Contents lists available at ScienceDirect

### Physica A

journal homepage: www.elsevier.com/locate/physa

# Canonical ensemble in non-extensive statistical mechanics, q > 1



Institute of Theoretical Physics and Astronomy, Vilnius University, A. Goštauto 12, LT-01108 Vilnius, Lithuania

#### HIGHLIGHTS

- Reservoir having large generalized heat capacity leads to the Tsallis statistics.
- Short-range interactions with such a reservoir lead to *q*-exponential factor.
- Generalized heat capacity with q > 1 leads to a negative physical heat capacity.
- The condition of applicability of canonical ensemble is the same for all values of q.

#### ARTICLE INFO

Article history: Received 23 February 2016 Available online 19 April 2016

Keywords: Generalized statistical mechanics Tsallis entropy Canonical ensemble Nonextensivity

#### ABSTRACT

The non-extensive statistical mechanics has been used to describe a variety of complex systems. The maximization of entropy, often used to introduce the non-extensive statistical mechanics, is a formal procedure and does not easily lead to physical insight. In this article we investigate the canonical ensemble in the non-extensive statistical mechanics by considering a small system interacting with a large reservoir via short-range forces and assuming equal probabilities for all available microstates. We concentrate on the situation when the reservoir is characterized by generalized entropy with non-extensive statistical mechanics occurring when q > 1 and show that there is a limit on the growth of the number of microstates of the system that is given by the same expression for all values of q. © 2016 Elsevier B.V. All rights reserved.

#### 1. Introduction

The standard, Boltzmann–Gibbs statistical mechanics has been successfully applied to describe a huge variety of systems. The cornerstone of the standard statistical mechanics is the functional form of the entropy

$$S_{\rm BG} = -k_{\rm B} \sum_{\mu} p(\mu) \ln p(\mu),$$
(1)

where  $p(\mu)$  is the probability of finding the system in the state characterized by the parameters  $\mu$ . However, there are systems exhibiting long-range interactions, long-range memory, and anomalous diffusion, that possess anomalous properties in view of traditional Boltzmann–Gibbs statistical mechanics. To understand such systems a generalization of statistical mechanics has been proposed by Tsallis [1]. The non-extensive statistical mechanics has been used to describe phenomena in many physical systems: dusty plasmas [2], trapped ions [3], spin-glasses [4], anomalous diffusion [5,6], high-energy physics [7], Langevin dynamics with fluctuating temperature [8,9], cold atoms in optical lattices [10], turbulent

http://dx.doi.org/10.1016/j.physa.2016.04.020 0378-4371/© 2016 Elsevier B.V. All rights reserved.





PHYSICA

E-mail address: julius.ruseckas@tfai.vu.lt.

flows [11]. This generalized framework has found applications also in chemistry, biology, geology, and economics [12-15]. Instead of Eq. (1) the non-extensive statistical mechanics is based on the generalized functional form of the entropy [1]

$$S_q = k_{\rm B} \frac{1 - \sum_{\mu} p(\mu)^q}{q - 1}.$$
(2)

Here the parameter q describes the non-extensiveness of the system. The Boltzmann–Gibbs entropy can be obtained from Eq. (2) in the limit  $q \rightarrow 1$  [1,16]. More generalized entropies and distribution functions are introduced in Refs. [17,18]. It is convenient to write the equations of non-extensive statistical mechanics using the q-logarithm

$$\ln_q x = \frac{x^{1-q} - 1}{1-q}$$
(3)

and its inverse, the *q*-exponential [1]

$$\exp_q(x) \equiv [1 + (1 - q)x]_+^{\frac{1}{1 - q}}.$$
(4)

Here  $[x]_+ = x$  if x > 0, and  $[x]_+ = 0$  otherwise. For example, using the *q*-logarithm one can write Eq. (2) in a form similar to the Boltzmann–Gibbs entropy (1) [1]:

$$S_q = k_{\rm B} \sum_{\mu} p(\mu) \ln_q \frac{1}{p(\mu)}.$$
(5)

The exponential Boltzmann factor in the non-extensive statistical mechanics is replaced by a *q*-exponential. In the limit  $q \rightarrow 1$  the *q*-logarithm becomes an ordinary logarithm and the *q*-exponential function becomes the ordinary exponential  $e^x$ .

In the non-extensive statistical mechanics the canonical ensemble is often described in a formal way, starting from the maximization of the generalized entropy (2) [1]. The physical content enters as a form of constraints in the maximization procedure. In Ref. [19] the canonical ensemble in the non-extensive statistical mechanics has been considered starting from a physical situation of a small system interacting with a large reservoir via short-range forces. Assuming that the *q*-heat capacity of the reservoir instead of the ordinary heat capacity is large, the equations of the non-extensive statistical mechanics have been obtained. However, in Ref. [19] only the case of q < 1 has been investigated. The goal of this paper is to consider the situation when q > 1.

As in Ref. [19] we are investigating a small system interacting with a large reservoir via short-range forces. Such description is not directly applicable to a subsystem of a large system with long-range interactions, where the non-extensive statistical mechanics has been usually applied. However, the much simpler situation of short-range forces allows us to highlight the differences from the standard statistical mechanics and to gain a deeper insight into non-extensive statistics.

The case of q > 1 presents additional difficulties compared with the situation when q < 1. For example, let us consider the microcanonical ensemble where the probability of a microstate  $\mu$  is  $p(\mu) = 1/W$ , with W being the number of microstates. If the generalized entropy  $S_q$  is extensive and proportional to the number of particles N in the system, the number of microstates W behaves as  $(1 - (q - 1)AN)^{-1/(q-1)}$ . Thus the number of microstates becomes infinite when the number of particles N approaches a finite maximum number  $N_{\text{crit}}$  and the macroscopic limit  $N \to \infty$  cannot be taken. In this situation one can try to take a different limit,  $N \to N_{\text{crit}}$ , instead of the limit  $N \to \infty$ . Additional problem is that the q-exponential distributions with q > 1 lead to divergences in the thermodynamic limit for classical Hamiltonian systems [20].

The paper is organized as follows: In Section 2 we consider the canonical ensemble in the non-extensive statistical mechanics describing a small system interacting with a large reservoir via short-range forces. In Section 3 we investigate possible divergences arising in the description of the system using canonical ensemble. We analyze properties of the generalized thermodynamical quantities for the case of q > 1 in Section 5 summarizes our findings.

#### 2. Canonical ensemble in non-extensive statistical mechanics when q > 1

As in Ref. [19] we will consider a composite system consisting of a small system S interacting with a large reservoir R. We assume that the interaction between the system S and the reservoir R is via short-range forces, however the reservoir R is not described by the Boltzmann–Gibbs statistics. We require that the *q*-heat capacity  $C_q^{(R)}$  of the reservoir, defined by Eq. (14), instead of standard heat capacity should be large. In this article we consider only the situation when q > 1.

Similar investigation of the canonical ensemble in the non-extensive statistical mechanics has been performed in Ref. [21], however the reservoir has been considered as a heat bath. A system weakly coupled to a finite reservoir has been considered in Ref. [22]. Assuming that the number of microstates of the reservoir with energy less than  $E_R$  grows as a power-law of  $E_R$ , the *q*-exponential distribution of the energy of the system has been obtained. In Ref. [22] the parameter *q* tends to 1 when the number of particles of the reservoir increases. Here we do not assume any particular dependence of the parameter *q* on the number of particles in the reservoir.

Download English Version:

## https://daneshyari.com/en/article/974339

Download Persian Version:

https://daneshyari.com/article/974339

Daneshyari.com