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## Effects of ZrH<sub>2</sub> doping and sintering temperature on the critical current density of MgB<sub>2</sub> wires

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#### Abstract

 $ZrH_2$ -doped MgB<sub>2</sub> wires, with the composition of Mg: $ZrH_2$ :B = (1-x):x:2 (x = 0%, 5%, 8%, 10%, 12%, 15%), were fabricated through the in situ powder-in-tube method by using low carbon steel tubes as sheath materials. Samples were sintered at 700 °C, 750 °C, 800 °C and 900 °C for 2 h respectively in a flow of high purity argon. It is found that the amount of  $ZrB_2$  increases with the doping ratio, and the content of  $Fe_2B$  increases with the sintering temperature. Critical current density ( $J_c$ ) of all the doped samples decreases with the increase of sintering temperature, but the optimum doping ratio goes up correspondingly. Among all the doped samples, the samples of x = 5% and x = 8% exhibit better  $J_c$  properties at the sintering temperature of 700 °C, but the samples of x = 10%, 12% and 15% present higher  $J_c$  values when sintered at 900 °C.

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#### 1. Introduction

The discovery of superconductivity in MgB<sub>2</sub> with critical temperature of 39 K [1] has stimulated a lot of interests from scientists. Compared with

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high temperature superconductors, MgB<sub>2</sub> has no weak-link problem at grain boundaries [2], and the cost of raw materials is low. Therefore, MgB<sub>2</sub> is believed a promising candidate for the engineering applications in the temperature range of 20–30 K.

Many methods have been tried to fabricate MgB<sub>2</sub> wires [3–8], in which powder-in-tube (PIT) technique is an effective one. However, the critical current density (J<sub>c</sub>) of MgB<sub>2</sub> usually decreases rapidly with the increase of magnetic field (B), due to the lack of pinning centers. Fortunately, it is reported that chemical doping can improve the  $J_c$ and flux pinning of MgB<sub>2</sub> superconductors effectively [9-13]. Ma et al. [13] demonstrated that the  $J_c$  of MgB<sub>2</sub>/Fe tape had been improved by the 5 at% doping of ZrSi2, ZrB2 and WSi2, respectively, but the doping ratios have not been optimized in their investigation. Feng et al. [11] confirmed that all the Zr-doped MgB<sub>2</sub> bulk samples have better performance than the undoped one, and the highest  $J_c$  of  $2.1 \times 10^6$  A/cm<sup>2</sup> in 0.56 T at 5 K and  $1.83 \times 10^6$  A/cm<sup>2</sup> in self-field at 20 K had been achieved in the Mg<sub>0.9</sub>Zr<sub>0.1</sub>B<sub>2</sub> sample.

In this paper,  $MgB_2$  wires were fabricated by using  $ZrH_2$  as dopant. As  $ZrH_2$  decomposes to be Zr and  $H_2$  at high temperature, we believe it can act as good as Zr powders for the improvement of  $MgB_2$  performances. Therefore, we have paid our attentions to the optimum doping ratio of  $ZrH_2$  and its relation with the sintering temperatures.

#### 2. Experimental

ZrH<sub>2</sub>-doped MgB<sub>2</sub> wires were fabricated by using the in situ PIT method. Mg (100–200 mesh, 99%), ZrH<sub>2</sub> (1250 mesh, 90%) and B (325 mesh, 99%) powders with the composition of (1-x): x:2 (doping ratio x = 5%, 8%, 10%, 12%, 15%) were ground together in agate mortar for 3 h in air. Mixtures were properly filled into low carbon steel tubes (outer diameter: 16 mm, inner diameter: 12 mm). These composite tubes were grooverolled, with four times of intermediate annealing performed at 400 °C, 450 °C, 590 °C and 580 °C for 1 or 2 h, respectively, to recover the plasticity and ductility of the low carbon steel. Subse-

quently, they were drawn to wires with the diameter of 1.6 mm. In the end, short samples were heated up to 700 °C, 750 °C, 800 °C, 900 °C and sintered for 2 h respectively in a flow of high purity argon. Then furnace-cooled to room temperature. Undoped MgB<sub>2</sub> wire (x=0) was fabricated by using the same method, but only having two times of intermediate annealing at 590 °C and 580 °C, respectively.

Phase identification was performed by using the X-ray diffractometer. Microstructure analysis was carried out by scanning electron microscopy (SEM). Critical current ( $I_c$ ) was measured by the standard four-probe resistance method in magnetic fields up to 7 T at 4.2 K. The criterion for the  $I_c$  definition was 1  $\mu$ V/cm.  $J_c$  was obtained by dividing  $I_c$  by the cross-sectional area of the superconducting core.

#### 3. Results and discussion

Fig. 1 illustrates X-ray diffraction (XRD) patterns of x mol ZrH<sub>2</sub>-doped samples. MgB<sub>2</sub>, ZrB<sub>2</sub>, MgO, Fe<sub>2</sub>B,  $\delta$ -ZrH<sub>2</sub> and Mg can be identified in the samples sintered at 700 °C. The appearance of ZrB<sub>2</sub> is because that ZrH<sub>2</sub> decomposed and reacted with B during the sintering process, as shown following:

$$ZrH_2 \rightarrow Zr + H_2$$
  
 $Zr + 2B \rightarrow ZrB_2$ 

As the relative peak intensity of  $ZrB_2$  increases with the increasing of x, we believe the amount of  $ZrB_2$  rises up correspondingly. However, the existence of  $\delta$ - $ZrH_2$  indicates that  $ZrH_2$  did not decompose thoroughly when sintered at 700 °C.

Fe<sub>2</sub>B appears mainly as the interface layer between the sheath materials and the superconducting core. It is the formation of Fe<sub>2</sub>B that consumed some B atoms and led to the surplus of Mg.

As the sintering temperature increasing, the phase compositions of these samples change slightly. No  $\delta$ -ZrH<sub>2</sub> can be found in all samples sintered at 750 °C, such as shown in the sample x = 8%. MgB<sub>4</sub> appears in the samples sintered above 850 °C, especially at 900 °C. The concentration of

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