

Synthesis of tri- and tetracyclic diterpenes. Cyclisations promoted by SmI₂

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Abstract—Several tri- and tetracyclic diterpenes have been synthesised from zamoranic acid. The key step is the cyclisation of a dicarbonyl 13,14-secoderivative by SmI₂.

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1. Introduction

Tri- and tetracyclic diterpenes are very common in nature, showing a variety of skeletons, and many of them are very important not only due to their structures but also from the biological point of view.¹ Abietanes, pimaranes, isopimaranes, cassanes, cleistanthanes, totaranes and the trinor-derivatives, podocarpanes, are tricyclic diterpenes. Among the tetracyclic systems are kauranes, beyeranes or hibaenes and isohibaenes (Fig. 1). They show biological profiles as

antifungal,² antiinflammatory,³ antioxidant,⁴ antibiotic,⁵ anti-tumoural,⁶ antimalarial,⁷ cardioactive,⁸ and antiviral⁹ compounds and are inhibitors of nitric oxide.¹⁰ Additionally some of this class of species are active as phytoalexins,¹¹ making these compounds particularly interesting. Due to their wide distribution and interesting bioactivities many synthetic studies have been reported.¹²

As part of our continuing efforts to synthesise this kind of biologically active compound, a series of tri- and tetracyclic diterpenes were synthesised from zamoranic acid. This compound is a labdane diterpene,¹³ which has already been used to prepare biologically active natural products, such as sesquiterpene drimanes,¹⁴ labdanolides, such as limonidilactone,¹⁵ diterpenes with the isofregenedane skeleton, such as chrysollic acid and isofregenedol.¹⁶ Recently we have reported the synthesis of tri- and tetracyclic diterpenes¹⁷ and (+)-totarol.¹⁸

In this paper we show that when the trinorderivative of zamoranic acid, the dialdehyde (14,15,16-trinorlabdan-13,17-dial), **23**, is treated with SmI₂, it gives rise to the podocarpanic diol **36** as the major product, an advanced intermediate for the synthesis of tricyclic compounds with the isopimarane, cassane, cleistanthane and rearranged abietane skeletons.

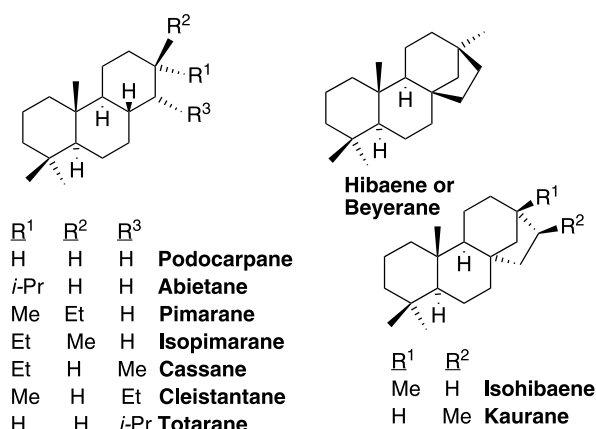


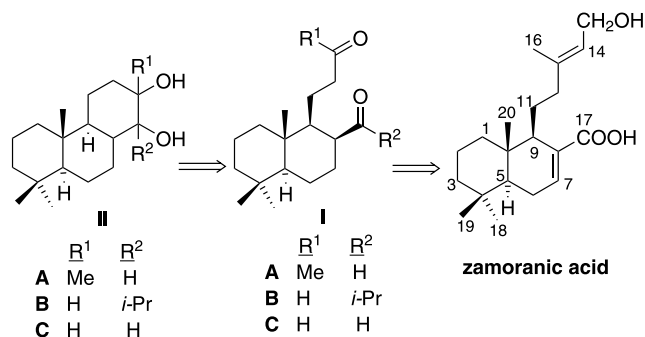
Figure 1. Diterpene tri- and tetracyclic skeletons.

Keywords: Samarium iodide; Stereoselective cyclisation; Tri- and tetracyclic diterpenes; Zamoranic acid.

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2. Results and discussion

Our general approach to the tricyclic diterpenes from zamoranic acid is outlined in Scheme 1. To allow access to tricyclic diterpenes of diverse skeletons such as **II**, our plan



Scheme 1.

leads to the corresponding dicarbonyl compounds **I** as intermediates for subsequent cyclisation by SmI₂. The synthesis of the dicarbonyl compounds **I** was planned as resulting from the degradation of the side chain of zamoranic acid and then adjustment of the functionalisation or substitution at C-17. The synthesis of tricyclic diterpenes, such as pimaranes, or tetracyclic species, such as hibaenes **IIA**, was planned via an intermediate obtained by cyclisation of **IA**. The synthesis of totaranes, and particularly (+)-totarol, can be envisaged from diol **IIIB** as intermediate, this being obtained by cyclisation of **IB** with SmI₂. As highlighted above in order to obtain tricyclic diterpenes with a range of different skeletons, the synthesis of the podocarpanic diol **IIC** was planned from the dialdehyde **IC**.

