



## Short communication

## Treatment of pasture topsoil with alum to decrease phosphorus losses in subsurface drainage

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## ABSTRACT

Phosphorus loss from land can impair surface water quality. Losses *via* subsurface flow can be substantial, but most strategies to mitigate P losses focus on surface runoff. Aluminium sulphate (alum) was applied at 25 and 50 kg Al ha<sup>-1</sup> to a flat, podzol soil under pasture regularly grazed by dairy cattle. Over a year, losses of filtered (<0.45 μm) reactive P (FRP) and total filtered P (TFP) intercepted at 35-cm depth by Teflon suction cups were c. 0.6 and 1.0 kg P ha<sup>-1</sup>, respectively for the control treatment. The 50 kg Al ha<sup>-1</sup> treatment decreased FRP and TFP by 26 and 27%, respectively: no significant difference to the control was noted for alum applied at 25 kg Al ha<sup>-1</sup>. The cost-effectiveness was estimated at 190–952 USD kg<sup>-1</sup> P mitigated. While more cost-effective strategies should be practised first, surface applying alum may provide an option where sub-surface P losses must be lowered further especially if applied to a small area of high P loss.

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## 1. Introduction

The loss of P from land can impair surface water quality *via* eutrophication (Carpenter *et al.*, 1998). In a review of New Zealand land uses, McDowell and Wilcock (2008) found that P losses were enriched from dairy-farmed land. Under certain soil physical (*e.g.* low slope) or chemical (*e.g.* poor P sorption capacity) conditions, subsurface flow can be a more important mechanism of P loss than surface runoff. In the New Zealand soil classification (Hewitt, 1998), many podzol soils are prone to subsurface P losses (*viz.* leaching) due to a combination of high hydraulic conductivity, poor anion storage capacity (ASC: an indicator of P sorption capacity; McDowell and Condon, 2004), and macropore flow especially if intercepted by artificial drainage (Monaghan and Smith, 2004).

Strategies to mitigate P loss from dairy systems have focused on minimising losses by surface runoff. However, there are far fewer strategies to mitigate P losses *via* subsurface flow. In a review of potential technologies to remove P in drainage water, Buda *et al.* (2012) highlighted several structures that filter-out P from drainage water such as flue gas desulfurization gypsum ditch filters (Bryant *et al.*, 2012), the use of iron oxides in and around tile drains (McDowell *et al.*, 2008; Chardon *et al.*, 2012), and steel slag filter beds to remove P from drainage water at catchment outlets

(Penn *et al.*, 2012). However, McDowell and Nash (2012) also noted that as scale and complexity of flow-paths increased the cost-effectiveness of strategies to mitigate P loss decreased. Strategies that aim to decrease P loss most cost-effectively focus on decreasing the availability of P at source; for example, by decreasing soil Olsen P concentration (McDowell and Nash, 2012).

In many soils, especially those used for intensive pastoral grazing, maintaining soil Olsen P as low as agronomically possible may still lead to substantial P loss *via* subsurface flow due to poor P sorption capacity and continual inputs from excretal returns (McDowell and Nash, 2012). One amendment, aluminium sulphate (alum) has been used to decrease the potential for P loss in surface runoff (*e.g.* McDowell and Norris, 2014). Alum was therefore tested to determine if it could also be used to cost-effectively decrease P loss in subsurface flow from a grazed pasture.

## 2. Methods and materials

## 2.1. Preliminary leaching trial

A preliminary leaching trial was conducted using a Tisbury silt loam soil (Typic Perch-gley podzol in the New Zealand soil classification – equivalent to an Aquod in US Taxonomy; Hewitt, 1998) from pasture-based grazed dairy farm >10 years old. Soil was sampled of the 0–7.5-cm depth, broken up by hand to remove plant roots and passed through a 2-mm sieve. A sub-sample was dried and analysed for ASC (Saunders, 1965) as a measure of the soil's

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**Table 1**

Soil test values with depth at the field site. The least significant difference at the  $P < 0.05$  level of significance ( $LSD_{05}$ ) is given for the comparison of means for each depth.

