



# Streptoxamine, an unprecedented benzoisindole-deferoxamine hybrid from the locust-derived *Streptomyces* sp. HKHCa2

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## ABSTRACT

An unusual benzoisindole-deferoxamine hybrid, streptoxamine (1), has been isolated from the ethyl acetate crude extract of the fermentation broth of a locust-associated actinomycete, *Streptomyces* sp. HKHCa2, which was isolated from an insect, *Oxya chinensis*. The structure of this secondary metabolite was elucidated on the basis of its one-dimension, two-dimension NMR, and mass spectroscopic data. This natural product features a hybrid pattern of a benzoisindole with an “iron carrier” deferoxamine B through C-N linkage. Compound 1 showed weak antibacterial activity against the gram-positive bacteria, *Staphylococcus aureus* and *Mycobacterium smegmatis*.

## 1. Introduction

Insects, possessing the most rich groups, are often associated with symbiotic microorganisms which provide nutrition, immunological regulators, antibiotics, etc. to support their host survival and adaptation [1,2]. In the insect-symbiont associations, the host organisms often harbor specific microbial symbionts [3], providing a tremendous discovery opportunity for isolating new species of bacteria and fungi. Explorations of the microbes in symbiotic insect ecosystems can give new insights into the natural products discovery and have led to the discovery of plentiful of bioactive natural products with novel chemical scaffolds. For example, a series of immunosuppressive polyketides with different novel carbon skeletons were reported from the fungus *Daldinia eschscholzii* IFB-TL01 isolated from the gut of the mantis species *Tenodera aridifolia* [4,5]. Bacterial symbionts with insect are also a promising source of novel bioactive natural products and a great many bioactive metabolites from actinomycete are being reported such as the Na<sup>+</sup>/K<sup>+</sup>-ATPase inhibitor, coprisidin A, and NAD(P)H:quinone oxidoreductase inducer, coprisidin B from a dung beetle (*Copris tripartitus*)-associated *Streptomyces* strain [6], and antimicrobial 20-membered glycosylated macrolactams, macrotermycins A-D from a termite-associated *Amycolatopsis* sp. M39 [7]. In addition, *Deinococcus* sp. from the carpenter ant *Camponotus japonicus* and *Microbacterium* sp. from the carrion beetle *Nicrophorus concolor* produced novel aminoglycolipids

deinococcucins A-D and chlorinated cyclic peptides nicrophorusamides A and B, respectively [8,9]. Our previous chemical investigation of insect-associated actinomycetes also led to the discovery of chemically intriguing bioactive molecules like novel enediyne-derived compounds amycolamycins A and B, and unusual macrolactams rifamorpholines A-E from the locust (*Locusta migratoria*)-associated rare actinobacterium *Amycolatopsis* sp. Hca4, and tetrasaccharide derivatives actinotetraoses A-H and angucyclines from *Amycolatopsis* sp. HCa1 derived from the locust, *Oxya chinensis* [10–14]. In our continuing efforts to search for insect-associated bacterial symbionts, recently we isolated an actinomycete strain *Streptomyces* sp. HKHCa2 from a healthy insect *Oxya chinensis*, collected from Haikou, People's Republic of China. Subsequent chemical study on the large-scale fermentation broth by a combination of column chromatography led to the isolation and identification of an unusual benzoisindole-deferoxamine hybrid, streptoxamine (1) (Fig. 1). This compound features a hybrid pattern of a benzoisindole with an “iron carrier” deferoxamine B through C-N linkage and showed weak antibacterial activity against the gram-positive bacteria, *Staphylococcus aureus* and *Mycobacterium smegmatis*. Herein, we report the isolation and structure elucidation of this novel compound, as well as its antibacterial activity.

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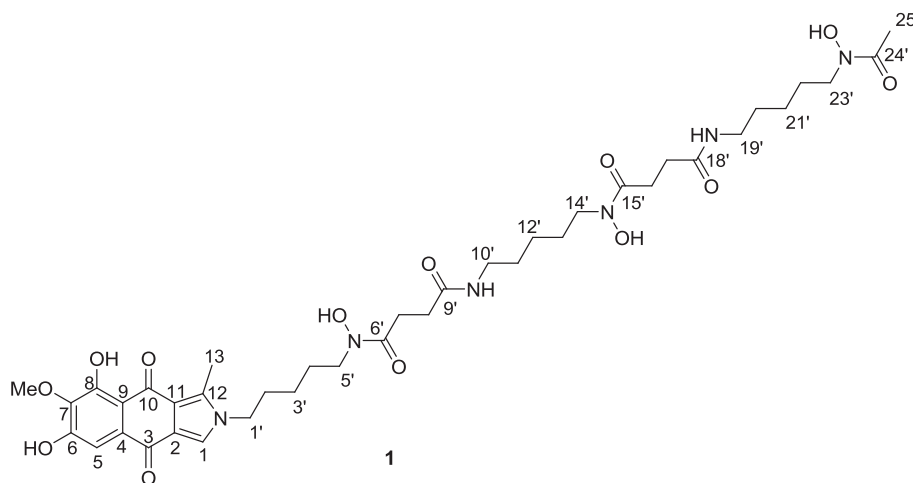


Fig. 1. Structure of streptoxamine (1).

