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Particulate matter emissions and gaseous air toxic pollutants from commercial meat cooking operations

Nicholas Gysel, William A. Welch, Chia-Li Chen, Poornima Dixit, David R. Cocker III, Georgios Karavalakis*

5 Department of Chemical and Environmental Engineering, University of California, Riverside, CA 92521, USA 6 Center for Environmental Research and Technology, University of California, Riverside, CA 92507, USA

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ABSTRACT

This study assessed the effectiveness of three novel control technologies for particulate 16 matter (PM) and volatile organic compound (VOC) removal from commercial meat cooking 17 operations. All experiments were conducted using standardized procedures at University of 18 California, Riverside's commercial test cooking facility. PM mass emissions collected using 19 South Coast Air Quality Management District (SCAQMD) Method 5.1, as well as a dilution 20 tunnel-based PM method showed statistically significantly reductions for each control 21 technology when compared to baseline testing (i.e., without a catalyst). Overall, particle 22 number emissions decreased with the use of control technologies, with the exception 23 of control technology 2 (CT2), which is a grease removal technology based on boundary 24 layer momentum transfer (BLMT) theory. Particle size distributions were unimodal with 25 CT2 resulting in higher particle number populations at lower particle diameters. Organic 26 carbon was the dominant PM component (>99%) for all experiments. Formaldehyde and 27 acetaldehyde were the most abundant carbonyl compounds and showed reductions with 28 the application of the control technologies. Some reductions in mono-aromatic VOCs were 29 also observed with CT2 and the electrostatic precipitator (ESP) CT3 compared to the baseline 30 testing. 31

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46 Introduction

47 Commercial cooking has been shown to be an important 48 contributor to ambient particle emissions (with particulate 49 matter less than 2.5 μ m in size, PM_{2.5}) in urban environments and megacities (Allan et al., 2010; Schauer et al., 1999, 2002; 50Sun et al., 2011; Zhao et al., 2015). Emission inventory data 51showed that PM2.5 emissions from restaurant operations in 52the Los Angeles Basin contributed approximately 9.15 tons 53 per annual average day for 2014, with an estimate to exceed 5410 tons per annual average day for 2023 (AQMP, 2012). In the 55greater Los Angeles Basin, restaurant operations including 56

charbroilers (chain-driven and under-fired) are responsible for 57 about 84% of the $PM_{2.5}$ emissions from this source category 58 (AQMP, 2012). With an environmental problem of this magnitude, the South Coast Air Quality Management District 60 (SCAQMD) was forced to implement rules as part of the Air 61 Quality Management Plan for reducing 7 tons per day of PM_{10} 62 from charbroilers. At present time, SCAQMD evaluates rule 63 development efforts for restaurants including under-fired 64 charbroilers to install control devices with at least 85% reduction in $PM_{2.5}$ emissions. 66

Recently, there is an intense research activity within the 67 scientific community for the understanding of cooking organic 68

* Corresponding author. E-mail: gkaraval@cert.ucr.edu (Georgios Karavalakis).

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aerosol contribution to total organic aerosol in urban settings 69 due to the importance of airborne particulate emissions and 70 negative effects on human health (Mohr et al., 2012; Li et al., 71 2014; Robinson et al., 2006; Schauer et al., 1999). Numerous 72studies have found associations between particulate air 73 pollution with asthma exacerbations, increased respiratory 74 symptoms, decreased lung function, increased medication 75 use, and increased hospital admissions (BeruBe et al., 2007; 76 77 Kreyling et al., 2006; Utell and Frampton, 2000). Epidemiolog-78 ical studies have shown that exposure to particulate air pollution is associated with increased cardiovascular and 79 respiratory morbidity and mortality (Pope, 2000; Sioutas et al., 80 2005). Oberdorster et al. (2005) have shown that ultrafine 81 particles are more biologically active than larger particles due 82 to their greater surface area per mass. It was also found that the 83 small size facilitates uptake into cells and transcytosis across 84 epithelial cells into the blood circulation to reach potentially 85 sensitive areas, as well as penetrating the skin distribute via 86 87 uptake into lymphatic channels.

