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Maximizing $CO₂$ biofixation and lipid productivity of oleaginous microalga Graesiella sp. WBG-1 via $CO₂$ -regulated pH in indoor and outdoor open reactors

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HIGHLIGHTS

rate under pH 8.0–9.0.

• The lipid production of WBG1 under 4 CO2-regulated pH levels was evaluated. • The culture has a maximum $CO₂$ fixation

• Highest biomass, lipid content and productivity were obtained under pH 8.0–

• There is a positive correlation between CO2 utilization efficiencies and pH

• Provide a strategy for coupling and maximizing the $CO₂$ fixation and lipid

GRAPHICAL ABSTRACT

 15% CO. Graesiella sp. WBG1 $CO₂$ -regulated pH Maximization of CO₂ biofixation and

biodiesel production simultaneously

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9.0.

values.

production.

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Carbon dioxide $(CO₂)$ and pH are two interdependent factors that greatly impact the growth and lipid accumulation of microalgae. However, the effects of these two factors are usually studied separately. The use of exogenous CO2, such as flue gas derived, to regulate pH in the large-scale cultivation of microalgae provides an ideal means for combining CO₂ biofixation and biodiesel production. In this study, the CO₂ biofixation and lipid production of oleaginous microalga Graesiella sp. WBG-1 was explored for four pH levels regulated by exogenous 15% CO₂ (flue gas concentration) in 10 L circular culture ponds and 5 m² open raceway reactors. Results revealed that pH 8.0–9.0 was the optimum pH for $CO₂$ fixation and lipid production, attaining the highest $CO₂$ fixation rates of 0.26 g L^{–1} day^{–1} and 18.9 g m^{–2} day^{–1}, respectively, lipid contents of 46.28% and 32.38%, and lipid pro-
ductivities of 64.8 mg L^{–1} day^{–1} and 3.14 g m^{–2} day^{–1}. A positive correlation between CO₂ and pH in open reactors was also suggested in this research, and thus provides direction for screening of $CO₂$ fixation by microalgae. The present study provides an excellent strategy for coupling CO₂ fixation and lipid production via microalgae in large-scale cultivation.

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1. Introduction

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The emissions of $CO₂$, a typical greenhouse gas, mainly derived from extensive fossil fuel consumption pose grave challenges to worldwide sustainability ([Karl et al., 2009](#page--1-0)). Of the various approaches for capturing CO2, biofixation by microalgae is considered one of the most promising

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technologies [\(Cheah et al., 2015; Kao et al., 2014\)](#page--1-0). Microalgae can convert $CO₂$ to massive amounts of biomass and biological products efficiently through photosynthesis. Many microalgal strains with different tolerance to $CO₂$ have been isolated and studied ([Cabello et al., 2017;](#page--1-0) [Kassim and Meng, 2017; Solovchenko and Khozin-Goldberg, 2013;](#page--1-0) [Zhao and Su, 2014](#page--1-0)); however, high cost and land demand greatly limit the biosequestration of $CO₂$ via microalgae [\(Lam et al., 2012](#page--1-0)). Another important application of these organisms, biodiesel derivation from microalgae, has been considered a potential renewable energy alternative to traditional fossil fuels for coping with energy shocks [\(Chisti,](#page--1-0) [2007; Hu et al., 2008; Mata et al., 2010\)](#page--1-0). Unfortunately, this application faces the same difficulties as $CO₂$ biofixation does. Thus, combining biodiesel production with $CO₂$ biosequestration of microalgae to reduce costs has become an urgent issue whether for the biodiesel industry or to reduce carbon emissions ([Kumar et al., 2010; Yen et al., 2015](#page--1-0)).

Though many species of microalgae have the ability to capture carbon under different concentrations of $CO₂$ owing to their unique carbon-concentrating mechanisms (CCMs) ([Barber and Price, 2003](#page--1-0)) and therefore have relatively high $CO₂$ biofixation rates compared to terrestrial plant [\(Chen et al., 2013](#page--1-0)), the low $CO₂$ concentration in air seriously limits the metabolic activity and biomass productivity of microalgae, especially in large-scale cultivation. The addition of exogenous $CO₂$ is not only beneficial to carbon biofixation, but also beneficial to biodiesel production in microalgae culture. It has been reported that the synthesis and accumulation of lipids and TAGs in microalgae could be triggered by stress conditions, such as nitrogen starvation and osmotic stress ([Brennan and Owende, 2010; Hu et al., 2008; Ördög et al.,](#page--1-0) [2013; Zhao and Su, 2014](#page--1-0)), and enhanced by suitable concentrations of inlet $CO₂$ [\(Breuer et al., 2013; Moheimani, 2013](#page--1-0)). Using the $CO₂$ in flue gas as a carbon source for large-scale cultivation of oleaginous microalgae to reduce biodiesel costs and attain $CO₂$ fixation has been considered for microalgae biodiesel production [\(Yen et al., 2015](#page--1-0)).

