soil moisture content and soil type on SP values from soil-emitted  $\text{N}_2\text{O}$  have been observed, supporting that SP could be a useful tool to source-partition  $\text{N}_2\text{O}$  emitted from soil. While process-specific SP values based on pure culture studies have been used in isotope mixing and fractionation models to source partition  $\text{N}_2\text{O}$  in environmental samples, effects of confounding factors such as unaccounted pathways, microbial community composition, process rate, and soil heterogeneity remain poorly quantified. This urges continued research to determine SP values for distinct  $\text{N}_2\text{O}$  producing and consuming processes under controlled laboratory conditions for soils from a variety of ecosystems and environments. As mechanisms underlying  $\text{N}_2\text{O}$  production and consumption are plentiful and complex, creation of large isotope databases should be complemented with the development of more advanced models that take into account  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  of precursors, variability of overall isotope effects, and bulk  $\delta^{15}\text{N}$ ,  $\delta^{18}\text{O}$ , and SP of  $\text{N}_2\text{O}$ , as well as traditional proxies such as soil moisture content and C and N availability.

© 2013 Elsevier Ltd. All rights reserved.

## 1. Introduction

As greenhouse gas concentrations in the atmosphere continue to increase, much research is now directed toward strategies to mitigate global climate change. This includes attempts to reduce emissions of  $\text{N}_2\text{O}$ , a potent greenhouse gas and major player in the depletion of stratospheric ozone (Crutzen, 1981; IPCC, 2007). Soils under agricultural management as well as under natural vegetation constitute the major global source of  $\text{N}_2\text{O}$  (Denman et al., 2007).  $\text{N}_2\text{O}$  emissions from soil are characterized by high temporal and

spatial variability and underlain by a multitude of microbial and abiotic pathways (Wrage et al., 2001; Hayatsu et al., 2008; Hénault et al., 2012). A better understanding of pathways causing  $\text{N}_2\text{O}$  fluxes at any given time or place could greatly advance the development of more targeted mitigation strategies and narrow the uncertainty around  $\text{N}_2\text{O}$  emission predictions (Wrage et al., 2001).

Hydroxylamine ( $\text{NH}_2\text{OH}$ ) oxidation, nitrifier-denitrification, denitrification and  $\text{N}_2\text{O}$  reduction to  $\text{N}_2$  are considered the major processes controlling  $\text{N}_2\text{O}$  emissions (Firestone and Davidson, 1989; Wrage et al., 2001).  $\text{NH}_2\text{OH}$ -oxidation-derived  $\text{N}_2\text{O}$  is  $\text{N}_2\text{O}$  that is created as a byproduct during  $\text{NH}_4^+$  oxidation to  $\text{NO}_2^-$  by autotrophic ammonium-oxidizers (Hooper and Terry, 1979; Chalk and Smith, 1983). The same organisms are capable of reducing  $\text{NO}_2^-$  to  $\text{N}_2\text{O}$  and/or  $\text{N}_2$  under high  $\text{NO}_2^-$  partial pressure or low oxygen conditions, a process known as nitrifier-denitrification (Wrage et al., 2001).  $\text{N}_2\text{O}$  is also produced as an intermediate during the reduction of  $\text{NO}_3^-$  to  $\text{N}_2$  by heterotrophic denitrifying bacteria under anaerobic conditions, i.e. denitrification-derived  $\text{N}_2\text{O}$  (Knowles, 1982). Moreover, alternative pathways such as

List of abbreviations: SP, site preference; AOA, ammonia-oxidizing archaea; AOB, ammonia-oxidizing bacteria;  $\text{N}_2\text{O}_\text{N}$ ,  $\text{N}_2\text{O}$  produced by processes for which SP is around 32.8‰, including hydroxylamine oxidation by ammonia-oxidizing bacteria, fungal denitrification, abiotic  $\text{N}_2\text{O}$  production and  $\text{N}_2\text{O}$  production by ammonia-oxidizing archaea;  $\text{N}_2\text{O}_\text{D}$ ,  $\text{N}_2\text{O}$  produced by processes for which SP is around −1.6‰, including denitrification and nitrifier-denitrification.

\* Corresponding author.