Commercial cooking can generate particulate emissions, 88 volatile organic compounds (VOCs), heterocyclic aromatic 89 amines, and polycyclic aromatic hydrocarbons with the 90 quantities of these pollutants strongly dependent on cooking 91 92procedures, such as cooking temperature, ingredients, dura-93 tion, and other factors (Lewtas, 2007; McDonald et al., 2003; 94 Nolte et al., 1999; Saito et al., 2014). Many studies have 95 evaluated the effects of different cooking styles on PM and 96 VOC emissions (Abdullahi et al., 2013; Cheng et al., 2016; He et al., 2004). Western cooking operations involve the con-97 sumption of beef and chicken, whereas Chinese cooking 98 mainly involves frying with pork, poultry, beef, seafood, and 99 vegetables. Zhao et al. (2007) showed a dominant presence 100 of β-sitosterol and levoglucosan in PM_{2.5} confirming that 101 vegetable oils are consumed during Chinese cooking opera-102 tions. Huang et al. (2011) reported a significant production 103 of formaldehyde, acetaldehyde, and benzene during residen-104tial cooking activities in Hong Kong. Mugica et al. (2001) 105reported the non-methane organic compounds, including 106 some monoaromatic hydrocarbons, of cooking emissions 107 from tortillerias, restaurants, rotisseries, and fried food places 108 in Mexico. They found that food cooking can be an important 109 110 source of these species. Schauer et al. (1999) showed that 111 formaldehyde and acetaldehyde were the predominant aldehydes from commercial charbroiling meat cooking opera-112tions. Buonanno et al. (2009) conducted a study to characterize 113particle emissions during grilling and frying and they found 114 higher emission factors at higher food temperatures, as well 115116as higher particle emissions as a function of the oil used. Rogge et al. (1991) reported increasing organic acids and 117 higher PM emissions for meats with higher fat contents. 118 119 McDonald et al. (2003) compared cooking methods and 120 identified under-fired charbroiling meat cooking emitted the 121 highest amount of PM_{2.5} per pound of meat cooked. They also found that charbroiling emissions were almost exclusively 122123 composed of organic carbon (OC) in nature with almost no elements or inorganic ions. Hildemann et al. (1991) estimated 124 125that approximately 21% of all organic PM_{2.5} in Los Angeles was 126 from meat cooking, while Schauer et al. (2002) estimated that 23% of the $PM_{2.5}$ organic carbon mass emitted in Los Angeles 127was contributed from meat cooking activities. 128

Although previous studies have provided substantial data 129 about indoor and outdoor cooking emissions, there is very 130 limited data on the effects of aftertreatment control technol-131 ogies on emissions from commercial cooking operations. In 132 California, and most of the United States, smaller restaurant 133 chains operating with under-fired charbroilers are not required 134 to control their PM emissions, which are an environmental 135 burden and also complicates the human risk assessment on 136 cooking emissions. Thus, it is necessary to study emissions 137 from under-fired charbroiled meat cooking operations with and 138 without aftertreatment control technologies. This work exam-139 ines the physical and chemical characteristics of PM_{2.5}, particle 140 number emissions and gaseous toxic pollutants from meat 141 cooking processes.

1. Materials and methods

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1.1. Test facility and protocol

The meat cooking experiments were conducted at the University 146 of California Riverside, Center for Environmental Research and 147 Technology (CE-CERT) commercial cooking facility. The facility 148 was equipped with a Nieco Model 9025 conveyorized charbroiler 149 fired with natural gas. Total emissions were captured by a 150 48-inch by 48-inch Captive-Aire stainless steel hood and ducted 151 to the second level of the facility with an upblast blower. The 152 blower had a variable speed drive and controller, which was used 153 to adjust the velocity and flow rates through the stack to meet 154 the Uniform Mechanical Code (UMC) and National Fire Protection Association (NFPA).

Prior to testing, the hamburger patties were prepared by 157 loading them onto sheet pans lined with freezer paper. The 158 1/3-pound meat patties used in this study were finished grind, 159 pure beef hamburger, 21% fat by weight, 58%–62% moisture, 160 3/8-inch-thick, and 5 in. in diameter. The fat and moisture 161 content of the patties were verified in accordance with recognized laboratory procedures (Association of Official Analytical 163 Chemists, AOAC, Official Actions 960.39 and 950.46, respectively). 164 Patties were cooked to an average internal temperature of 165 175 ± 5 °F, to confirm a medium-well condition. Internal meat 166 temperature was determined with a stack of hamburger patties 167 placed in a temperature measurement system. 168

Cooking cycles were developed in conjunction with the 169 California and National cooking restaurant associations and 170 private entities to best mimic commercial cooking processes 171 and were six minutes in duration. 172

1.2. Sampling and analysis 173

A sampling system (Fig. 1) was devised to simultaneously 174 collect multiple filter and gas samples. A sample was iso-175 kinetically withdrawn from the stack at a fixed flow rate and 176 diluted with VOC and particle-free air using a partial flow 177 venturi dilution system. The dilution system included quartz 178 filters (Q1–Q3), Teflon filters (T1–T4), equipped with orifices 179 to control flow rate through the filters and differential pres-180 sure (P1–P7) to measure filter loading. The total PM mass 181 was determined by gravimetric analysis of 47 mm (Teflo®, 182 Pall Gelman, USA) filters. The filters were conditioned and 183

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