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ACCEPTED MANUSCRIPT

Effects of water-borne copper and lead on metabolic and excretion rate of Bahaii loach (*Turcinoemacheilus bahaii*)

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Highlights

- Different lead treatments increased aerobic scope while all copper treatments showed no significant changes.
- The specific rate of ammonia excretion was declined following all lead or copper treatments as compared with control.
- Lead and copper can cause damages to fish gills and kidneys, hence, affecting the metabolic and excretory capacities of fish.

Abstract

Beyond the role of anthropogenic activities, natural sources of metal contaminations are still controversial, together counting, however, as a major threat to inland and coastal waters, becoming an even more prominent stressor for aquatic life. To address the effects of metals on the physiological response of fish, standard metabolic rate (SMR), maximum metabolic rate (MMR), aerobic scope (AS) and factorial aerobic scope (FAS) as well as specific rate of ammonia excretion (J_{amm}) of Turcinoemacheilus bahaii were determined following different water-borne Cu²⁺ and Pb²⁺ treatments. Following LC₅₀-96h determination, 72 fish (BW=1.153±0.56 g and TL=6.155±0.97 cm) were exposed to different amounts of Cu²⁺ and Pb²⁺ in 9 different treatments (eight fish/treatment), including $0.910 \text{ mg } l^{-1} \text{ Cu}^{2+} \text{ for } 24\text{h}, 0.455 \text{ mg } l^{-1} \text{ Cu}^{2+} \text{ for } 7d, 0.182 \text{ mg } l^{-1} \text{ Cu}^{2+} \text{ for } 14d \text{ and } l^{-1} \text{ cu}^{2+} \text{ cu}^{2+}$ $0.091 \text{ mg } 1^{-1} \text{ Cu}^{2+} \text{ for } 30 \text{ d as well as } 124.430 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 24\text{h}, 62.215 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^{2+} \text{ for } 7\text{d}, 12.443 \text{ mg } 1^{-1} \text{ Pb}^$ for 14d, 6.221 mg l⁻¹ Pb²⁺ for 30d and control. The SMR of fish was reduced following exposures to all Cu²⁺ and Pb2+ treatments (P<0.05), except for 30d exposure as compared with the control. The MMR remained steady following all Cu²⁺ treatments while it was raised significantly (P<0.05) following Pb²⁺ treatments at 7, 14 and 30d exposure. Although the AS showed a similar pattern to MMR, the FAS was elevated (P<0.05) following all the treatments when compared with control. Lower J_{amm} were observed following all metalstreated fish in comparison with control (P<0.05). In addition, higher (P<0.05) levels of injuries were observed following all Cu²⁺ and Pb²⁺ treatments in gills and kidneys. The results suggest that Cu²⁺ and Pb²⁺ over the experimental period could impair the metabolic and excretory capacities, hence affecting the possible physiological performance of fish.

Keywords

Metabolic Capacity, Aerobic Scope, Metabolic Waste, Gill and Kidney Histopathology, Metals

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