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Physica A

A conservative multi-group approach to the Boltzmann equations for reactive gas mixtures

A B S T R A C T

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h i g h l i g h t s

- We consider a quaternary gas mixture undergoing a bimolecular chemical reaction.
- We discuss a reactive kinetic model based on Boltzmann like equations.
- We resort to a probabilistic formulation in terms of collision frequencies and kernels.
- We propose a multi-group approach (speed discretization) with suitable weights.
- We prove that the scheme is conservative at any approximation step.

a r t i c l e i n f o

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1. Introduction

Modelling and analysis of gas mixtures constitute very stimulating and challenging problems of practical significance, especially if one wants to account for non-conservative effects like chemical reactions. Indeed we are surrounded by gas mixtures in everyday life. They also play an important role in science and technology. Increasingly better tools for their $\frac{4}{3}$ analysis and numerical simulation are required in order to control and predict their behaviour in different environments and ⁵ physico-chemical situations. On the other hand gas flows with chemical reactions play a crucial role in several applications of ϵ paramount importance, like combustion processes, detonation and deflagration waves, chemical reactors, power generation, ⁷ and space vehicles.

The matter can be handled in the frame of the continuum theory of fluids $[1,2]$ $[1,2]$, but of course a kinetic approach is highly $\overline{9}$ desirable, not only for a consistent derivation of the relevant fluid-dynamic equations, but also for a better description and 10

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the particle distribution functions of all species. The procedure takes advantage of a suitable probabilistic formulation, based on the underlying collision frequencies and transition probabilities, of the relevant reactive kinetic equations of Boltzmann type. Owing to an appropriate choice of a sufficiently large number of weight functions, it is shown that the proposed multi-group equations are able to fulfil exactly, at any order of approximation, the correct conservation laws that must be inherited from the original kinetic equations, where speed was a continuous variable. Future developments are also discussed. © 2015 Published by Elsevier B.V.

Starting from a simple kinetic model for a quaternary mixture of gases undergoing a bimolecular chemical reaction, multi-group integro-differential equations are derived for

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 1 a deeper understanding of the underlying problems $[3-5]$. Work on reactive kinetic equations is in progress, starting from 2 suitable models for chemical reactions and state transitions $[6-10]$ for different types of simple reactions.

³ The problem that will be addressed here concerns the reversible bimolecular reaction

$$
A_1 + A_2 \rightleftarrows A_3 + A_4,\tag{1}
$$

 5 starting for simplicity from the simplified reactive kinetic model [\[6\]](#page--1-3), the so-called Rossani–Spiga model [\[11\]](#page--1-4), in which ⁶ non-translational degrees of freedom are neglected. The set of integro-differential Boltzmann-like equations governing the distribution functions *fⁱ* making up the distribution vector *f* ⁷ of the considered mixture reads as

$$
\frac{\partial f_i}{\partial t} + \mathbf{v} \cdot \nabla_{\mathbf{x}} f_i = \sum_{j=1}^4 Q_{ij} [f_i, f_j] + J_i [f_j] \equiv Q_i [f_j] \quad i = 1, ..., 4
$$
\n(2)

where *Qij* is the collision integral for mechanical encounters of the (*i*, *j*) pair, and *Jⁱ* ⁹ is the net gain by collision for species *i* in ¹⁰ the chemical reaction (*i*, *j*, *h*, *k*), with admissible sequences (1, 2, 3, 4), (2, 1, 4, 3), (3, 4, 1, 2), (4, 3, 2, 1). They are given $_1$, by five fold integrals involving the differential cross sections for pair elastic scattering σ_{ij}^E , and the reactive cross sections *i*₂ for the chemical reactions σ_{ij}^R (with $\sigma_{ji}^*=\sigma_{ij}^*$, a star standing for either *E* or *R*). Each species is endowed with an energy of 13 chemical bond E_i , and the heat of reaction is given by

$$
\Delta E = -\sum_{i=1}^{4} \Lambda_i E_i \qquad \mathcal{A} = (1, 1, -1, -1) \tag{3}
$$

 15 in terms of the stoichiometric coefficients Λ_i . Being only matter of convention, we may always assume Δ $E > 0$, so that ¹⁶ an energetic threshold occurs for the direct reaction in [\(1\).](#page-1-0) Possible activation energies are included in the cross sections, 17 which account for both possible transitions $(i, j) \rightarrow (h, k)$ and $(i, j) \rightarrow (k, h)$. We shall set $M = m_1 + m_2 = m_3 + m_4$ for n_8 the total mass, $r_i=m_i/M$ for mass fractions, $\mu_{ij}=m_im_j/M=r_ir_jM$ for reduced masses, and $\delta_{ij}=\sqrt{2\Delta E/\mu_{ij}}>0.$

