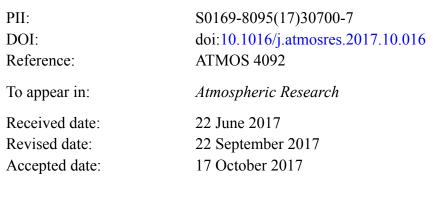
Accepted Manuscript

Ion balance and acidity of size-segregated particles during haze episodes in urban Beijing

Shili Tian, Yuepeng Pan, Yuesi Wang



Please cite this article as: Shili Tian, Yuepeng Pan, Yuesi Wang, Ion balance and acidity of size-segregated particles during haze episodes in urban Beijing. The address for the corresponding author was captured as affiliation for all authors. Please check if appropriate. Atmos(2017), doi:10.1016/j.atmosres.2017.10.016

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



ACCEPTED MANUSCRIPT

Ion balance and acidity of size-segregated particles during haze

episodes in urban Beijing

Shili Tian, Yuepeng Pan, Yuesi Wang

State Key Laboratory of Atmospheric Boundary Layer Physics and Atmospheric Chemistry (LAPC), Institute of Atmospheric Physics, Chinese Academy of Sciences, BJ 100029, China
*Corresponding author. Tel.: +86 01082020530; fax: +86 01062362389.
E-mail address: wys@mail.iap.ac.cn (Y. Wang); panyuepeng@mail.iap.ac.cn (Y. Pan).

Abstract: In this study, we investigated how the ion balance of size-segregated particles causes variations in aerosol acidity and atmospheric processing on clean versus hazy days using a 9-stage sampler. We calculated the ratios (in charge equivalents, R_{C/A}) between measured cations (Na⁺, NH₄⁺, K⁺, Mg²⁺, and Ca²⁺) and anions $(SO_4^{2-}, NO_3^{-} \text{ and } Cl^{-})$ for different aerosol size fractions. The ratios were typically close to unity in the accumulation mode, and increased significantly when the particle size increased or decreased. In the coarse size range (aerodynamic diameter > 2.1 μ m), high R_{C/A} values were most likely caused by the CO₃²⁻ and HCO_3^- content of the mineral dust. In contrast, the high $R_{C/A}$ values for submicron aerosols (< 1.1 μ m) were likely caused by the presence of water-soluble organic anions. The R_{C/A} values for all size fractions were lower on hazy days than clean days, indicating that aerosol acidity was enhanced on polluted days. The SO_4^{2-} and NO_3^{-} contents in fine particles were completely neutralized as the R_{C/A} values for PM_{2.1} approached unity, and mean values of R_{C/A} were 1.34 and 1.16 during the transition and polluted periods, respectively. The lowest $R_{C/A}$ values were observed in the size fraction with the highest concentrations of SO_4^{2-} , NO_3^{-} and NH_4^{+} (SNA) and concentrations of SNA increased with the increasing aerosol acidity. Significant correlations between $[NO_3^-]/[SO_4^{2-}]$ and $[NH_4^+]/[SO_4^{2-}]$ during NH_4^+ -rich conditions in fine size fractions indicated fine mode NO_3^- in Beijing was mainly formed by gas-phase homogeneous reaction between the ambient NH₃ and HNO₃.

Key words: Ion balance; acidity; size distribution; haze

1. Introduction

Atmospheric particulate matter (PM) is an important pollutant that has been linked to adverse health effects, visibility reduction and climate change (Liang et al., 2016; Download English Version:

https://daneshyari.com/en/article/8864841

Download Persian Version:

https://daneshyari.com/article/8864841

Daneshyari.com