E-mail address: [charlotte.decock@usys.ethz.ch](mailto:charlotte.decock@usys.ethz.ch) (C. Decock).

fungal denitrification, abiotic N<sub>2</sub>O production, archaeal N<sub>2</sub>O production, heterotrophic nitrification and co-denitrification could contribute to the N<sub>2</sub>O flux from soils (Laughlin and Stevens, 2002; Venterea, 2007; Hayatsu et al., 2008; Schleper, 2010; Spott et al., 2011).

Given the highly variable emission patterns and the complexity of N<sub>2</sub>O production and consumption pathways, source-partitioning N<sub>2</sub>O is methodologically challenging at both the ecosystem and global scale (Groffman et al., 2006; Baggs, 2008). Previous efforts to identify N<sub>2</sub>O production and consumption pathways have mostly relied on C<sub>2</sub>H<sub>2</sub> inhibition and isotope labeling techniques (Baggs, 2008). These methods have some important limitations, including difficulties with homogeneous diffusion of the label or inhibitor, disturbance of the system during label or inhibitor application, and a narrow time frame during which N<sub>2</sub>O production and consumption can be monitored before the label or inhibitor turns over (Groffman et al., 2006; Ostrom and Ostrom, 2011). Variation in the natural abundance <sup>15</sup>N and <sup>18</sup>O of N<sub>2</sub>O has been explored as a non-invasive method to assess N<sub>2</sub>O production and reduction mechanisms, but interpretation of bulk  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  of N<sub>2</sub>O is challenging because of the dependency on  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  of the precursors and the uncertainty around isotope fractionation factors for various processes (Perez, 2005).

It has been suggested that the intramolecular distribution of <sup>15</sup>N in the N<sub>2</sub>O molecule could serve as a tool to discern various N<sub>2</sub>O producing and consuming processes and to help constrain the global N<sub>2</sub>O budget (Yoshida and Toyoda, 2000; Perez et al., 2001). Molecules with differing distribution of <sup>15</sup>N in N<sub>2</sub>O are sometimes referred to as 'isotopomers', i.e. molecules of the same mass in which trace isotopes are arranged differently (Ostrom and Ostrom, 2011). The term 'isotopologue' more generally refers to molecules that differ in their isotopic composition (Ostrom and Ostrom, 2011). Since N<sub>2</sub>O is an asymmetric molecule, a central and terminal N atom can be distinguished (Toyoda and Yoshida, 1999). The central and terminal N atoms have been referred to as  $\alpha$  and  $\beta$  (Toyoda and Yoshida, 1999) or 2 and 1 (Brenninkmeijer and Röckmann, 1999), respectively. In addition, the terminology 456 and 546 has been used to refer to the isotopomers <sup>14</sup>N<sup>15</sup>N<sup>16</sup>O and <sup>15</sup>N<sup>14</sup>N<sup>16</sup>O, respectively (Stein and Yung, 2003). In this review, we will use the terminology  $\alpha$  and  $\beta$  because this terminology is most prevalent in studies focused on source partitioning N<sub>2</sub>O emitted from soil. The site-specific distribution of <sup>15</sup>N in N<sub>2</sub>O has been expressed as site-preference (SP), and is calculated as the difference in  $\delta^{15}\text{N}$  in the central compared to the terminal N atom ( $\text{SP} = \delta^{15}\text{N}_{\alpha} - \delta^{15}\text{N}_{\beta}$ ) (Toyoda and Yoshida, 1999). A major advantage of the use of SP is the minimal disturbance associated with sampling, facilitating experimentation under field conditions (Ostrom and Ostrom, 2011). Determining SP could complement  $\delta^{15}\text{N}_{\text{bulk}}$  (average <sup>15</sup>N content) and  $\delta^{18}\text{O}$  data of N<sub>2</sub>O, with the additional advantage that SP is presumed to be independent of  $\delta^{15}\text{N}$  of the precursors (Toyoda et al., 2002). The <sup>15</sup>N content in the  $\alpha$  and  $\beta$  positions can be determined based on mass analyses of the molecular (N<sub>2</sub>O<sup>+</sup>) and fragment (NO<sup>+</sup>) ions of N<sub>2</sub>O on an isotope ratio mass spectrometer (Toyoda and Yoshida, 1999). In addition, quantum cascade laser based absorption spectroscopy to determine the intramolecular distribution of <sup>15</sup>N within the N<sub>2</sub>O molecule is under development (Waechter et al., 2008; Mohn et al., 2010). An extensive overview of analytical methods and challenges associated with measuring SP can be found in Ostrom and Ostrom (2011).