For carbon biofixation and biodiesel production via microalgae, the close relationship between the inorganic carbon source represented by CO₂ and the pH values of the medium is particularly noteworthy. The pH of microalgal culture systems fluctuates throughout the day/ night cycle because of the effects of photosynthesis and respiration. $CO₂$ consumption via photosynthesis raises pH values during the light period, and respiration at night lowers the pH. In the large-scale cultivation of oleaginous microalgae, the strong photosynthesis of microalgal cells can raise the pH to more than 10.0, which is obviously hazardous to the growth of most microalgal species ([Oswald, 1988](#page--1-0)). pH usually has been studied as an independent factor regarding its effects on biomass production and lipid accumulation of microalgae in previous studies [\(Bartley et al., 2014; Breuer et al., 2013; Moheimani, 2013](#page--1-0)). Sodium bicarbonate, other chemical buffers, and hydrochloric acid are typically used to regulate and maintain the desired pH levels in culture systems in most studies, and $CO₂$ is rarely used ([Bartley et al., 2014](#page--1-0)). However, the use of these chemicals to adjust pH is not only unsafe for the environment and does not easily control pH, but also further increases the biodiesel cost and is hence difficult to apply ([Wang et al., 2011\)](#page--1-0). Regulating and maintaining the pH of culture systems at an appropriate level for growth by injecting $CO₂$ is an optimum choice, whether considered from the perspective of cost or actual effect ([Bartley et al., 2014;](#page--1-0) [Han et al., 2013](#page--1-0)). Based on this, exogenous $CO₂$ and pH should be considered a combined factor in the combination of biodiesel production with $CO₂$ biosequestration via microalgae.

CO2-based pH regulation of cultivation systems to optimum levels seems a good method for coupling biodiesel production and $CO₂$ fixation by microalgae, and maximizing $CO₂$ fixation and oil production of microalgae simultaneously should be of high interest. In the present study, the effects of four pH levels (7.5–8.0, 8.0–9.0, 9.0–10.0, and 10.0–10.5) (the suitable pH range for algae growth under natural conditions) regulated by exogenous 15% CO₂ (average concentration in flue gases) on the $CO₂$ fixation rates, growth, lipid contents, fatty acid profiles, and lipid productivities of the oleaginous microalga Graesiella sp.

WBG-1 were explored. The aim of this study was to provide a valuable strategy for combining the goals of the algal biodiesel industry and that of reducing $CO₂$ emissions. The utilization efficiencies of injected $CO₂$ at different pH levels were also measured in these experiments, which was carried out in indoor circular culture ponds (10 L) and outdoor 5 m² raceway reactors (1000 L).

2. Materials and methods

2.1. Microalgae strains and medium

The oleaginous microalgal strain Graesiella sp. WBG-1, which exhibits excellent growth rate and lipid production and is suitable for large-scale cultivation, was used as experimental material. This strain was isolated from Chenghai Lake (N26°17′–26°28′, E100°38′– 100°41′), which contains half-alkaline waters (nearly 1% alkalinity), located in Yunnan Province, China, and was deposited in the Algae Culture Collection of Wuhan Botanical Garden, Chinese Academy of Sciences (Wuhan, China). Modified BG11 medium [\(Wen et al., 2014\)](#page--1-0) was used for the cultivation of Graesiella sp. WBG-1 both indoors and outdoors. The composition of this modified medium was similar to that in the literature except for the addition of 1 $g L^{-1}$ NaHCO₃ and a concentration of 0.10 g L^{-1} NaNO₃. All the chemicals used for medium preparation were analytically pure (Sinopharm Group Co. Ltd., Shanghai, China). The culture medium was sterilized through autoclaving at 121 °C for 30 min in the laboratory. The water from Chenghai Lake was used to prepare the medium for outdoor culture after filtration and chemical disinfection, as described in our previous study ([Wen et al., 2016](#page--1-0)).

2.2. Experimental photobioreactor and procedures

Self-designed circular culture ponds with culture volume of 10 L (indoor) and outdoor 5 m^2 open raceway reactors (1000 L) (Fig. 1) were used for the indoor and outdoor experiments, respectively. The 10 L reactors were manufactured from plexiglass and employed fluorescent lamps (30 W) to provide light. A water recycling system controlled by a thermostatic circulator (Hanon FCL6-20, China) was used to regulate the culture temperature. The steps and conditions of the indoor experiment are as follows. First, the WBG-1 cells were cultured to logarithmic

Fig. 1. Photobioreactors for microalgal cultivation. a: Scheme of 10 L circular culture pond; b: Photograph of 5 $m²$ open raceway reactors.

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