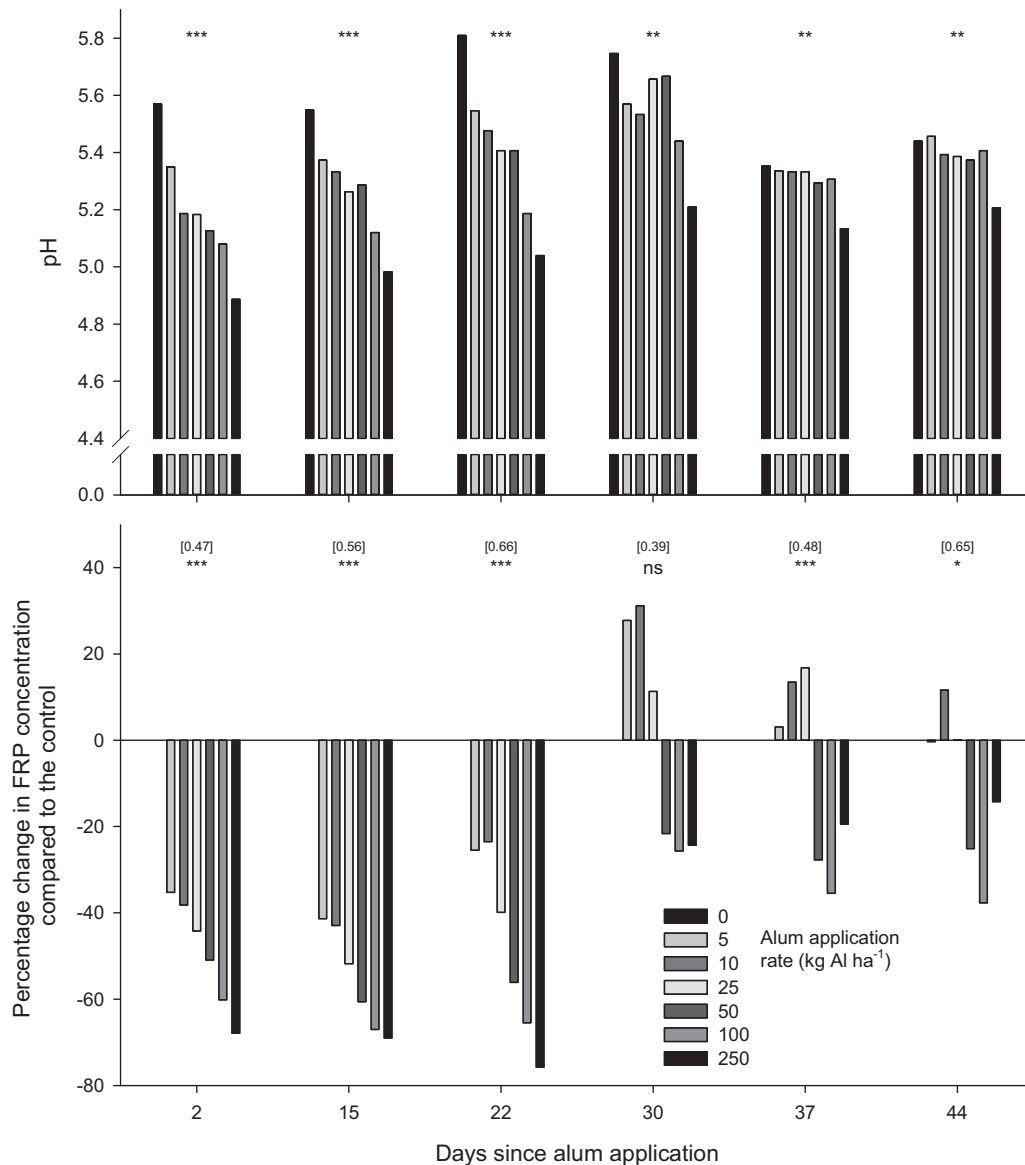
Depth (cm)	Soil test			
	Olsen P ( $\text{mg L}^{-1}$ soil)	WSP ( $\text{mg L}^{-1}$ extract)	ASC (%)	pH
0–7.5	51	0.18	47	5.7
7.5–15	46	0.15	61	5.0
15–30	31	0.10	60	4.9
30–45	21	0.09	60	4.7
45–60	11	0.05	82	4.5
60–100	4	0.02	91	4.3
$LSD_{05}$	5	0.03	12	0.3

P-sorbing Al and Fe oxide concentration, Olsen P concentration (Olsen et al., 1954) and water extractable P (WEP; McDowell and Condon, 2004). Eighty grams of field moist soil was placed inside  $100\text{-cm}^3$  syringes (3-cm diameter) with some glass wool at the end to prevent soil loss. Alum was applied at rates of 0, 32, 64, 160, 320,

640,  $1600\text{ kg ha}^{-1}$ , equivalent to 0, 5, 10, 25, 50, 100,  $250\text{ kg Al ha}^{-1}$  to the surface of 6 replicates each – yielding 42 syringes in total. Soils were leached with 40-mm of deionised over 8 h, 2, 15, 22, 29, 37 and 44 days after alum was applied. Leachate samples passed through a  $0.45\text{ }\mu\text{m}$  filter and analysed for filtered reactive P (FRP) and pH (1:2 soil to water ratio).

## 2.2. Field trial

Alum was applied in October 2012 at 0, 25 and  $50\text{ kg Al ha}^{-1}$  to the same paddock used to collect the Tisbury silt loam (Table 1). The site had a slope of  $<1\%$  and artificial drains installed at 70–80-cm depth to facilitate drainage to adjacent 1-m deep open channel drains. The paddock received regular annual fertiliser of a single application of  $32\text{ kg P ha}^{-1}$ ,  $45\text{ kg K ha}^{-1}$  and  $100\text{ kg Ca ha}^{-1}$  (as lime) as part of the wider dairy farm's milking platform in December. Split applications of  $30\text{ kg N ha}^{-1}$  as urea were applied in September, November, January and March. Paddocks on the milking platform were rotationally grazed every 24–28 days



**Fig. 1.** Mean leachate pH and percentage change compared to the control for filtered reactive P (FRP) concentration with time after different rates of aluminium sulphate was applied. Values in parentheses are the mean concentration of FRP in leachate for the control. An asterisk indicates a significant difference in mean FRP concentrations between rates for each event (\*\* and \*\*\* indicate significance at the  $P < 0.01$  and  $<0.001$  level, respectively).

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