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Chemical characteristics and source apportionment of PM_{2.5} between heavily polluted days and other days in Zhengzhou, China

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40 Introduction

51 In recent years, China has experienced unprecedented eco-52 nomic developments along with a substantial quantity of 53 energy consumption, which results in serious air pollution 54 problems, especially extreme haze episodes (Wang et al., 55 2015). A haze comprising suspended solids, liquid particles, smoke, and vapor in the atmosphere which is defined as a 56 weather phenomenon with the horizontal visibility less than 57 10 km and the relative humidity (RH) less than 80% (Tan et al., 58 2009; Yang et al., 2015). Haze episodes have frequently 59 occurred in China along with the characteristics of large 60 polluted areas, long duration, and high concentration level 61 (Wang et al., 2014b). Recently, the occurrence of haze has 62

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ABSTRACT

PM_{2.5} samples were collected in Zhengzhou during 3 years of observation (2013–2015), and 17 Q7 chemical characteristics and source contribution were analyzed. Approximately 96% of the 18 daily PM2.5 concentrations and annual average values exceeded the Chinese National 19 Ambient Air Quality Daily and Annual Standards, indicating serious PM_{2.5} pollution. The 20 average concentration of water soluble inorganic ions was 2.4 times higher in heavily 21 polluted days (daily PM_{2.5} concentrations > 250 μ g/m³ and visibility < 3 km) than that in 22 other days, with sulfate, nitrate, and ammonium as major ions. According to the ratio of 23 NO_3^-/SO_4^{2-} , stationary sources are still the dominant source of $PM_{2.5}$ and vehicle emission 24could not be ignored. The ratio of secondary organic carbon to organic carbon indicated that 25 photochemical reactivity in heavily polluted days was more intense than in other days. 26 Crustal elements were the most abundant elements, accounting for more than 60% of 23 27 elements. Chemical mass balance results indicated that the contributions of major sources 28 (i.e., nitrate, sulfate, biomass, carbon and refractory material, coal combustion, soil dust, 29 vehicle, and industry) of PM2.5 were 13%, 16%, 12%, 2%, 14%, 8%, 7%, and 8% in heavily 30 polluted days and 20%, 18%, 9%, 2%, 27%, 14%, 15%, and 9% in other days, respectively. 31 Extensive combustion activities were the main sources of polycyclic aromatic hydrocarbons 32 during the episode (January 1-9, 2015) and the total benzo[a]pyrene equivalency 33 concentrations in heavily polluted days present significant health threat. Because of the 34 effect of regional transport, the pollution level of $PM_{2.5}$ in the study area was aggravated. 35 © 2017 The Research Center for Eco-Environmental Sciences, Chinese Academy of Sciences. Published by Elsevier B.V.

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rapidly increased in the Yangtze River Delta region and other 63 regions (Wang et al., 2014a). Five heavy pollution episodes, 64 whose frequencies were far greater than those in other years, 65 were recorded in January 2013 in northern China (Ji et al., 2014). 66 Particular matter (PM) played a major role in the formation 67 and evolution of haze (Tan et al., 2009). PM not only affects 68 climate, environment, and visibility but also has a severe 69 effect on human health (Pope et al., 2002). PM can be classified 70 71 by size into PM_{10} (aerodynamic diameter < 10 μ m) and $PM_{2.5}$ 72(aerodynamic diameter < 2.5 μ m). As PM_{2.5} is associated with more adverse health effects and influences on haze than 73 larger particles (Zhao et al., 2009), numerous analyses of the 74 haze mainly concentrate around chemistry compositions and 75 the source appointment of PM_{2.5} (Pateraki et al., 2012; F. Zhang 76 et al., 2013; Zhou et al., 2016). Sulfate, nitrate, and ammonium 77 (secondary inorganic aerosols, SIAs) were found to be the 78 major components of water soluble inorganic ions (WSIIs), 79 which accounted for one-third or more of PM_{2.5} (Kim et al., 80 2006; Cao et al., 2012; Kong et al., 2014). Organic carbon (OC) 81 and elemental carbon (EC) are also important constituents of 82 PM_{2.5}, particularly in highly industrialized and urbanized 83 areas (X.H.H. Huang et al., 2014). Elements were divided into 84 crustal elements and anthropogenic pollution elements, 85 86 including trace heavy metals harmful to the human body 87 (Gao et al., 2015). Although polycyclic aromatic hydrocarbons 88 (PAHs) constitute a small part of PM_{2.5}, they are well known to 89 be carcinogenic and mutagenic (Hu et al., 2007).

