



Flavonoid glycosides and alkaloids from the embryos of *Nelumbo nucifera* seeds and their antioxidant activity

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ABSTRACT

Chemical investigation of the embryos of *Nelumbo nucifera* afforded four new flavone C-glycosides, named nelumbosides A–D (1–4), together with nine known ones, comprising five flavonoids (5–9) and four alkaloids (10–13). The chemical structures of the new compounds were elucidated by 1D, 2D-NMR and HR-ESI-MS techniques, together with chemical methods. Nelumbosides A–D (1–4) are rarely present in naturally occurring flavone C-glycosides featuring a 4-hydroxystyrene unit connected to the flavonoid skeleton. Compounds 2–13 were evaluated for their antioxidant activity by ABTS and DPPH radical-scavenging assay. Among them, compounds 2, 6, 7 and 11 exhibited strong scavenging activity with SC₅₀ values ranging from 12.07 to 25.68 μM compared with the positive control L-ascorbic acid.

1. Introduction

Nelumbo nucifera GAERTNER, considered to be an economically important perennial aquatic crop, has been cultivated all over China [1]. Many parts of this plant, such as leaves, flowers, rhizomes and embryos, have been used as common vegetable, tea and healthcare product [2]. Meanwhile, the different organs of this plant have also been used in traditional herbal medicine since ancient times to cure different diseases [3]. The embryos of *N. nucifera* seeds, also called “Lian Zi Xin”, were primarily used to remove heat from the heart, anchor the mind, improve spermatorrhea and arrest bleeding [4]. Modern pharmaceutical studies showed that alkaloids from the embryos of *Nelumbo nucifera* seeds possessed various biological activities, including anti-tumor [5–6], anti-inflammatory [7], anti-oxidation [8–9] and obvious sedation activities [10–11].

Previous phytochemical studies on the embryos of *N. nucifera* seeds indicated the presence of alkaloids [11–13], sterols and fatty alcohols [14]. A recent study reported the profiling of flavonoids in Lotus (*Nelumbo nucifera*) by HPLC-DAD and HPLC-ESI-MSⁿ [15]. However, detailed phytochemical study on the flavonoids in this plant has rarely been reported. Moreover, flavonoids and alkaloids in this plant were considered to be the bioactive components. The lack of chemical

entities greatly limited biological evaluation of this important resource. Therefore, the present study focused on the investigation of these two kinds of compounds in the embryos of *N. nucifera* seeds in order to provide a scientific foundation for further use of this plant. As a result, four novel flavone C-glycosides (1–4), with an unusual structure featuring a 4-hydroxystyrene unit connected to the flavonoid skeleton, together with nine known compounds (5–13) were isolated from the embryos of *N. nucifera* seeds (Fig. 1). Herein, the isolation and identification of these compounds and their antioxidant activities against ABTS⁺ and DPPH assay are reported.

2. Experimental

2.1. General experimental procedures

Optical rotations were determined on a Perkin-Elmer 341 polarimeter at room temperature. IR spectra were carried out on a Perkin-Elmer 1725X-FT spectrometer with KBr disks. UV spectra were taken in MeOH on a Shimadzu UV 260 Spectrometer. ¹H NMR and ¹³C NMR, HSQC and HMBC spectra were recorded with a Bruker Avance-600 spectrometer and a Bruker Avance-400 spectrometer (Bruker Corporation, Stockholm, Sweden). The chemical shift (δ) values are

