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# Investigations on preparation, upper critical field and low temperature thermal expansion of LiTi<sub>2</sub>O<sub>4</sub> superconductor

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### Abstract

A new preparation route for LiTi<sub>2</sub>O<sub>4</sub> superconductor was presented. The upper critical field and low temperature thermal expansion of LiTi<sub>2</sub>O<sub>4</sub> were investigated. The upper critical field  $H_{C2}(0)$  determined by the measurements of resistance and magnetization versus temperature are  $8.9 \pm 0.5$  T and  $9.3 \pm 0.3$  T, respectively. The linear coefficient of thermal expansion in the temperature range of 90–298 K determined by low temperature X-ray diffraction is  $7.97(\pm 0.13) \times 10^{-5}$  K<sup>-1</sup>.

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### 1. Introduction

LiTi<sub>2</sub>O<sub>4</sub> superconductor with the spinel structure and  $T_{\rm C} \sim 12$  K was discovered by Johnston in 1973 [1]. This superconductor is still interesting due to some unusual properties in both physical and chemical respects. The preparation and crystal structure of the superconductor have been investigated in Refs. [2–12]. In previous publications, the polycrystalline LiTi<sub>2</sub>O<sub>4</sub> superconductor was prepared by solid state reaction method using the starting materials of Li<sub>2</sub>Ti<sub>2</sub>O<sub>5</sub> + Ti<sub>2</sub>O<sub>3</sub> + TiO<sub>2</sub>, or Li<sub>2</sub>CO<sub>3</sub> + TiO<sub>2</sub> and Li<sub>2</sub>CO<sub>3</sub> + TiO<sub>2</sub> + Ti, with a more complex procedure. The upper critical field  $H_{C2}(0)$  (2–32.8 T) of LiTi<sub>2</sub>O<sub>4</sub> was reported by several groups [3,5,8]. Recently, the  $H_{C2}(0)$  of LiTi<sub>2</sub>O<sub>4</sub> determined by specific heat measurement is 11.7 T in Ref. [13].

In this paper, a new preparation route for  $\text{LiTi}_2\text{O}_4$  superconductor was presented. The upper critical field and low temperature thermal expansion of  $\text{LiTi}_2\text{O}_4$  superconductor were investigated.

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# 2. Experimental

Ceramic LiTi<sub>2</sub>O<sub>4</sub> samples were prepared by a new preparation route. The chemical reagent TiO and Li<sub>2</sub>Ti<sub>3</sub>O<sub>7</sub> monocrystal powders with the purities better than 99.99% were used as the starting materials. The preparation method of TiO was reported (see the powder diffraction file (23-1078)). The stoichiometric powders were mixed, ground, pelletized and calcined in an evacuated quartz tube at 805 °C for 50 h, then cooled slowly to room temperature, according to the chemical reaction:

# $2Li_2Ti_3O_7 + 2TiO = 4LiTi_2O_4$

The phase structure of the prepared samples were characterized by powder X-ray diffraction (XRD) analysis on an MXP18A-HF-type diffractometer with Cu-K<sub> $\alpha$ </sub> radiation. Powder X program was used for lattice parameter calculations. DCmagnetization *M* and electrical resistance *R* were measured versus temperature *T* using a DC-SQUID magnetometer and four-probe technique, respectively. Low temperature linear thermal expansion was investigated by powder X-ray diffraction at low temperature on Rigaku D/ max-2500 diffractometer with Cu-K<sub> $\alpha$ </sub> radiation (40 kV × 200 mA) and a graphite monochromator.

#### 3. Results and discussion

## 3.1. Sample characterization

The used Li<sub>2</sub>Ti<sub>3</sub>O<sub>7</sub> monocrystal was grown by Czochralski technique in our institute. Li<sub>2</sub>CO<sub>3</sub> and TiO<sub>2</sub> with purities better than 99.99% were used as the starting materials. This crystal has lightly yellow color and is a transparent body. The XRD pattern of the monocrystal powder is shown in Fig. 1. Its lattice parameter  $a = 5.015 \pm$ 0.004 Å,  $b = 9.539 \pm 0.004$  Å,  $c = 2.946 \pm 0.005$ Å, which is agreement with those of a =5.0182 Å, b = 9.5523 Å, c = 2.9455 Å reported in the powder diffraction file (34-393), indicating that the crystal has good quality.

The composition (Li 40 mol% and Ti 60 mol%) of  $Li_2Ti_3O_7$  is the closest to that (Li 33.3 mol% and Ti 66.7 mol%) of  $LiTi_2O_4$  among the compounds



Fig. 1. XRD pattern of Li<sub>2</sub>Ti<sub>3</sub>O<sub>7</sub> monocrystal powder.

in  $Li_2O-TiO_2$  system. This preparation route is beneficial to repress the volatilization of Li component, to control the composition of  $LiTi_2O_4$  superconductor and to simplify preparing processes.

Fig. 2 shows the XRD pattern of the prepared LiTi<sub>2</sub>O<sub>4</sub> superconductor, indicating that it is single phase. Its lattice parameter  $a = 8.400 \pm 0.002$  Å, which is good agreement with that (8.405 Å) of LiTi<sub>2</sub>O<sub>4</sub> with stoichiometric composition [5]. R-T and M-T curves of the prepared LiTi<sub>2</sub>O<sub>4</sub> are shown in Figs. 3 and 4, indicating that  $T_{\rm C}$  is 12.1 K,  $\Delta T = 0.3$  K and superconducting transition is very sharp.



Fig. 2. XRD pattern of LiTi<sub>2</sub>O<sub>4</sub>.

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