Accepted Manuscript

Oxidation mechanisms and chemical bioavailability of chromium in agricultural soil – pH as the master variable

Inka Reijonen, Helinä Hartikainen

PII: S0883-2927(16)30268-2

DOI: 10.1016/j.apgeochem.2016.08.017

Reference: AG 3711

To appear in: Applied Geochemistry

Received Date: 15 April 2016

Revised Date: 17 August 2016

Accepted Date: 29 August 2016

Please cite this article as: Reijonen, I., Hartikainen, H., Oxidation mechanisms and chemical bioavailability of chromium in agricultural soil – pH as the master variable, *Applied Geochemistry* (2016), doi: 10.1016/j.apgeochem.2016.08.017.

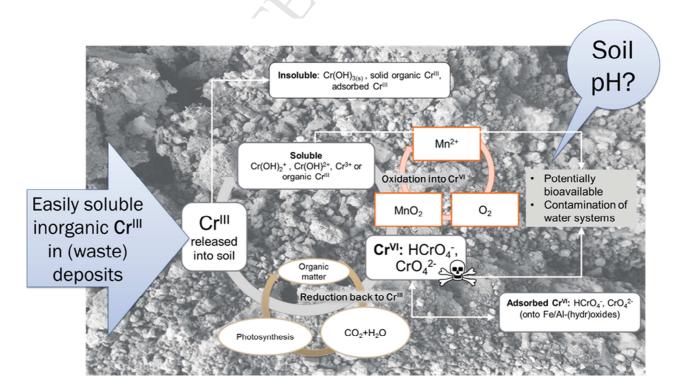
This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



ACCEPTED MANUSCRIPT

- Oxidation mechanisms and chemical bioavailability of
- 2 chromium in agricultural soil pH as the master
- 3 variable
- 4 KEYWORDS: chromium; oxidation; speciation; soil; bioavailability, chromate; pH; Cr; VI; III;
- 5 species
- 6 HIGHLIGHTS:
- Oxidation of Cr^{III} into Cr^{VI} by MnO₂ is regulated by soil pH
- Net-oxidation of Cr^{III} at low pH is diminished by the subsequent reduction of Cr^{VI}
- Oxidation of Cr^{III} is limited by its low solubility and precipitation onto MnO₂
- Potential bioavailability of Cr^{III} is enhanced by its oxidation into Cr^{VI}
- Risk of Cr^{III} oxidation was negligible in agricultural field soil samples without added MnO₂

12 GRAPHICAL ABSTRACT:



13

Download English Version:

https://daneshyari.com/en/article/8863288

Download Persian Version:

https://daneshyari.com/article/8863288

Daneshyari.com