During N<sub>2</sub>O production, two NO<sup>•</sup> ions bind to form the intermediate hyponitrite (<sup>•</sup>O–N=N–O<sup>•</sup>), followed by cleavage of one N–O bond (Stein and Yung, 2003). An N–O bond is also broken during N<sub>2</sub>O reduction to N<sub>2</sub> (Zumft, 1992). Ultimately, preferential cleavage of bonds between lighter atoms determines the degree of

SP and can vary between microbial groups and/or enzymes involved (Stein and Yung, 2003; Schmidt et al., 2004). This variation across microbial groups or enzymes forms the basis for the use of SP to source partition N<sub>2</sub>O emitted from soil. Values of SP have been determined for N<sub>2</sub>O production and consumption pathways in pure microbial cultures and soil incubation experiments (Sutka et al., 2006; Well et al., 2006). Partial compilations of observed SP values are provided in Park et al. (2011), Toyoda et al. (2011) and Well et al. (2012). Values of SP observed in pure cultures have been used in a number of studies to interpret SP values observed in environmental samples (Opdyke et al., 2009; Toyoda et al., 2011). In a recent review, Ostrom and Ostrom (2011) promoted the use of SP values from pure cultures and a graphical analysis based on  $\delta^{18}\text{O}$  and  $\delta^{15}\text{N}$  of N<sub>2</sub>O to estimate the contribution of different N<sub>2</sub>O producing and consuming pathways to N<sub>2</sub>O emissions from environmental samples. However, before SP can be used with confidence for source partitioning N<sub>2</sub>O at ecosystem and global scales, a sequence of validation steps is warranted (Fig. 1). These include validation of the robustness of SP values presumed to be characteristic for various N<sub>2</sub>O consuming and producing processes, identification of confounding factors and quantification of their effect, development and evaluation of models and strategies to interpret SP data, and assessment of the uncertainty in sources of N<sub>2</sub>O estimated using SP. We synthesize here the progress that has been made in each of those validation steps, leading up to recommendations for future research.

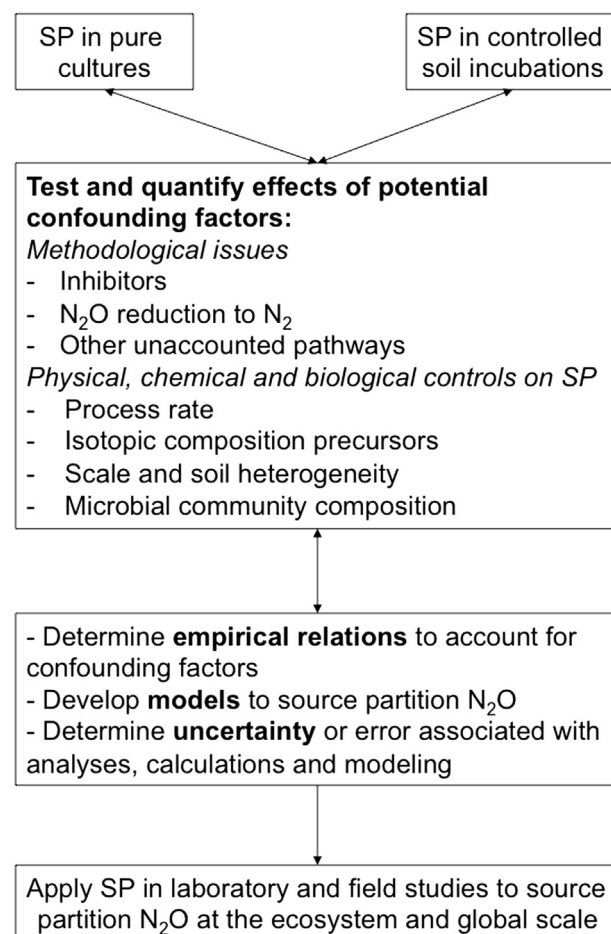


Fig. 1. Flowchart for SP method validation.

Download English Version:

<https://daneshyari.com/en/article/8365204>

Download Persian Version:

<https://daneshyari.com/article/8365204>

[Daneshyari.com](https://daneshyari.com)