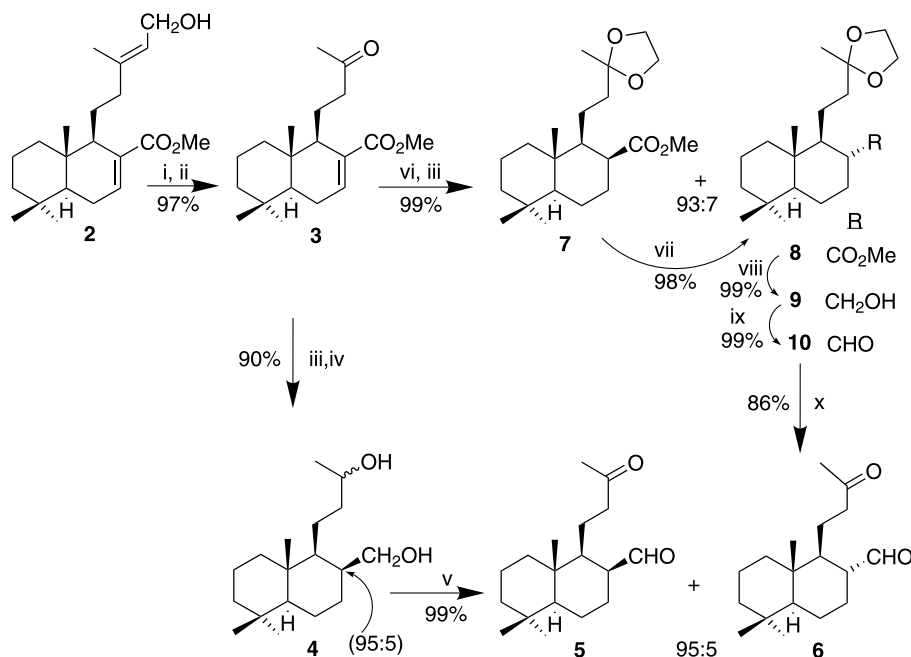
In the paper we describe the synthesis of compounds with pimarane and hibaene skeletons and of (+)-totarol, and demonstrate the potential of the podocarpanic diol **36** as an intermediate for the synthesis of different tricyclic diterpene skeletons. Initially, we describe the synthesis of the dicarbonyl compounds, then the cyclisation with SmI₂ of each of these, and finally the synthesis of the diterpenes.

2.1. Degradation of the side chain and synthesis of dicarbonyl compounds

2.1.1. Synthesis of dicarbonyl compounds 5 and 6. The synthesis of the intermediates **5** and **6** is shown in Scheme 2. Chemoselective oxidation of the allyl alcohol olefin unit of zamoranic acid **2** with OsO₄¹⁹ afforded the corresponding diol, which was subsequently cleaved with LTA (lead tetracetate)²⁰ to yield the methyl ketone **3**. Hydrogenation of **3** in the presence of PtO₂,²¹ followed by reduction with DIBAL-H²² led to compound **4** with very good diastereoselectivity, we realised that LAH could also be used for the latter part of this transformation. Oxidation of **4** using Swern technique²³ gave **5** and **6**; Collins's reagent²⁴ could also facilitate this transformation.

However, aldehyde **6** was alternatively synthesised from **7** in an excellent yield (Scheme 2). Hydrogenation of the dioxolane derivative of **3** gives **7** and **8** stereoselectively (93:7, 98%). The coupling constants of H-8 in the ¹H NMR spectra for **7** (*J*=1.6, 4.8 Hz) and **8** (*J*=4.3, 11.8 Hz) confirm the stereochemical assignment for C-8. Treatment of **7** with NaOMe²⁵ gave diastereomer **8**, which was easily converted to **6** by a reduction–oxidation sequence through the intermediates **9** and **10**, these transformations confirm the C-8 stereochemistry of **5** and **6**.

2.1.2. Synthesis of dicarbonylic compound 19. Addition of isopropylmagnesium chloride²⁶ to aldehyde **10** (Scheme 3) proceeded with complete stereoselectivity to give exclusively the 14*R*-hydroxy derivative **11**, in excellent yield. The structure of **11** was confirmed by X-ray analysis of one of its derivatives (compound **15**, Scheme 3), showing that the reaction of the nucleophile has taken place from the less hindered *Si* face of the carbonyl group.



Scheme 2. (i) OsO₄, NMO, ^tBuOH/THF/H₂O, rt, 18 h; (ii) LTA, benzene, rt, 2 h; (iii) H₂/PtO₂/Et₂O, rt, 36 h for **3** to **4** and 10 h for **3** to **7** and **8**; (iv) DIBAL-H, toluene, –78 °C, 30 min; (v) (COCl)₂, DMSO, Et₃N, DCM, –78 °C, 30 min; (vi) (CH₂OH)₂, *p*-TsOH, benzene, reflux, 21 h; (vii) (a) NaOMe/MeOH, reflux 12 h; (b) CH₂N₂; (viii) LAH, Et₂O, rt, 1 h; (ix) CrO₃/Py, DCM, rt, 30 min; (x) *p*-TsOH/acetone, rt, 14 h.

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