## 2. Results and discussion

Streptoxamine (1) was isolated as a yellow oil with the molecular formula of  $C_{39}H_{56}N_6O_{13}$  derived from the high-resolution electrospray ionization mass spectrum (HR-ESI-MS) and  $^{13}C$  NMR data (Table 1). The  $^{13}C$  NMR and DEPT135 spectra of 1 displayed signals of thirty nine carbons, including seven carbonyl carbons ( $\delta_C$  186.1–170.6), two aromatic/olefinic methine carbons ( $\delta_C$  126.1, 108.8), eight non-protonated aromatic/olefinic carbons ( $\delta_C$  157.8–111.5), two methyl carbons ( $\delta_C$  20.8, 11.3), one oxygenated methyl carbons ( $\delta_C$  60.3), and nineteen aliphatic methylene carbons ( $\delta_C$  47.5–23.4). The  $^1H$  NMR spectrum of 1 in DMSO- $d_6$  exhibited signals of two methyls at  $\delta_H$  2.59 (3H, s, H-13) and 1.95 (3H, s, H-25'), one oxygenated methyls at  $\delta_H$  3.79 (3H, s, 7-OMe), two aromatic protons [ $\delta_H$  7.63 (1H, s, H-1) and 7.08 (1H, s, H-5)], nineteen aliphatic methylenes ( $\delta_H$  3.96–1.20), two hydroxyl groups at  $\delta_H$  13.62 (1H, br s, 8-OH) and 9.63 (1H, br s, 6-OH), and two NH groups at  $\delta_H$  7.76 (1H, t,  $J = 5.6$ , 10'-NH) and 7.74 (1H, t,  $J = 5.6$ , 19'-NH). These data above showed that 1 could contain three other *N*-hydroxyl moieties and three rings.

The complete assignments for all protons and carbons, as shown in

**Table 1**  
 $^1H$  (700 MHz) and  $^{13}C$  NMR (175 MHz) spectroscopic data for streptoxamine (1) in DMSO- $d_6$ .

| Position | $\delta_H$ (mult, $J$ in Hz) | $\delta_C$            | Position | $\delta_H$ (mult, $J$ in Hz) | $\delta_C$            |
|----------|------------------------------|-----------------------|----------|------------------------------|-----------------------|
| 1        | 7.63, s                      | 126.1, CH             | 7'       | 2.25, m                      | 30.3, CH <sub>2</sub> |
| 2        |                              | 120.8, C              | 8'       | 2.56, m                      | 28.0, CH <sub>2</sub> |
| 3        |                              | 178.5, C              | 9'       |                              | 171.7, C              |
| 4        |                              | 131.7, C              | 10'      | 2.99, m                      | 38.9, CH <sub>2</sub> |
| 5        | 7.08, s                      | 108.8, CH             | 10'-NH   | 7.76, t ( $J = 5.6$ )        |                       |
| 6        |                              | 157.8, C              | 11'      | 1.37, m                      | 29.3, CH <sub>2</sub> |
| 6-OH     | 9.63, br s                   |                       | 12'      | 1.20, m                      | 24.0, CH <sub>2</sub> |
| 7        |                              | 139.3, C              | 13'      | 1.49, m                      | 26.5, CH <sub>2</sub> |
| 7-OMe    | 3.79, s                      | 60.3, CH <sub>3</sub> | 14'      | 3.44, t ( $J = 7.0$ )        | 47.5, CH <sub>2</sub> |
| 8        |                              | 157.8, C              | 15'      |                              | 172.4, C              |
| 8-OH     | 13.62, s                     |                       | 16'      | 2.25, m                      | 30.3, CH <sub>2</sub> |
| 9        |                              | 111.5, C              | 17'      | 2.56, m                      | 28.0, CH <sub>2</sub> |
| 10       |                              | 186.1, C              | 18'      |                              | 171.7, C              |
| 11       |                              | 117.2, C              | 19'      | 2.99, m                      | 38.9, CH <sub>2</sub> |
| 12       |                              | 137.8, C              | 19'-NH   | 7.74, t ( $J = 6.0$ )        |                       |
| 13       | 2.59, s                      | 11.3, CH <sub>3</sub> | 20'      | 1.37, m                      | 29.3, CH <sub>2</sub> |
| 1'       | 3.96, t ( $J = 7.0$ )        | 46.9, CH <sub>2</sub> | 21'      | 1.20, m                      | 24.0, CH <sub>2</sub> |
| 2'       | 1.71, p ( $J = 7.0$ )        | 29.8, CH <sub>2</sub> | 22'      | 1.49, m                      | 26.5, CH <sub>2</sub> |
| 3'       | 1.23, m                      | 23.4, CH <sub>2</sub> | 23'      | 3.44, t ( $J = 7.0$ )        | 47.2, CH <sub>2</sub> |
| 4'       | 1.54, p ( $J = 7.0$ )        | 26.2, CH <sub>2</sub> | 24'      |                              | 170.6, C              |
| 5'       | 3.47, t ( $J = 7.0$ )        | 47.2, CH <sub>2</sub> | 25'      | 1.95, s                      | 20.8, CH <sub>3</sub> |
| 6'       |                              | 172.4, C              |          |                              |                       |