¹⁹ The general *i*th reactive collision integral is given by

$$
J_{i}[\underline{f}](\mathbf{v}) = \iint U(g - \Lambda_{i}\delta_{ij})g\sigma_{ij}^{R}(g, \hat{\mathbf{n}} \cdot \hat{\mathbf{n}}') \left[\left(\frac{\mu_{ij}}{\mu_{hk}} \right)^{3} f_{h} \left(\mathbf{v}_{ij}^{hk} \right) f_{k} \left(\mathbf{w}_{ij}^{hk} \right) - f_{i}(\mathbf{v})f_{j}(\mathbf{w}) \right] d_{3}\mathbf{w} d_{2}\hat{\mathbf{n}}' \tag{4}
$$

where $g=|\textbf{v}-\textbf{w}|$, $\hat{\textbf{n}}=(\textbf{v}-\textbf{w})/g$, and $\textbf{v}^{hk}_{ij}=r_i\textbf{v}+r_j\textbf{w}+r_k$ $g^{hk}_{ij}\hat{\textbf{n}}'$, $\textbf{w}^{hk}_{ij}=r_i\textbf{v}+r_j\textbf{w}-r_h$ $g^{hk}_{ij}\hat{\textbf{n}}'$, with $g^{hk}_{ij}=\left[\frac{\mu_{ij}}{\mu_{hl}}\right]$ where $g=|\textbf{v}-\textbf{w}|$, $\hat{\textbf{n}}=(\textbf{v}-\textbf{w})/g$, and $\textbf{v}_{ij}^{hk}=r_i\textbf{v}+r_j\textbf{w}+r_k g_{ij}^{hk}\hat{\textbf{n}}'$, $\textbf{w}_{ij}^{hk}=r_i\textbf{v}+r_j\textbf{w}-r_h g_{ij}^{hk}\hat{\textbf{n}}'$, with $g_{ij}^{hk}=\left[\frac{\mu_{ij}}{\mu_{hk}}(g^2-\varLambda_i\delta_{ij}^2)\right]^{1/2}$. $_{22}$ Here *U* denotes the unit step function, and it actually introduces a threshold for the collision when $\bar{A_i}>0.$ In the ranges $_{\rm 23}$ allowed for reaction, cross sections σ_{ij}^{hk} are related by the micro-reversibility condition

$$
\mu_{ij}^2 g^2 \sigma_{ij}^R(g, \hat{\mathbf{n}} \cdot \hat{\mathbf{n}}') = \mu_{hk}^2 \left(g_{ij}^{hk} \right)^2 \sigma_{hk}^R (g_{ij}^{hk}, \hat{\mathbf{n}} \cdot \hat{\mathbf{n}}'). \tag{5}
$$

25 The elastic collision term Q_{ij} may be considered as a specialization of [\(4\)](#page-1-1) to the sequence (*i*, *j*, *i*, *j*), by using the elastic rather μ_{26} than the reactive cross section. In this case thus $\Delta E=0$, $\mu_{hk}=\mu_{ij}$, $\delta_{ij}=0$, and $g^{hk}_{ij}=g.$

 $_{27}$ Collision invariants, namely continuous test functions $\phi(\mathbf{v})$ satisfying, for any *f*

$$
\sum_{i=1}^{4} \int \phi_i(\mathbf{v}) Q_i(\mathbf{v}) d_3 \mathbf{v} = 0,
$$
\n(6)

²⁹ constitute a seven-dimensional linear space, spanned by the independent scalar vectors

30
$$
\begin{pmatrix} 1 \\ 0 \\ 1 \\ 0 \end{pmatrix}
$$
, $\begin{pmatrix} 1 \\ 0 \\ 0 \\ 1 \end{pmatrix}$, $\begin{pmatrix} 0 \\ 0 \\ 1 \\ 1 \end{pmatrix}$, $\begin{pmatrix} m_1 \mathbf{v} \\ m_2 \mathbf{v} \\ m_3 \mathbf{v} \\ m_4 \mathbf{v} \end{pmatrix}$, $\begin{pmatrix} \frac{1}{2} m_1 v^2 + E_1 \\ \frac{1}{2} m_2 v^2 + E_2 \\ \frac{1}{2} m_3 v^2 + E_3 \\ \frac{1}{2} m_4 v^2 + E_4 \end{pmatrix}$ (7)

31 Particle conservation is given by the sum of the first and the third one, mass conservation is recovered by linear combination ³² of the first three multiplied by *m*3, *m*1−*m*3, *m*2, respectively. The following three vectors represent momentum conservation,

- 33 and the last one gives conservation of total (kinetic plus reactive) energy.
- ³⁴ A Boltzmann lemma and an *H*-theorem hold [\[6,](#page--1-3)[7\]](#page--1-5). Collision equilibria, defined by

$$
Q_i[f](\mathbf{v}) = 0, \quad \forall \mathbf{v}, \forall i \tag{8}
$$

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