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Construction of five zinc coordination polymers with 4-

substituted bis(trizole) and multicarboxylate ligands:

Syntheses, structures and properties

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ABSTRACT

Five zinc coordination polymers $\{[Zn_2(mtrb)_2(btec)] \cdot H_2O\}_n$ (1), $\{[Zn(mtrb)(Meip)] \cdot H_2O\}_n$ (2), $[Zn(mtrb)(nip)]_n$ (3), $\{[Zn(mtrb)(hip)] \cdot H_2O\}_n$ (4) and $[Zn(mtrb)(nbdc)]_n$ (5) (mtrb = 1,3-bis(1,2,4-triazol-4-ylmethyl)benzene, btec = 1,2,4,5benzenetetracarboxylate, Meip = 5-methylisophthalate, nip = 5-nitroisophthalate, hip = 5-hydroxyisophthalate, nbdc = 4-nitro-1,2-benzenedicarboxylate) were synthesized and characterized. 1 shows a (4,4)-connected self-catenated 3D network with a point symbol of $\{6^2.8^4\}\{6^4.8^2\}_2$, which belongs to the 4,4T32 topological type. 2 is a 1D chain structure containing [Zn(mtrb)(Meip)] 20-membered cycle. 3 is a 1D chain structure containing [Zn(nip)₂] 16-membered cycle and [Zn(mtrb)₂] 24-membered cycle. 4 and 5 exhibit the undulated 2D (4,4) network. 1 is a highly sensitive and selective luminescence sensor for detection of $Cr_2O_7^{2-}$, CrO_4^{2-} and Fe^{3+} in the aqueous solution. The detection limits for $Cr_2O_7^{2-}$, CrO_4^{2-} and Fe^{3+} are 3.05 μ M, 5.72 μ M, and 6.28 µM, respectively. The luminescence sensor is stable and can be recycled for detection at least five times. The possible quenching mechanisms were discussed.

Keywords: Coordination Polymer; Structure; Luminescent sensor; Detection of dichromate; Detection of ferric

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