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JOURNAL OF ENVIRONMENTAL SCIENCES XX (2017) XXX-XXX



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Impacts of sodium hydroxide and sodium hypochlorite aging on polyvinylidene fluoride membranes fabricated with different methods

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10 ARTICLEINFO

12 Article history:

- 13 Received 18 March 2017
- 14 Revised 19 July 2017
- 15 Accepted 20 July 2017
- 16 Available online xxxx
- 38 Keywords:
- 39 Micro/ultrafiltration membrane
- 40 Membrane aging
- 41 NIPS
- 42 TIPS
- 43 NaOH
- 44 NaClO
- 45

ABSTRACT

This study compared the effects of chemical aging on the polyvinylidene fluoride (PVDF) 17 membranes fabricated with the methods of non-solvent induced phase separation (NIPS) 18 (named NIPS-PVDF) and thermally induced phase separation (TIPS) (named TIPS-PVDF). The 19 chemical solutions of sodium hypochlorite (NaClO) and sodium hydroxide (NaOH) were 20 chosen at the concentration of 5000 mg/L. The equivalence of 5 and 10 years was respectively 21 selected as the time of aging. The physicochemical evolutions of membrane aging are 22 characterized on the base of morphology analysis, chemical components, permeation ability 23 and mechanical properties. The aging of NIPS-PVDF membrane led to the elimination of 24 surface hydrophilic additives, while NaOH focused on the dehydrofluorination process 25 resulting in the formation of conjugated chains of polyene on the skeleton structure. The 26 chemical components of the surface of TIPS-PVDF membrane were removed continuously 27 during the aging processes of both NaClO and NaOH, which was caused by the saponification 28 of surface additives and the chain scissions of skeleton structure, but without producing any 29 obvious conjugated chains of polyene. All the aging processes led to the increase of contact 30 angle and the decrease of mechanical properties, and the permeability was reduced first and 31 increased later due to the enlargement of surface membrane pores and membrane block. With 32 the influence of membrane aging, selectivity of membrane was decreased (except coliform 33 bacteria). At the beginning of filtration, the turbidity and particle count were at relatively high 34 levels and declined with the filtration process. 35

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50 Introduction

52 Micro/ultrafiltration (MF/UF) is known as a high effective 53 technology for drinking water treatment in removing 54 suspended particles, colloids, and microorganisms (Liden 55 et al., 2016; Bergamasco et al. 2011). At present, there is a growing interest in using polymer hollow fiber membranes in 56 the drinking water treatment for micro/ultrafiltration, which 57 can afford stable and good permeate water (Ravereau et al. 58 2016; Konieczny et al. 2009; Ma et al. 2014). Many kinds of 59 polymers such as the polyvinyl chloride, polyacrylonitrile, 60 aromatic polysulfones, and polyvinylidene fluoride are used 61

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http://dx.doi.org/10.1016/j.jes.2017.07.014

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Please cite this article as: Wu, Q., et al., Impacts of sodium hydroxide and sodium hypochlorite aging on polyvinylidene fluoride membranes fabricated with different methods, J. Environ. Sci. (2017), http://dx.doi.org/10.1016/j.jes.2017.07.014

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for membrane material (Regula et al. 2014; Zhang et al. 2016a). 62 More recently, polyvinylidene fluoride (PVDF), a semi-crystalline 63 64 polymer, has appeared as a favorable material for polymer membrane material due to its outstanding characteristics 65 in chemical resistance, mechanical strength, flexibility and 66 thermal stability (Jung et al. 2016; Zhao et al. 2016; Zhang et al. 67 2016b). Polyvinylidene fluoride membrane can be fabricated 68 mainly using several methods, including the methods of 69 70 electro-spinning, non-solvent induced phase separation (NIPS), 71 and thermally induced phase separation (TIPS) (Kim et al. 2016). 72 NIPS and TIPS are two common methods widely used in 73 fabricating various MF/UF membranes. Electro-spinning is 74 used to fabricate nanofiber membranes for membrane distillation (Jung et al. 2016; Xu et al. 2014). The membranes prepared 75 76 with the NIPS method typically exhibit an integrally skinned asymmetric structure when membranes prepared with the TIPS 77 method show a highly porous and symmetric structure. These 78 two membranes can be used for various applications because of 79 their different characteristics (Xiao et al. 2015). 80

One of the main problems associated with drinking water 81 micro/ultrafiltration treatment process is membrane fouling, 82 which is susceptible to the modification of membrane properties, 83 such as permeate flux, pore size distribution, and hydrophilicity, 05 85 and may influence the characteristics of membrane filtration (Hwang et al. 2015; Kimura et al. 2014; Yu et al. 2016). To restore 86 the membrane flux and properties, chemical cleaning is 87 88 routinely performed during the operation process of filtration 89 (Regula et al. 2013). The corrosive chemical agent, sodium ٩N hypochlorite (NaOCl) and sodium hydroxide (NaOH), are still 91 popular due to their high efficiency and acceptable price 92 (Ravereau et al. 2016).