90 Zhengzhou, which is the capital of Henan province, is faced with severe air pollution accompanied by the rapid 91 development of the economy and a substantial number of 92people living demand, with a long-term coal-dominated 93 energy structure (http://tongji.cnki.net/kns55/Navi/YearBook. 94 aspx?id=N2016010114&floor=1###). According to satellite re-9596 mote sensing data, China is the most serious global PM polluted area. Moreover, most of the Henan Province experi-97 enced high aerosol optical depth problems, particularly 98 Zhengzhou (Tao et al., 2014), which is similar to the report 99 by Luo et al. (2014). According to the data from the Ministry of 08 Environmental Protection of the People's Republic of China 101 (MEPPRC, http://www.mep.gov.cn/), Zhengzhou is among the 102 10 Chinese cities with the worst air quality in 2013-2015 and 103 104 the primary pollutant was PM_{2.5} (http://www.zzepb.gov.cn/ 105Information/Content/?id=32853). Several studies have been reported on the chemical composition and source apportion-106ment of PM_{2.5} in Zhengzhou; however, only one year of data 107 was available in most of the literature. For example, F.H. Geng 09 et al. (2013) and N.B. Geng et al. (2013) determined that soil 109dust, secondary aerosol, coal combustion, biomass burning/ 110 oil combustion/incineration, vehicle emissions, and industrial 111 emissions contributed approximately 26%, 24%, 23%, 13%, 112 113 10%, and 4%, respectively, to $PM_{2.5}$ mass by positive matrix 114 factorization in Zhengzhou in 2010. A few studies on the long-term observation of PM2.5 in Zhengzhou, such as the 115WSIIs, carbonaceous components (J. Wang et al., 2016), and 116 117 PAHs (Wang et al., 2015) of PM_{2.5} were investigated in Zhengzhou from 2011 to 2013. However, the source contribu-118 tion percentages were obtained during normal days excluding 119 episodes. There was no report on source apportionment of 120 PM_{2.5} during heavily polluted days in Zhengzhou, as well as 121 122the research in long-term episode days.

In this study, 174 PM_{2.5} samples were collected and WSIIs, 123 OC, EC, and elements of the samples were detected. Chemistry 124 characteristics and potential source percentage contribution of 125 PM_{2.5} were contrasted between heavily polluted days and other 126 days during 2013–2015 (including a typical haze episode). 127

1. Materials and methods

1.1. Site description and sample collection

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Zhengzhou belongs to the warm temperate zone continental 131 climate, which has four distinctive seasons with an average 132 temperature of approximately 15°C (http://tongji.cnki.net/kns55/ 133 Navi/YearBook.aspx?id=N2014120113&floor=1). The sampling 134 site is located on the roof of the Collaborative Innovation Center 135 of Henan Resources and Materials Industry in Zhengzhou 136 University in Zhengzhou High-tech Zone (34°48' N, 113°31' E) 137 (Fig. 1). The sampling height is approximately 13 m above the 138 ground and there is no tall building around. 139

One hundred seventy-four samples were collected by using 140 quartz microfiber filters (20.3 cm \times 25.4 cm, Pall, USA) with a 141 high-volume PM_{2.5} sampler (TE-6070D Tisch Environmental, 142 USA), which was accompanied by the flow rate 1.13 m³/min 143 during the periods from 8 December 2012 to 24 October 2015. 144 Sampling was conducted from 9:00 am to 8:00 am of the 145 following day. 146

Before each sampling, the quartz filters were wrapped in 147 aluminum foil and baked at 450°C for 4.5 hr in a Muffle 148 furnace. All filters were then placed in a super clean room 149 (temperature: 25 ± 5 °C; RH: 50% ± 5%) for at least 48 hr and 150 weighed in a micro-balance (Mettler Toledo XS205, Switzer- 151 land) before and after each sampling. All filters were stored in 152 a freezer at -20°C before analysis. 153

1.2. Chemical analysis

1.2.1. WSIIs analysis

Two pieces of circular membranes (each piece 10.9 cm^2) were cut 156 from each sample filter and dipped in 25 mL ultra-pure water in 157 100 mL bottles. WSIIs from the samples were extracted by the 158 ultrasonic bath for 30 min (bath temperature < 30°C). The 159 extracted solutions were filtered with hydrophilic membranes 160 (0.22 µm) before determination. Four anion species (F^- , Cl^- , NO_3^- , 161 and SO_4^{2-}) were analyzed by using an ion chromatography 162 (ICS-900, Dionex, USA) with an IonPacASII-HC4 mm anion 163 separation column and an IonPacAGII-HC4 mm guard column. 164 The eluents were 8.0 mmol/L Na₂CO₃ and 1.0 mmol/L NaHCO₃ 165 mixtures with 0.8 mL/min flow rate. Five cation species (Na⁺, K⁺, 166 NH₄⁺, Ca²⁺, and Mg²⁺) were detected by ion chromatography 167 (ICS-90, Dionex, USA) with an IonPacCS12A cation separation 168 column and an IonPacCG12A guard column. The eluents were 169 20 mmol/L of methane sulfonic acid with 1.0 mL/min flow rate. 170

1.2.2. EC and OC analysis

Two 2.0 cm² membranes were punched from each filter and 172 used for analysis using a carbon analyzer (Sunset Laboratory, 173 USA). The analysis proceeds essentially in two stages. In the 174 first, OC is volatilized from the sample in a pure helium 175

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