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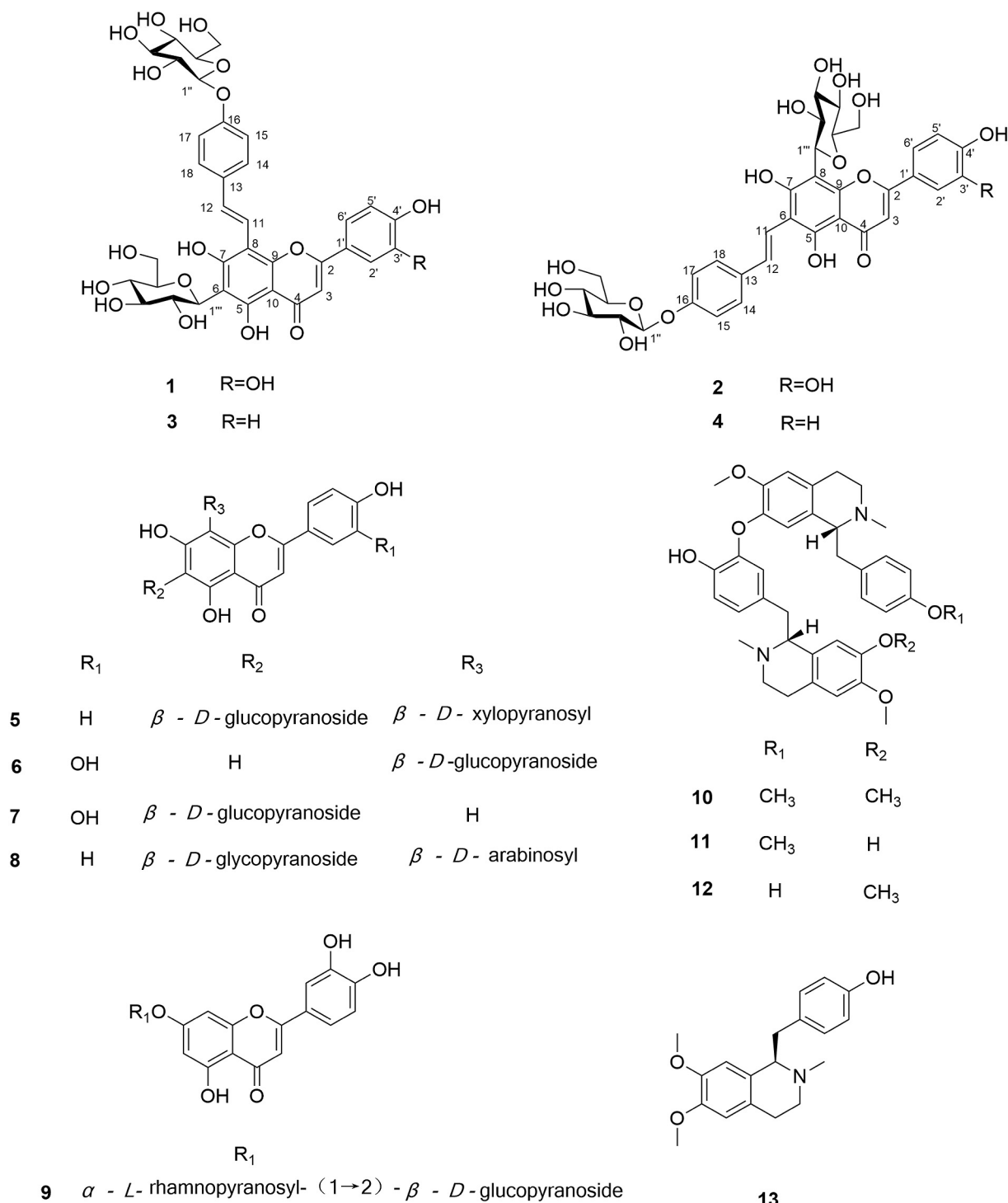


Fig. 1. Structures of compounds 1–13.

given in ppm with TMS as internal standard, and coupling constants (J) are in Hz. ESI-MS and HR-ESI-MS were measured on a Finnigan LCQ^{DECA} (San Jose, CA, USA) and a Bruker Apex-III mass spectrometer (Bruker Corporation, Stockholm, Sweden) respectively. Preparative HPLC was performed using Ultimate * XB-C₁₈ (250 mm \times 21.2 mm, 10 μ m) preparative column. ABTS and DPPH free radical-scavenging activities were measured by using Thermo Scientific Microplate Reader (Thermo Fisher Scientific). Sephadex LH-20 (GE); RP-C₁₈ (Welch); silica gel (100–200 mesh and 200–300 mesh, Qingdao Marine Chemical Factory) and silica gel 60 (40–63 μ m, Merck, Germany) were used for column chromatography (CC). TLC was carried out on precoated silica

gel GF₂₅₄ (10–40 μ m, Qingdao Marine Chemical Factory) plates; Spots were visualized under UV light and by spraying with 10% H₂SO₄ in C₂H₅OH (v/v) followed by heating. The solvents used for extraction and isolation were analytical grade (Chengdu Kelong Chemical Reagent Factory, Chengdu, China). The ratios of solvent are described as a mixture by v/v. ABTS (2,2'-Azinobis (3-ethylbenzothiazoline-6-sulfonic acid), potassium persulfate (K₂S₂O₈) and chromatographic solvents (methanol and acetonitrile) for analytical HPLC were purchased from Sigma (St Louis, MO, USA). DPPH (1,1-Diphenyl-2-picryl-hydrazyl) free radical were purchased from TCI (St Louis, MO, USA). L-ascorbic acid ($\geq 98\%$) was purchased from Chengdu Kelong Chemical Reagent

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