Table 1, were deduced by comprehensive analysis of HSQC,  $^1H$ - $^1H$  COSY, HMBC, and ROESY spectra (Fig. 2). In 1, The naphthoquinone substructure could be identified by the observation of HMBC correlations from 8-OH to C-7 ( $\delta_C$  139.3), C-8 ( $\delta_C$  157.8) and C-9 ( $\delta_C$  111.5), from 7-OMe to C-7, from H-5 to C-3 ( $\delta_C$  178.5), C-4 ( $\delta_C$  131.7), C-6 ( $\delta_C$  157.8), C-7 and C-9, from H-1 to C-2 ( $\delta_C$  120.8), C-3 and C-11 ( $\delta_C$  117.2), from H<sub>3</sub>-13 to C-11 and C-12 ( $\delta_C$  137.8). The  $^1H$ ,  $^1H$  three-bond couplings observed in the COSY spectrum from H<sub>2</sub>-2' to H<sub>2</sub>-1' and H<sub>2</sub>-3', from H<sub>2</sub>-4' to H<sub>2</sub>-3' and H<sub>2</sub>-5', from H<sub>2</sub>-11' to H<sub>2</sub>-10' and H<sub>2</sub>-12', from H<sub>2</sub>-13' to H<sub>2</sub>-12' and H<sub>2</sub>-14', from 10'-NH to H<sub>2</sub>-10', from H<sub>2</sub>-20' to H<sub>2</sub>-19' and H<sub>2</sub>-21', from H<sub>2</sub>-22' to H<sub>2</sub>-21' and H<sub>2</sub>-23', from 19'-NH to H<sub>2</sub>-19', together with the chemical shifts of the  $^{13}C$  resonances (C-1', C-5', C-10', C-14', C-19' and C-23') observed at relative higher field ( $\delta_C$  47.5–38.9), revealed the presence of three mono-*N*-hydroxylated diaminopentane moieties [15].  $^1H$ - $^1H$  COSY correlations from H<sub>2</sub>-7' ( $\delta_H$  2.25, m) to H<sub>2</sub>-8' ( $\delta_H$  2.56, m) and from H<sub>2</sub>-16' ( $\delta_H$  2.25, m) to H<sub>2</sub>-17' ( $\delta_H$  2.56, m) and HMBC correlations from H<sub>2</sub>-7' to C-9' ( $\delta_C$  171.7), from H<sub>2</sub>-8' to C-6' ( $\delta_C$  172.4), from H<sub>2</sub>-17' to C-15' ( $\delta_C$  172.4) and from H<sub>2</sub>-16' to C-18' ( $\delta_C$  171.7) identified two unsymmetrical succinyl residues in 1.

In the HMBC spectrum, correlations from H<sub>2</sub>-5' ( $\delta_H$  3.47, t,  $J = 7.0$ ) to C-6', from 10'-NH to C-9' and C-10' ( $\delta_C$  38.9), from H<sub>2</sub>-10' ( $\delta_H$  2.99, m) to C-9', from H<sub>2</sub>-14' to C-15', from 19'-NH to C-18' and C-19' ( $\delta_C$  38.9), and from H<sub>2</sub>-19' ( $\delta_H$  2.99, m) to C-18' suggested that each of the two succinyl residues was linked to two pentanediamine fragments via amide bonds. An acetyl group was connected with the 23'-*N*-hydroxyl on the basis of the HMBC correlations from H<sub>2</sub>-23' ( $\delta_H$  3.44, t,  $J = 7.0$ ) and H<sub>3</sub>-25' to C-15' ( $\delta_C$  170.6). These deductions above were also supported by the ROESY correlations from 10'-NH to H<sub>2</sub>-8', H<sub>2</sub>-10' and H<sub>2</sub>-11', from 19'-NH to H<sub>2</sub>-17', H<sub>2</sub>-19' and H<sub>2</sub>-20'. The key HMBC correlations observed from H<sub>2</sub>-1' ( $\delta_H$  3.96, t,  $J = 7.0$ ) to C-1 and C-12 suggested that the nitrogen on the pentanediamine fragment was linked to C-1 and C-12 on the naphthoquinone substructure, thus forming the last ring and completing the structure of 1. This connectivity was also secured by the observation of the ROESY correlations from H-1 and H<sub>3</sub>-13 to H<sub>2</sub>-1' and H<sub>2</sub>-2'. Therefore, the complete structure of streptoxamine (1) was determined as shown in Fig. 1. This unusual natural product features a hybrid pattern of a naphthoquinone with an “iron carrier” deferoxamine B through C-N linkage, which could be generated by non-enzymatic condensation of the terminal amino group in deferoxamine B with the isofurano moiety with the same strategy for pyridine ring biosynthesis [16,17].

Compound 1 was assessed for antibacterial activity against selected Gram-positive *Staphylococcus aureus*, *Bacillus subtilis*, and *Mycobacterium smegmatis*. As a result, it only displayed weak antibacterial activity

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