However, it has been reported by some researchers that the 93 caustic environment would impact PVDF (Ravereau et al. 2016; 94 Rabuni et al. 2015; Zhang et al. 2017; Hashim et al. 2011). For 95 instance, the PVDF material became brittle in the extremely 96 caustic environment for the chemical attacking from NaOH 97 (Antón et al. 2015; Ross et al. 2000). It was explained as a 98 dehydrofluorination reaction, which removed hydrogen fluo-99 100 ride unit from polymer chain structure and forming carboncarbon double bonds (Ross et al. 2000; Zhang et al. 2006). But 101 most of the experiments were carried out under the very strong 102 103 caustic condition of 12 mol/L, high temperature of 80°C, or long exposure time of a few months (Rabuni et al. 2015; Zhang et al. 104 2017; Hashim et al. 2011). The reaction must be much more 105 reactive than the relatively low concentration at ambient 106 107 temperature (Hashim et al. 2011).

Limited literature is available concerning degradation of the 108 PVDF by hypochlorite (Ravereau et al. 2016; Regula et al. 2013; 109 Zhang et al. 2017; Puspitasari et al. 2010). Most of the researchers 110 focus on the altering of functional and mechanical properties, 111 the mechanisms still focus on the dehydrofluorination, and 112 some modifications of the functional properties have been 113 found (Zhang et al. 2017; Hashim et al. 2011). With the 114 decomposition of NaClO into Cl₂, ClO., and OH. free radicals, 115 116 macromolecule chain scission and crosslinking of PVDF membranes were observed under the exposure to NaClO 117 solution of the concentration of 4000 mg/L at 40°C, at three 118 distinct pH values: 6.0, 7.5 and 11.5, during about 5 months 119 (Ravereau et al. 2016; Regula et al. 2014). The temperature, 120 concentration and exposure time are far from the chemical 121

clean strength in the whole lifespan of most polymer mem- 122 branes (Ravereau et al. 2016; Hashim et al. 2011; Ross et al. 2000). 123 The removal or degradation of the surface hydrophilic additives 124 of PVDF membrane is often presented by spectroscopy analysis, 125 but the removal process is seldom observed, which would 126 modify surface morphology and chemical characteristics of 127 membrane (Robinson et al. 2016). 128

The effects of NaOH and NaClO aging on PVDF membranes 129 fabricated by various methods have been published by several 130 authors. For instance, Zhang et al. (2017) found that the effects 131 of NaClO aging on PVDF membrane fabricated with the NIPS 132 method would change the surface characteristics, permeability, 133 retention ability and fouling behavior of membrane. Hashim 134 et al. (2011) compared the effects of sodium hydroxide on PVDF 135 membranes made of three kinds of commercial raw material 136 (Solef 1015, Solef 6010, and Kynar 761) by the NIPS method. It 137 demonstrated the decreases of different levels in mechanical 138 integrity and crystallinity of PVDF membrane. Ravereau et al. 139 (2016) compared the effects of NaClO aging on two PVDF base 140 membranes fabricated by TIPS. One contained hydrophilic 141 additives, and the other was additive free. The chemical 142 modifications of PVDF membranes were observed, but the 143 changes were too low to induce membrane embrittlement and 144 porosity change. The polymer additives were completely 145 destroyed, which led to the decrease of hydrophilicity and the 146 improvement on membrane selectivity. The effects of chemical 147 aging on membranes fabricated with the electro-spinning 148 method, which was a nanofiber technology for membrane 149 distillation (MD), had not been studied in details. It may impact 150 the surface hydrophobic property, pore size and surface 151 porosity of the membrane (Liao et al. 2013). 152

Research mentioned above mainly focus on harsh condi- 153 tion, only one type of PVDF membrane, whilst many kinds of 154 PVDF membranes have been produced by manufacture, and 155 single cleaning agent of NaClO or NaOH (Regula et al. 2014; 156 Robinson et al. 2016). The comprehensive investigation is 157 seldom reported. Rabuni et al. (2015) assessed the impacts of 158 NaOH and NaClO cleaning on the stability of plate sheet PVDF 159 membrane, and found the increase of contact angle and the 160 decrease of protein retention, against with Ravereau et al. 161 (2016) research results of the improvement in virus selectivity 162 of membrane with chemical aging, which may be due to the 163 differences in membrane kind and aging condition. Some 164 researchers did not find water flux and polyethylene glycol 165 retention change significantly after the PVDF membrane 166 immersed in NaOH solution of 10 wt% for 1 month or NaOH 167 solution of 1 wt% for 24 months, due to the aging condition 168 not so harsh (Regula et al. 2014). 169

In the study, the effects of sodium hydroxide and sodium 170 hypochlorite solutions on two PVDF hollow fiber membranes 171 have been researched: one fabricated by the non-solvent 172 induced phase separation process and the other by the 173 thermally induced phase separation process. Since polymer 174 membranes have a designed lifespan, in order to gain a better 175 understanding of the relationship between aging extent and 176 usage history, an equivalence of lifespan in the aging condition 177 is introduced and varied. The changes of the chemical 178 components and permeate water quality (instead of bovine 179 serum albumin (BSA) or dextran) of membranes are monitored, 180 and the links between each other are established. 181

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